

Differential sensing for the regio- and stereoselective identification and quantitation of glycerides

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Glycerides are of interest to the areas of food science and medicine because they are the main component of fat. From a chemical sensing perspective, glycerides are challenging analytes because they are structurally similar to one another and lack diversity in terms of functional groups. Furthermore, because animal and plant fat consists of a number of stereo- and regioisomeric acylglycerols, their components remain challenging analytes for chromatographic and mass spectrometric determination, particularly the quantitation of species in mixtures. In this study, we demonstrated the use of an array of cross-reactive serum albumins and fluorescent indicators with chemometric analysis to differentiate a panel of mono-, di-, and triglycerides. Due to the difficulties in identifying the regio- and stereochemistry of the unsaturated glycerides, a sample pretreatment consisting of olefin cross-metathesis with an allyl fluorescein species was used before array analysis. Using this simple assay, we successfully discriminated 20 glycerides via principal component analysis and linear discriminant analysis (PCA and LDA, respectively), including stereo- and regioisomeric pairs. The resulting chemometric patterns were used as a training space for which the structural characteristics of unknown glycerides were identified. In addition, by using our array to perform a standard addition analysis on a mixture of triglycerides and using a method introduced herein, we demonstrated the ability to quantitate glyceride components in a mixture.

array sensing | glyceride | chemometrics | serum albumin | differential sensing

Glycerides are the primary component of animal fats and vegetable oils (1). They consist of one, two, or three fatty acids esterified on glycerol, and hence are referred to as mono-, di-, and triglycerides, respectively. The structural diversity of glycerides derives in part from their fatty acid alkyl groups, which can differ in carbon number (i.e., chain length), the degree of unsaturation, the position of olefins, and the configuration of the olefins (i.e., *cis/trans*). Furthermore, these fatty acid alkyl groups can be connected to the *sn*-1, -2, or -3 carbons on glycerol. Hence, a variety of regio- and stereoisomers can exist for glycerides, posing a challenge for mass spectrometry (2). Further, because the differences in chain length primarily result from the presence of greater or fewer methylene groups, NMR spectroscopy can be ambiguous (3).

The analysis of glycerides is primarily important to the food and nutrition industries for tasks such as authenticating edible oils (4), designing foods with certain physical properties (5), and studying how fats are digested and absorbed (6). In particular, classifying all of the various kinds of regio- and stereoisomers of glycerides is biologically important because lipases, enzymes that catalyze the hydrolysis of glycerides into fatty acids and glycerol, exhibit selectivity based on these features of the glyceride substrates. As examples, the position and configuration of olefins, the identity of fatty acid alkyl groups, as well as their position on glycerol (i.e., *sn*-1,3 versus *sn*-2), all contribute to differing biological activity (7, 8). Studying the selectivity of these lipases has applications in understanding diseases, including fat malabsorption disorders, hypercholesterolemia, atherosclerosis, and

diabetes (9, 10). Research on metabolic disorders has shown that fatty acid accumulation can exert a toxic or a protective effect on a tissue, depending on the specific tissue type (e.g., liver, cardiac, or skeletal muscle) (11, 12) and health state (e.g., diabetic) (13, 14) as well as on the fatty acids (e.g., saturated or unsaturated) (15). Sequestration of fatty acids by esterification to glycerides is one pathway by which these effects are regulated (16). Thus, a deeper understanding of the distinct roles of the cellular storage of structurally different glycerides in normal and disease states is a desirable avenue of research (17). However, currently only limited information is available about the composition of glycerides in adipose and nonadipose tissue.

The most common method of glyceride identification is mass spectrometry (MS) (2, 18). However, as alluded to above, this approach has drawbacks. Because glycerides are neutral molecules, they must be ionized to be analyzed by MS. Saponification can be used to obtain the fatty acids, which are both volatile and charged, thereby facilitating MS analysis, but information about the glyceride structure is lost in this process (18). Electrospray ionization and atmospheric pressure chemical ionization are used to ionize glycerides directly; however, the ion yields are low compared with preionized lipids (19, 20). Furthermore, the ability of a glyceride to be ionized using these methods often varies. For example, ion abundance generally increases with increasing number of double bonds in the fatty acid alkyl chain and can also depend on fatty acid alkyl chain length (21). These significant variations in ion abundance mean that ionization methodologies must be developed and tailored to a specific application to satisfactorily detect each glyceride of interest (19). Finally,

Significance

Lipid metabolism is a growing area of biochemical research because understanding these pathways could lead to treatments for metabolic disorders such as obesity and type 2 diabetes. To study lipid metabolism, researchers need tools to identify and quantitate glycerides, the main component of animal fat. However, it can be difficult to tell one glyceride apart from another subtly different glyceride using current analytical methods such as mass spectrometry. Thus, we developed a method of discriminating glycerides using an array of cross-reactive proteins in conjunction with pattern recognition algorithms. By incorporating an olefin metathesis pretreatment step, we were able to distinguish glyceride regio- and stereoisomers and to predict these structural features. Finally, we achieved quantitation of glycerides in mixtures.

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these variations render the quantification of glycerides, particularly in a complex mixture, quite challenging when using MS (22).

Regio- and stereoisomers further confound the discrimination of glycerides by MS, because isomers share the same mass. Other techniques such as chemical derivitization of the glycerides, ion fragmentation, and specialized HPLC must be coupled with MS to effect differentiation of isomeric species. For example, ozonolysis has been used to cleave the double bonds in unsaturated glycerides before ionization to deduce the positions of double bonds (23). Nonaqueous reverse-phase (NARP)-HPLC can resolve *cis/trans* isomers of triacylglycerols and double-bond positional isomers after treatment of the olefins with bromine (24). Silver ion chromatography has been used to separate triacylglycerol positional isomers under specifically developed solvent and column temperature conditions (25). Silver cationization as a postcolumn treatment in conjunction with NARP-HPLC and ion fragmentation has also been used for triglyceride positional isomer determination (26, 27). Thus, although these current approaches to glyceride isomer analysis have been successful, they are complicated, labor intensive, time-consuming, and at times inconsistent in their results (26).

Because glycerides are structurally very similar to one another, we believed that a differential sensing array-based approach would be most suitable for their classification. Our hypothesis was that if a cross-reactive array could be created that was responsive to the subtle structural differences inherent in glycerides, it could be used to pattern individual glycerides, identify structural features of unknown glycerides, and potentially quantitate glycerides in a mixture. Cross-reactive arrays have been successfully used in a number of sensing applications (28–33). Differential sensing mimics the mammalian senses of olfaction and gustation by detecting the pattern of response of an analyte to a collection of semiselective receptors (34, 35). In mammals, the characteristic pattern for a scent or taste is interpreted and stored by the brain (36). In the laboratory, chemometric routines such as principal component analysis (PCA) and linear discriminant analysis (LDA) are used to extract the relevant information from the array. Both PCA and LDA are multivariate methods that reduce the dimensionality of a data set. PCA does so by finding unbiased orthogonal axes that describe decreasing extents of variance in the data derived from different samples (classes) and repetitions of the samples (37). Any grouping of like samples

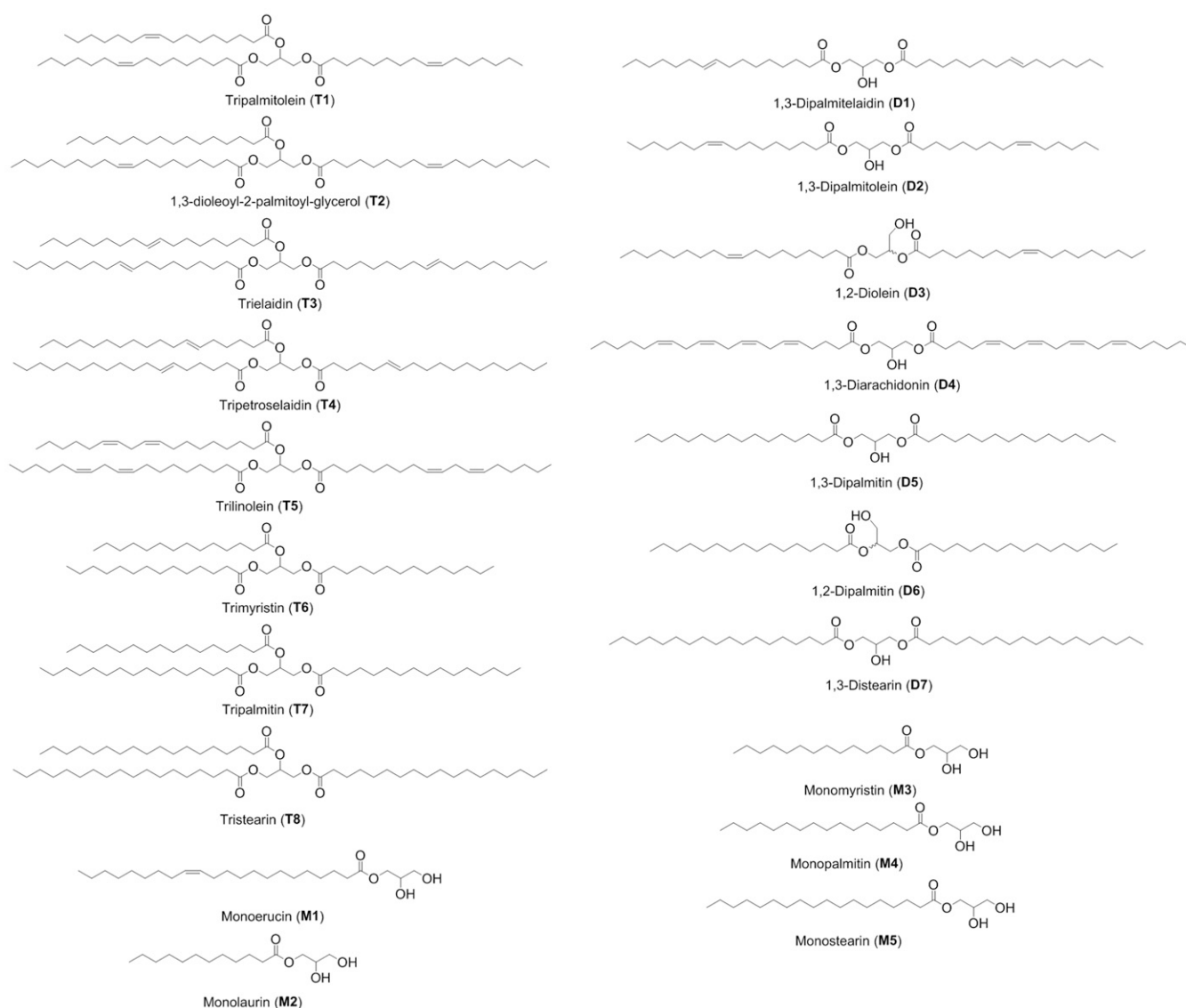


Fig. 1. Glyceride panel with structures and names.

represents intrinsic similarities between the sample datasets whereas separate classification represents differences in that variable space. LDA classifies samples by calculating discriminant functions that maximize the separation between predetermined classes and minimizes the separation within these classes (38, 39). Thus, LDA is a supervised method, meaning that the classes are provided as inputs into the algorithm. For this reason, a validation method called a leave-one-out cross-validation is used to test the predictive value of the model. Further, LDA can be used to predict the identity of unknowns by identifying which classes in the training set the unknowns most resemble.

Therefore, the goal of this project was to develop an array of cross-reactive receptors that could discriminate glycerides. The glycerides selected are shown in Fig. 1. The panel includes commercially available mono-, di-, and triacylglycerols with fatty acid alkyl groups that are relevant to mammalian biology (40). Moreover, the panel consists of examples of each of the following stereo- and regioisomers: (i) *cis/trans* olefins (**D1** and **D2**; **T2** and **T3**), (ii) differing position of the olefin (**T3** and **T4**), and (iii) differing position of the fatty acid alkyl groups on the glycerol (**D5** and **D6**). Clearly, it would be extremely challenging to create highly selective receptors for each individual glyceride, and thus a differential sensing method seems the only reasonable approach to creating an optical sensing routine to identify and classify these structures.

Because glycerides are extremely hydrophobic analytes, we postulated that serum albumins (SAs) would be suitable cross-reactive receptors with which to test our hypothesis. SA is a common plasma protein that binds hydrophobic molecules to transport them through the hydrophilic environment of blood plasma (41). The protein binds a number of endogenous compounds: long-chain fatty acids ($K_a = 10^6$ – 10^7 M $^{-1}$) (42), bile acids ($K_a = 10^3$ – 10^5 M $^{-1}$) (43), and steroids ($K_a = 10^3$ – 10^5 M $^{-1}$) (44–46), as well as many drugs, toxins, and fluorophores (41). Despite being composed of fatty acid alkyl groups, glycerides bind less tightly to SAs and in a different location than their fatty acid counterparts (47). The primary sequence of SAs differs between species, which thus exhibit differences in ligand binding (41). Previously, we have used arrays of SAs for the differentiation of other hydrophobic analytes including fatty acids (48), terpenes (49), and plasticizers (50). However, none of these previous studies involved differences between the analyte structures as subtle as glycerides do, nor had we challenged our methods to identify structural aspects of an unknown. Furthermore, we had never implemented a quantitation assay in a complex mixture. Because the binding of ligands to SAs is known to depend on subtle differences in their structure (41), we anticipated that success could be achieved but would be highly dependent upon the signaling modality and potentially analyte prederivatization.

Thus, herein we describe a method using SAs to fingerprint glycerides that classifies them as mono-, di-, or triglycerides. The glycerides were further classified based on fatty acid chain length, ester positions on glycerol, and olefin regio- and stereochemistry. For the unsaturated glycerides in the panel, differentiation based on olefin position and stereochemistry was achieved by the use of a pretreatment olefin metathesis. Using the protocols described herein, structural features of unknown glycerides could be identified. Furthermore, the quantitation of trilinolein in a mixture of triglycerides was achieved by application of the standard addition method using a net analyte signal technique (SANAS) presented herein.

Results and Discussion

Indicator Uptake Experiments with SA and Triglyceride. The optical signaling approach we envisioned required that fluorophores bind to the SAs, and signal the addition of glycerides upon binding to the SAs. To identify appropriate fluorophores for the sensing ensemble, several candidates were screened for their fluores-

cence modulation both in the presence of bovine serum albumin (BSA) and triglycerides. The triglycerides were insoluble in water but dissolved in the presence of SA, presumably due to binding in the hydrophobic sites of the protein. Thus, indicator uptake experiments were undertaken in which a SA/triglyceride solution was titrated with a fluorescent indicator. The change in emission of the indicator in the presence of the triglyceride was compared with the indicator's change in emission when it was added to SA alone. This protocol revealed which fluorophores bind the SAs and modulate in the presence of glycerides. The fluorophores screened included dansyl amide (DNSA), 2-anthracenecarboxylate (2-AC), 1-anilinonaphthalene sulfonic acid (ANS), and a nitrobenzoxadiazole fatty acid (NBD-FA). The first three compounds were known previously to bind to SA (41). We synthesized NBD-FA by a literature procedure (*SI Appendix*) (51).

An example of these indicator uptake experiments is shown in Fig. 2 for NBD-FA, whereas the titrations for the other indicators can be found in *SI Appendix*, Figs. S2–S4. When NBD-FA was added to SA, its emission increased (“BSA” in Fig. 2); however, when trimyristin (**T6**) was present with the BSA, this increase in emission was attenuated (“BSA/T6” in Fig. 2), indicating that the binding of **T6** to SA was inhibiting the binding of NBD-FA. It should be noted that the decrease in emission at concentrations above 100 μ M NBD-FA was attributed to self-quenching as multiple NBD-FAs bind to the SA.

DNSA and NBD-FA were inhibited in their binding to SA in the presence of triglycerides and therefore were suitable for use in the array. The other indicators, 2-AC and ANS, did not demonstrate significant inhibition in binding to SA in the presence of trimyristin. However, we chose to include ANS in the array as it is commonly used to probe the folding of SAs because it binds in the crevices between domains (41). We anticipated it would respond to other glycerides even if our initial tests with trimyristin did not reveal emission modulations.

First Iteration of the Well-Plate Array. The cross-reactive array was constructed using 96-well plates. As a preliminary test of the array's ability to differentiate the glycerides, an assortment of eight mono-, di-, and triglycerides from the total panel (Fig. 1) were selected. These glycerides were mixed with individual combinations of the three indicators (DNSA, NBD-FA, and ANS) and bovine and human serum albumin (six combinations: BSA/DNSA, human serum albumin (HSA)/DNSA, BSA/NBD-FA, HSA/NBD-FA, BSA/ANS, and HSA/ANS) in a 96-well plate with 8 replicates for each glyceride, and the emission of each well was recorded.

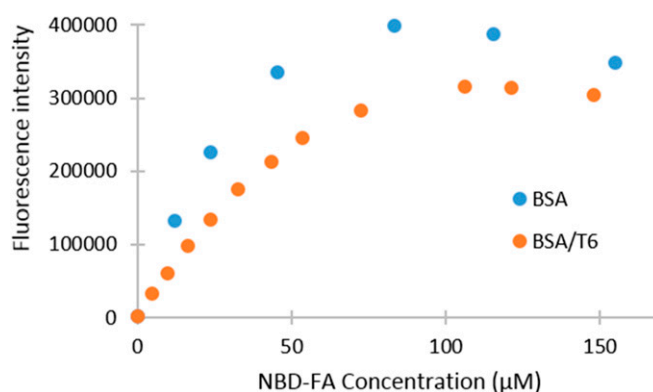


Fig. 2. Addition of NBD-FA (0–155 μ M) to BSA (100 μ M) and to BSA (100 μ M)/**T6** (90 μ M) in 10 mM phosphate buffer, pH 7.00, 0.02% NaN₃; <0.3% THF; λ_{ex} = 470 nm, λ_{em} = 540 nm.

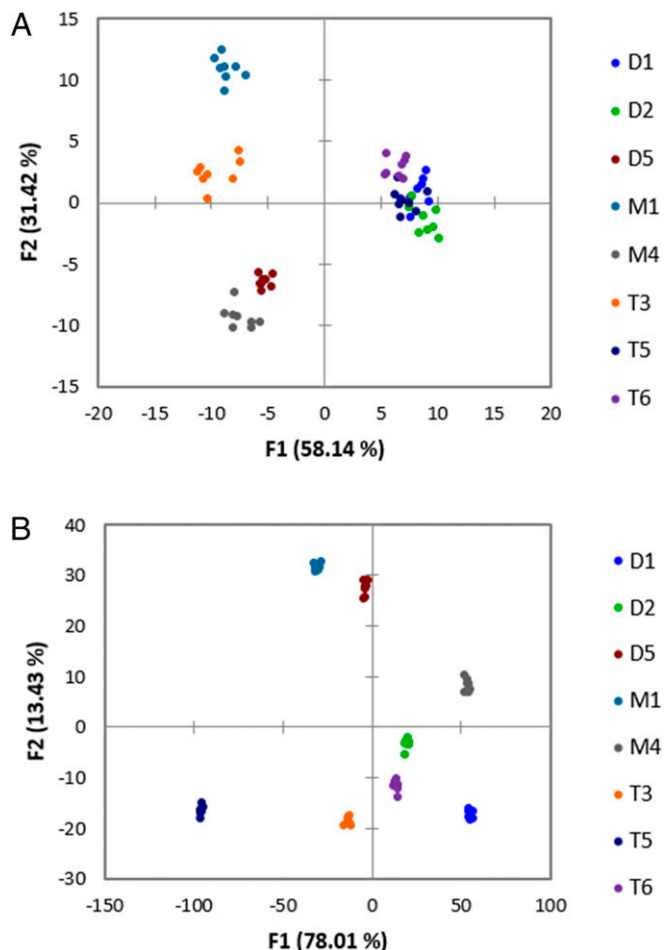


Fig. 3. LDA plots of data collected from 96-well plates without olefin metathesis (A) and with olefin metathesis (B). (A) BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), and NBD-FA (60 μ M) in phosphate buffer with <2% (wt/vol) THF (see [SI Appendix, Table S1](#) for read parameters) and (B) BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), and NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with <5% (vol/vol) THF (see [SI Appendix, Table S4](#) for read parameters).

A LDA score plot of the fluorescence data collected from this array is shown in Fig. 3A, which has a 98% cross-validation analysis. Whereas the eight glycerides were differentiated by this array with high accuracy, there was significant visual overlap along the F1 and F2 axes of three of the glycerides. **D1** and **D2**, which are *cis/trans* isomers of one another, were not well discriminated, and **T5** was also poorly separated. The fact that our approach demonstrated overlap of unsaturated glycerides with only 8 of the 20 total targets in the panel caused us to reconsider the approach. Because differences between the unsaturated glycerides were the hardest to discriminate, we anticipated even further problems when attempting to classify the position, stereochemistry, and number of double bonds.

Olefin Metathesis of Glycerides. One strategy to discriminate the differences in the olefins would be their derivitization. We contemplated the use of bromination and dihydroxylation, but ultimately chose olefin metathesis. Metathesis reactions of glycerides and fatty acids have been previously explored with the intent of using these compounds to make chemical products more sustainably (52, 53). Such a prior derivitization approach is similar in principle to the use of other olefin reactions, such as halogen

addition (24) and silver cationization, which are able to resolve stereo- and regioisomers of glycerides when combined with NARP-HPLC-MS methods (26).

Olefin metathesis was chosen for several reasons. First, using a metathesis reaction results in products of differing length depending upon the positions of the olefins in the glycerides, thus potentially making the products from such glycerides unique. Second, olefin metathesis catalysts have differing reactivity toward *cis* and *trans* stereoisomers (54), thus leading to different extents of metathesis depending upon the stereochemistry of the starting fatty acid chains. Third, the reaction conditions are relatively mild, and the reaction mixture can be used directly in the SA array without any purification. This factor allows the cross-metathesis reactions of multiple glycerides to be performed in parallel in a polypropylene well plate for efficient workflow. Lastly, this reaction could be used to introduce an additional fluorophore for optical analysis. Our strategy therefore used fluorescein conjugated to an olefin, resulting in mixed olefin products.

With these goals in mind, the allyl fluorescein derivative **AF** (Fig. 4) was synthesized according to a literature procedure (55, 56). As a model reaction, we screened several different reaction conditions for cross-metathesis between **AF** and monoerucin (**M1**) to optimize the reaction conditions (*SI Appendix, Table S2*). The conditions that were selected to be optimal for our purposes are shown in Fig. 4. LC-MS analysis of the reaction mixtures confirmed the conversion of **AF** to mixed glyceride and **AF**-containing compounds.

Once the optimal conditions were identified, all of the unsaturated glycerides in the panel were subjected to the cross-metathesis reaction in parallel in the same manner in a well plate, and the reaction mixtures were analyzed by LC-MS to determine that **AF** was metathesizing with the glycerides and that the conversion was reproducible. In these experiments, the **AF**-glyceride products observed in the LC-MS analysis contained the

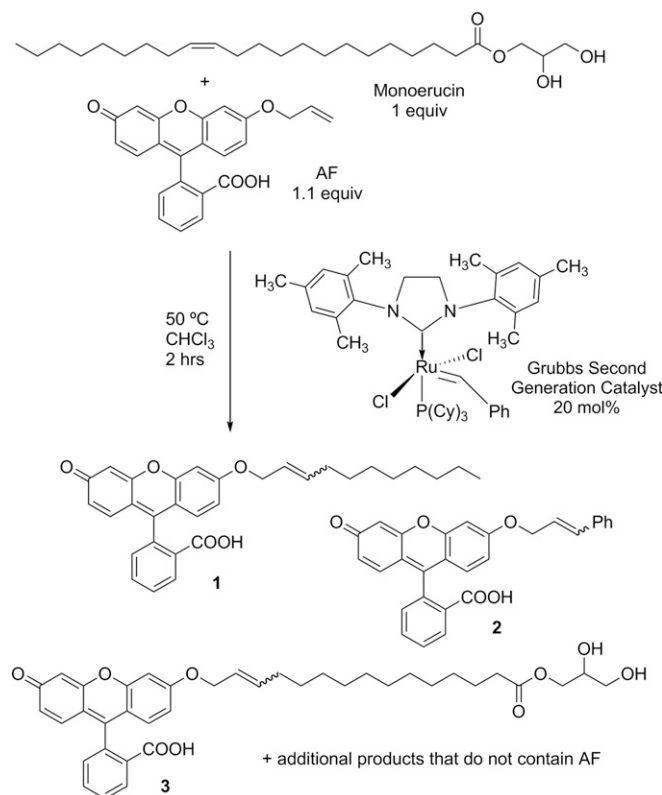


Fig. 4. Olefin metathesis of monoerucin (**M1**).

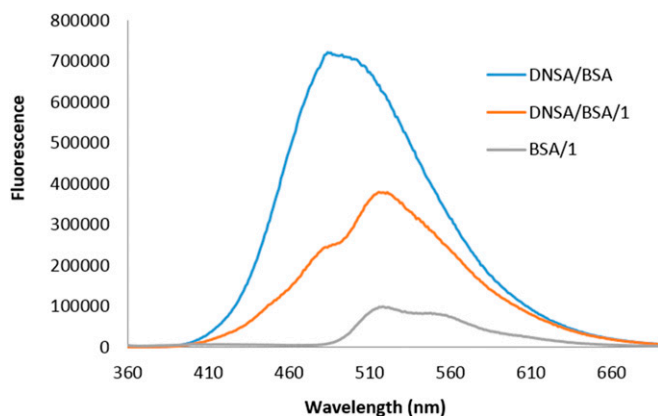


Fig. 5. Fluorescence spectra for DA/BSA (250 μ M/100 μ M), 1/BSA (90 μ M/100 μ M), and DA/1/BSA (250 μ M/90 μ M/100 μ M) in 10 mM phosphate buffer, pH 7.00, 0.02% NaN_3 ; $\lambda_{\text{ex}} = 350$ nm.

alkyl portion of the glyceride, as in **1**, and smaller amounts of products such as **3**. *SI Appendix, Table S3* summarizes the percent conversion of **AF** to the corresponding mixed product (**1** type) for each unsaturated glyceride. The chromatogram for absorption at 280 nm was used to quantitate the conversion, and we assumed that the extinction coefficient at that wavelength for **AF** and **AF**-glyceride species did not differ significantly. In each reaction, some unreacted **AF** remained, and some of product **2**, which results from the catalyst loading used, was also observed. Finally, it is highly likely that olefin combinations that did not involve **AF** (i.e., self-metathesis of the glyceride) are also generated in the reaction. However, these species were not observable using conventional LC-MS analysis. Although the conversion differed for each glyceride, all of the glycerides were found to metathesize with **AF**.

Exploring Förster Resonance Energy Transfer with DNSA and AF. Dansyl and fluorescein moieties have been reported to act as a Förster resonance energy transfer (FRET) pair (57, 58), and therefore we investigated this property with the metathesized glycerides. FRET would be advantageous to the chemometric patterning because it would add an additional facet of cross-reactivity to the array. To test whether FRET occurred between DNSA and **1** in the presence of BSA, we measured the emission spectra of solutions of DNSA/BSA, **1**/BSA, and DNSA/**1**/BSA by exciting at 350 nm (Fig. 5). These data support FRET because the DNSA/**1**/BSA sample exhibits emission at the λ_{max} of 520 nm for fluorescein that was much higher in intensity than the **1**/BSA sample. We also performed a titration in which DNSA was titrated into BSA/**1** while exciting at 350 nm (for dansyl) and observed an increase in emission of fluorescein with a λ_{max} at 520 nm (*SI Appendix, Fig. S5*), which supports FRET between DNSA and **1** when they are both bound to BSA. Hence, we added DNSA to the metathesized glycerides as a simple way to generate additional cross-reactivity.

Second Iteration of the Well-Plate Array. Once we had established that the olefin metathesis reaction was successfully mixing **AF** with fragments of the unsaturated glycerides in our panel, we were ready to incorporate this reaction into the well-plate array with SA. To do so, we performed the metathesis reaction on the glycerides in a well plate in chloroform. However, because chloroform is immiscible with buffered SA solutions, after metathesis the chloroform was allowed to evaporate, and the residues were taken up in tetrahydrofuran (THF). The THF solutions of the glyceride metathesis reactions were then exposed to BSA and HSA, separately, in a 96-well plate. DNSA was added to one set of

BSA and HSA plates, and buffer was added to another set of BSA and HSA plates. For the plates that contained DNSA, the emission of fluorescein, DNSA, and the FRET pair between the two fluorophores was measured. For the plates without DNSA, only the emission of the fluorescein was measured. We wanted to measure the fluorescein emission both in the absence and presence of DNSA to see if there was a significant difference in this signal and thus an additional variable in our array. The first iteration of the well-plate array was also performed exactly as described above for each glyceride. Both sets of fluorescence data were used in the pattern recognition algorithm for a total of 14 variables (Fig. 6).

The LDA plot generated for eight glycerides is shown in Fig. 3B. The cross-validation was 100%. The clustering within replicates of the same glyceride in this plot (Fig. 3B) was much tighter than in the plot generated using only the first iteration of the well-plate array (Fig. 3A), and the separation between different glycerides was more marked. In particular, compared with Fig. 3A, the *cis* isomer **D2** was discriminated from its *trans* isomer **D1**, and **T5** no longer overlaps with **D1** and **D2**.

From the factor-loading plot (Fig. 6), we concluded that the metathesis parameters contributed significantly to the discrimination. A factor-loading plot shows the contribution of each of the original input variables to each factor axis in the reduced variable space. The loading plot in Fig. 6 shows that the metathesis variables, which are marked in red, contribute significantly to F1 and F2. This result supports our hypothesis that the metathesis reaction would improve the differentiation of a panel of glycerides containing unsaturated species.

Array Reproducibility. Next, we wanted to be sure that we could replicate the results of the array. In other words, we wanted to know if we would get essentially the same pattern for the glycerides each time we performed an independent repetition of the entire experiment, including repeating the metathesis reaction and starting from newly prepared stock solutions of all of the array components. Such a complete reproduction has rarely, if ever, been discussed in papers describing differential sensing routines. Hence, we performed an independent repetition of the array on those same eight glycerides two more times for a total of three data sets. The LDA plots for these repetitions can be found in *SI Appendix, Fig. S6*. The relative position of all eight glycerides

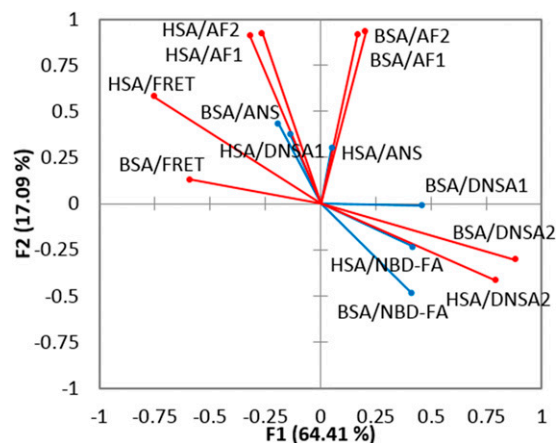


Fig. 6. Loading plot corresponding to the LDA in Fig. 3B. DNSA1 refers to the nonmetathesis part of the assay, whereas DNSA2 refers to the measurement of the DNSA emission in the presence of the metathesized glycerides. AF1 refers to the fluorescein emission in the absence of any DNSA, whereas AF2 refers to the fluorescein emission in the presence of DNSA.

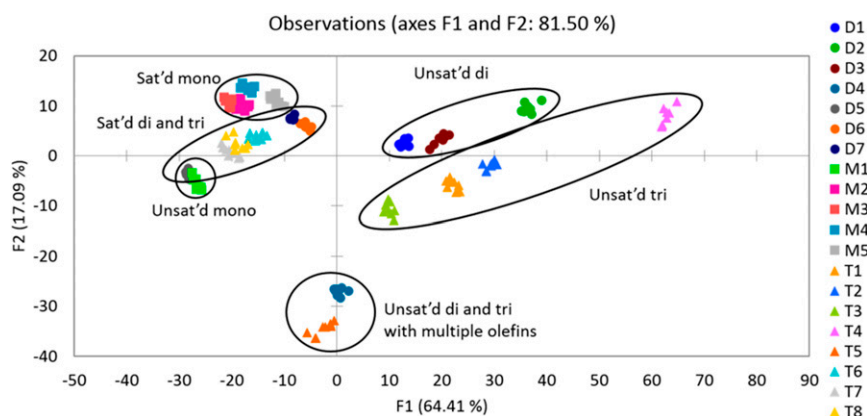


Fig. 7. LDA plot of data collected from 96-well plates. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with <5% (vol/vol) THF (see *SI Appendix, Table S4* for read parameters). Cross-validation: 98%.

was essentially the same for each plot, and we concluded that the array is reproducible.

To further test the consistency of the array, we used LDA as a predictive tool. Because LDA entails that the classes be given as inputs, one can take two sets of data, in our case the independent repetitions of the array on those eight glycerides, and assign one as a training set and the other as a prediction set. The training set teaches the algorithm which responses from the array correspond to which glyceride, and then the algorithm can assign the glyceride identity to the sets of array responses in the prediction set. Hence, we alternately treated each of the three repetitions as the training set or the prediction set. For 64 data points in each set (8 replicates of 8 glycerides), the average accuracy was 87% for the 6 combinations of the 3 independent replications (*SI Appendix, Table S5*).

Analysis of the Full Glyceride Panel. Finally, we performed the full 14-variable array on all 20 glycerides in the panel. A LDA plot was obtained with a cross-validation of 98% (Fig. 7). The only error in classification in the cross-validation was between one replicate of the saturated triglyceride **T7**, which was misclassified as another saturated triglyceride **T8**. In general, the unsaturated di- and triglycerides are on the right side of the plot, whereas the saturated glycerides are on the left side. Within the unsaturated glycerides, there is very clear separation between all of the mono-, di-, and triglycerides. The unsaturated monoglyceride **M1** is associated more closely with the saturated glycerides on the left side of the plot. However, this result is reasonable because **M1** metathesizes poorly (*SI Appendix, Table S3*). The two glycerides that consist of fatty acid alkyl groups with multiple double bonds (**T5**, **D4**) are found near one another at the bottom center of the plot.

Moreover, the stereo- and regioisomers of the unsaturated di- and triglycerides are successfully distinguished by the array. Within the unsaturated diglycerides, the *cis/trans* isomers **D1** and **D2** are discriminated from one another. Within the unsaturated triglycerides, the olefin positional isomers **T3** (olefin between carbons 9 and 10) and **T4** (olefin between carbons 6 and 7) are also clearly separated. Furthermore, **T2** contains fatty acid alkyl groups that are stereoisomeric (*cis*) to the fatty acid alkyl groups in **T3** (*trans*), and these species are also clearly differentiated from one another.

All of the saturated glycerides are also well discriminated, although the separation is less marked than between the unsaturated glycerides. This finding is unsurprising because the metathesis reaction does not occur with the saturated glycerides, and hence their discrimination would not be expected to be

as significantly improved by the pretreatment step. Within the saturated mono-, di-, and triglycerides, the difference between the compounds is simply the number of carbons in the fatty acid alkyl chains; however, the panel does contain one regioisomer pair, **D5** and **D6**. **D5** is esterified at the *sn*-1 and *sn*-3 positions of glycerol, whereas **D6** is esterified at the *sn*-1 and *sn*-2 positions of glycerol. The array is able to differentiate these regioisomers as well.

PCA (Fig. 8) also results in good discrimination of the glycerides and similar groupings of glycerides based on their structural features to the LDA plot. In PCA, the identities of the glycerides are not given as inputs, so the analysis only takes the variance in the variables (i.e., emission signals for each receptor) into account without knowing what the classes (i.e., identities of the glycerides) are supposed to be. It is apparent in the PCA that the SA-based receptors respond to the glycerides in a differential manner. In other words, the receptors behave differently from one another with each analyte. When receptors in an array behave similarly to one another, the PCA shows low dimensionality

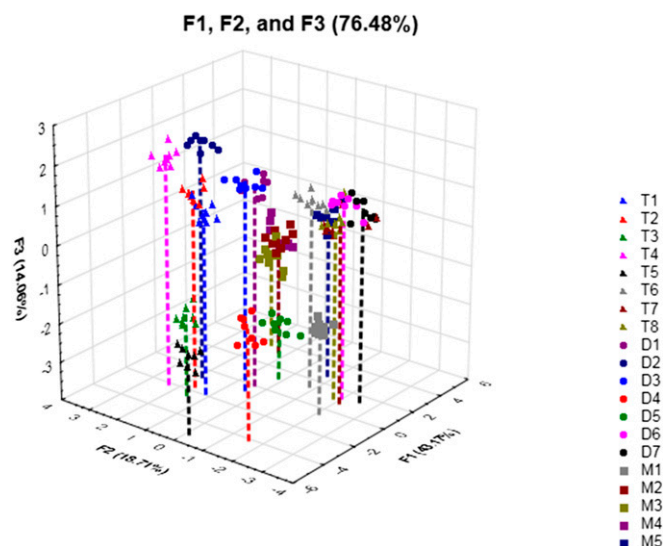


Fig. 8. PCA plot of data collected from 96-well plates. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with <5% (vol/vol) THF (see *SI Appendix, Table S4* for read parameters).

(i.e., most of the variance is described by one or two factor axes). For example, if the receptors all display their highest emission values in the presence of analyte 1 and the lowest emission values in the presence of analyte 2, the PCA would be able to describe the bulk of the variance in the data along a single factor axis. In our case, the first five factor axes (F1–F5) describe 90% of the variance in the data. The high dimensionality in the data is indicative of receptors that respond differently from one another to the analytes in the panel. High dimensionality in the data is indicative of a truly cross-reactive array of receptors (59).

Prediction of Glyceride Structural Features. Because glycerides that share structural features are grouped in the LDA and PCA plots, we postulated that our array could be used to identify these features in unknown glycerides. As a way of testing this hypothesis, an independent reproduction of the array was performed on all 20 glycerides (LDA plot for this repetition is shown in *SI Appendix, Fig. S7*), and the predictive feature of LDA was used. The data set from one experiment was used as the training set, and the other data set was used as the prediction set. The analyte classes that were put into the LDA were structural features of the glycerides: unsaturated triglyceride, saturated triglyceride, polyunsaturated triglyceride, unsaturated diglyceride, saturated diglyceride, polyunsaturated diglyceride, unsaturated monoglyceride, and saturated monoglyceride (Fig. 9 and *SI Appendix, Fig. S8*).

The LDA correctly assigned the glycerides in the training set to its structural feature class in an average of 89% of 160 cases. Because our array is primarily designed to target differences between unsaturated glycerides through the metathesis pretreatment, we postulated that separating the saturated and unsaturated glycerides from one another first would improve the accuracy of the predictions. The analyte classes were assigned more generally to either saturated, unsaturated, or polyunsaturated (*SI Appendix, Fig. S9*). The LDA correctly assigned the glycerides in the training set according to these classes in an average of 96% of 160 cases. Once it was known whether the glycerides were saturated or unsaturated, then whether they were mono-, di-, or triglycerides could be determined. The training and prediction sets were split into saturated and unsaturated. Within the unsaturated set, the analyte classes were triglyceride, diglyceride, monoglyceride, polyunsaturated di-

glyceride, and polyunsaturated triglyceride (*SI Appendix, Fig. S10*). With the saturated set, the analyte classes were triglyceride, diglyceride, and monoglyceride (*SI Appendix, Fig. S11*). For the unsaturated set, the glycerides were assigned correctly in an average of 95% of the 80 cases, whereas for the saturated set, the glycerides were assigned correctly in an average of 73% of the 80 cases. The lower accuracy of the prediction for the saturated glycerides is unsurprising because it is apparent in the LDA plot that the saturated glycerides are less clearly separated from one another compared with the unsaturated glycerides. However, overall, the array shows a good ability to predict the structural features of glycerides using LDA.

Quantitation of Trilinolein in Triglyceride Mixtures Using Standard Addition.

Introduction to standard addition and net analyte signal. Because quantitation is one of the major challenges that MS analysis faces in regard to glyceride analysis, we sought to demonstrate the use of our assay for quantitation of a glyceride of interest within a mixture of other glycerides. Because the receptors in the array are semiselective in nature, the presence of glycerides other than the species of interest in the mixture will interfere with the signal elicited from the receptor by that species. Hence, we chose to use the “standard addition” method to minimize the influence of the other glycerides in the mixture on the quantitation of the glyceride of interest.

Standard addition is a classic calibration method that is used when matrix effects are present in a sample that interfere with the signal the compound of interest elicits, confounding accurate quantitation of that compound (60). Ordinarily, the experimentalist tries to remove interference from the sample whenever possible; however, when it is not facile to physically remove the interfering background matrix, standard addition can be used to minimize the matrix's effect on the analysis. Standard addition is carried out by measuring the response of the mixture after successive additions of the pure analyte of interest. By plotting the response against the amount of the pure analyte in each addition, the concentration of the analyte of interest in the sample is obtained by a linear fit of the data. The x intercept of the line is the analyte concentration. This data analysis procedure is used

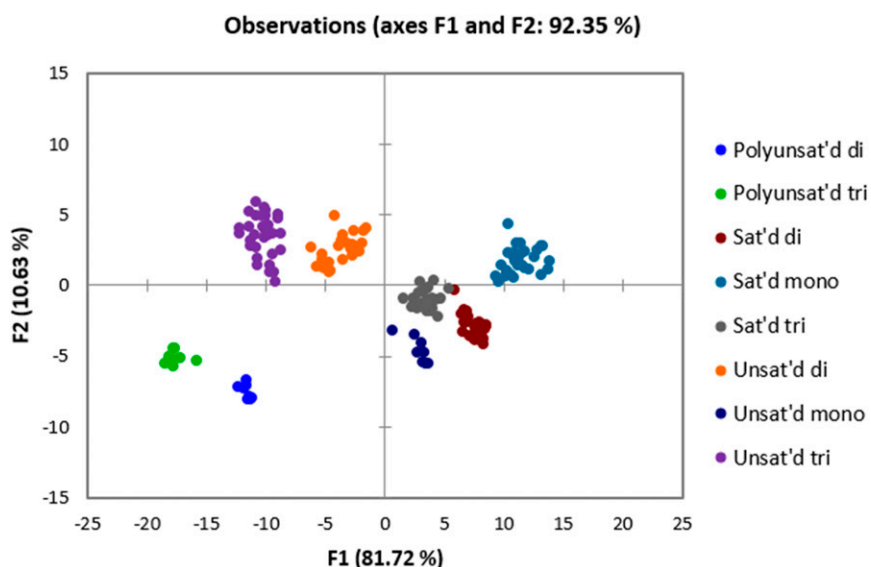


Fig. 9. LDA plot of training set of glycerides with analyte classes designated as structural features. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with <5% (vol/vol) THF (see *SI Appendix, Table S4* for read parameters). The fluorescence counts were expressed as a percentage of the total fluorescence for each receptor/variable. Cross-validation: 100%.

Mixture	Total concentration of the mixture, mg/mL	Actual T5 concentration, mg/mL	Predicted concentration, mg/mL	R ² of SANAS plot
A	0.04	0.0113	0.0338	0.7403
B	0.04	0.0158	0.0160	0.9517
C	0.01	0.0053	0.0056	0.9519
D	0.01	0.0042	0.0047	0.9664

By rebuilding the data set from fewer factor axes than the total number (for example, from F1–F3, discarding F4–F10), irrelevant information like noise, background, or interferences can be removed from **X**. From the rebuilt data set (**X'**), the NAS for the analyte of interest is calculated by using a type of singular value decomposition method called rank annihilation (see ref. 60 for a detailed account of this series of computations). The output is a vector that describes the NAS. By plotting the Euclidean norm of this vector (i.e., its magnitude) against the concentrations of the additions, one obtains a plot that is equivalent to the one described for the univariate case described above. The x intercept of the linear fit has the same meaning as in the univariate case, that is, the concentration of the analyte of interest in the mixture.

Applying SANAS to mixtures of glycerides. The first step in performing the quantitative analysis with our array was to generate mixtures of triglycerides with a known composition to study the accuracy

From the rebuilt data matrix, the net analyte signal ($||\text{NAS}||$) was calculated and plotted against the additions to give SANAS plots (Fig. 10 and *SI Appendix, Figs. S12–S14*). From these plots, the concentration of **T5** in the mixture was determined by finding the x intercept of the linear fit (Table 1). Good estimates of the concentration of **T5** in the mixtures were obtained for mixtures B, C, and D; however, a larger overestimation of the concentration of **T5** in mixture A was obtained. We attribute this error to poor linearity in the response of the sensors to the additions of **T5** that is evidenced by the lower correlation coefficient for the linear fit in the SANAS plot for mixture A (Table 1). Some of the nonlinearity was likely due to random experimental errors in solution and plate preparation. However, some deviations from linearity of the responses to the additions to all of the mixtures are inherent to the experimental method. The fluorescence signal collected from each well of the plate is mediated by the SA, fluorophore, and glycerides in the solution and is an indirect measurement of the **T5** content. Hence, we expect that nonlinearity in the response results from both the saturation behavior of ligand binding to SA as well as binding competition from glycerides in the mixture that are not **T5**. Nevertheless, for mixtures B, C, and D the array response showed a good linear dependence on the additions of **T5**. The SANAS method was



able to provide quite accurate values for the **T5** content without the necessity for removing or otherwise separating out the other triglycerides in the mixture.

Conclusions

In this work, we report an array of cross-reactive SAs and fluorescent indicators to discriminate a panel of structurally similar glycerides. An olefin cross-metathesis reaction with an AF species was used to pretreat the samples before array analysis to distinguish unsaturated glycerides. Using this assay, we successfully discriminated 20 mono-, di-, and triglycerides, including stereo- and regioisomeric pairs. These isomer types included *cis/trans* stereoisomers, double-bond positional regioisomers, and positional regioisomers on the glycerol core. Using LDA, we were able to predict structural characteristics of the glycerides. Finally, we demonstrated the use of multivariate standard addition with a cross-reactive array of chemosensors. We used our array to perform a standard addition of trilinolein to a mixture of triglycerides. By applying the established SANAS method, we were able to quantitate the trilinolein in the mixture.

Materials and Methods

General. The arrays were prepared in Costar 96-well plates (#3915). The plates were analyzed using a Biotek Synergy 2 Multimode Microplate Reader or a Biotek Cytation 3 Microplate Reader. LDA, PCA, and data normalization were done using XLSTAT 2011. The SANAS code was run on MATLAB 2014.

Glyceride Array (Part 1). Concentrated stock solutions of each glyceride were prepared in THF (concentrations on the order of millimolar). Stock solutions of human and BSA were prepared in 10 mM phosphate buffer (pH 7.0, 0.02% NaN₃) at a concentration of 500–700 μ M. SA-glyceride solutions were prepared from these stocks at concentrations of 150 μ M SA and 135 μ M glyceride [3% (vol/vol) THF, 97% (vol/vol) 10 mM phosphate buffer, pH 7.0, 0.02% NaN₃]. Concentrated stock solutions of DNSA and NBD-FA were prepared in DMSO. A concentrated stock solution of ANS was prepared in 10 mM NaOH (aq). Then, solutions of each of the three indicators at a concentration of 180 μ M in 10 mM phosphate buffer, pH 7.0, 0.02% NaN₃ were prepared.

The plates were made by placing 200 μ L of each SA-glyceride solution and 100 μ L of the 180 μ M indicator solution in each well [final concentrations: 100 μ M SA, 90 μ M glyceride, 60 μ M indicator, 2% (vol/vol) THF]. Each plate contained a column of indicator alone and a column of indicator and SA as controls. Eight replicates were performed for each glyceride/SA/indicator combination and for the controls.

The parameters for the reading of the well plates can be found in [SI Appendix, Table S1](#). The fluorescence data were normalized by first subtracting the emission of the control (indicator alone) from each data point, and then the data set was rescaled from 0 to 100.

Well-Plate Metathesis Reaction. Concentrated stock solutions of each glyceride were prepared in chloroform (concentrations were on the order of millimolar). Solutions of AF and Grubbs Second Generation Catalyst (G2) were also prepared in chloroform. These solutions were applied to a deep-well polypropylene plate such that each well contained 2 mM glyceride, 2.2 mM AF, and 0.4 mM G2 in a total volume of 1 mL of chloroform. Two controls were also included that contained only AF and G2. The plate was placed in an oven and heated at 50 °C for 2 h. Then the plate was removed and left in the fume hood to finish evaporating the chloroform overnight at ambient temperature and pressure. The next day, the dry material that remained in each well was taken up in 1 mL of THF with thorough mixing via a glass pipette to completely dissolve all of the material in the THF.

Glyceride Array with Metathesized Glycerides (Part 2). The solutions from the metathesis reaction in 1 mL of THF were used directly. Solutions of human and BSA were prepared in 10 mM phosphate buffer (pH 7.0, 0.02% NaN₃) at a concentration of 550 μ M. SA-glyceride solutions were prepared from these stocks at concentrations of 150 μ M SA, 135 μ M glyceride, 149 μ M AF, and 27 μ M G2 [6.8% (vol/vol) THF, 93.2% (vol/vol) 10 mM phosphate buffer, pH 7.0, 0.02% NaN₃]. A concentrated stock solution of DNSA was prepared in DMSO. Then a solution of DNSA at a concentration of 180 μ M in 10 mM phosphate buffer, pH 7.0, 0.02% NaN₃ was prepared.

The plates were made by placing 200 μ L of each SA-glyceride solution and 100 μ L of buffer or of the 180 μ M dansyl solution in each well [final concentrations: 100 μ M SA, 90 μ M glyceride, 99 μ M AF, 60 μ M DNSA, 4.5% (vol/vol) THF]. Each plate contained a column of indicator alone and a column of indicator and SA as controls. Eight replicates were performed for each glyceride-SA combination with either DNSA/AF or only AF and for the controls.

The parameters for the reading of the well plates can be found in [SI Appendix, Table S3](#). The fluorescence data were normalized by first subtracting the emission of the control (indicator alone) from each data point, and then the data set was rescaled from 0 to 100.

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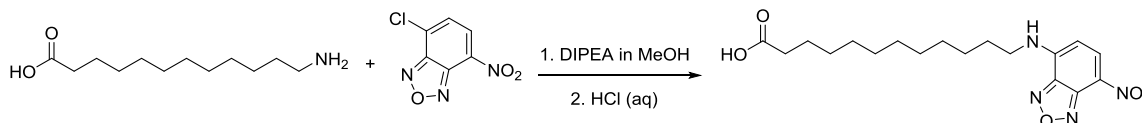
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Supplementary Information

General

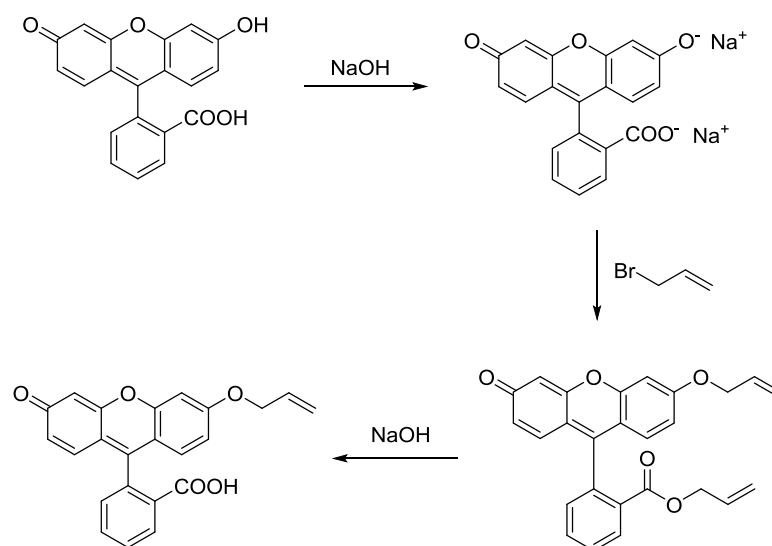
Human and bovine serum albumin were purchased fatty acid free from Sigma Aldrich. Glycerides were purchased from Nu-Chek Prep (**D1**, **D2**, **D4**), Sigma Aldrich (**T1**, **T2**, **T4**, **T8**, **D4**, **D7**), ABCR (**T3**, **T5**, **T6**, **M1**, **M2**, **M3**, **M4**, **M5**), Acros (**T7**), Bachem (**D6**), and Zerenex (**D5**). Grubbs Second Generation Catalyst and 1,8-anilinonaphthalene sulfonic acid was purchased from Sigma Aldrich. Dansyl amide and 2-anthracene carboxylate were purchased from TCI America. The cuvette experiments were performed with a PTI fluorimeter using an 814 photomultiplier detection system and a 75W xenon short arc lamp. NMR spectra were taken on a 500 MHz Varian NMR. LC-MS analysis was performed on an Agilent 1200 Series HPLC with an Agilent 6130 single quadrupole mass spectrometer (ESI and APCI ionization).

Synthesis of NBD-FA



To 100 mL methanol was added 12-aminododecanoic acid (5.61 mmol, 1.2 g) and Hünig's base (10.6 mmol, 1.85 mL). Then, 7-chloro-4-nitrobenzo-3-oxa-1,3-diazole (5.01 mmol, 1.0 g) was dissolved in 50 mL of methanol. This solution was added to the first one, and the reaction was stirred overnight at room temperature. The reaction slowly darkened from yellow to a brownish-red color. The methanol was removed by rotary evaporation leaving an oil. The oil was taken up in dichloromethane and washed several times with aqueous 1M HCl. Finally, the solvent was removed by rotary evaporation to yield a reddish-brown powder (1.23 g, 65% yield). No further purification was necessary. ¹H-NMR (400 MHz, CDCl₃, ppm): δ 8.50 (d, 1H, H-Ar), 6.37 (m, 1H, H-N), 6.17 (d, 1H, H-Ar), 3.48 (dt, 2H, -CH₂-NH₂), 2.35 (t, 2H, -CH₂-COOH), 1.8-1.25 (m, 18H, 9 -CH₂). MS (ESI): 377.2 (M-1).

Synthesis of allyl fluorescein

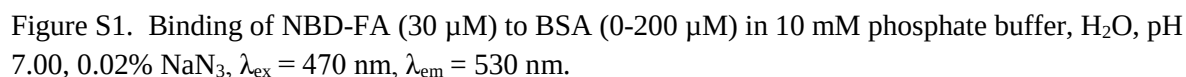


Fluorescein (4 g, 12.0 mmol) and sodium hydroxide (1 g, 12.0 mmol) were dissolved in 50 mL of water to yield the disodium salt. The water was removed by lyophilization or centrifugal evaporation. The disodium salt was dissolved in 150 mL water/80 mL acetone. Allyl bromide (2.4 mL, 27.6 mmol) was added to the reaction, and the reaction was refluxed for 2 hr. The reaction mixture was poured into 150 mL of water, and this solution was basified to yield an orange precipitate. A red oil remained in the reaction flask. This oil was dissolved in acetone and then poured into 100 mL of water. This solution was also basified to form the orange precipitate. The two orange precipitates were collected by filtration and combined. The fluorescein diether was recrystallized in ethanol/dilute NaOH (aq). Finally, the diether was dissolved in 30 mL acetone and heated to reflux. Then 20 mL of 1 M NaOH (aq) was added, and the reflux was continued for 12 min. The reaction mixture was poured into 100 mL of water, and this solution was acidified, becoming yellow and cloudy. The solution was extracted 3x with dichloromethane. The dichloromethane was dried with magnesium sulfate and removed by rotary evaporation. The resulting product was a yellow powder (357 mg, 8% yield). ¹H-NMR (400 MHz, CDCl₃, ppm): δ 8.00 (dt, 1H, H-Ar-COOH), 7.65 (td, 1H, H-Ar-COOH), 7.60 (td, 1H, H-Ar-COOH), 7.15 (dt, 1H, H-Ar-COOH), 6.76 (d, 1H, H-Ar), 6.71 (d, 1H, H-Ar), 6.67-6.50 (m, 6H, H-Ar), 6.02 (m, 1H, CH₂-CH=CH₂), 5.40 (dq, 1H, geminal cis-CH₂-CH=CH₂), 5.30 (dq, 1H, geminal trans-CH₂-CH=CH₂), 4.55 (dt, 2H, CH₂-CH=CH₂). MS (ESI): 373.2 (M+1).

Synthesis of 1

In 8 mL of CHCl₃ were dissolved 20 mg of monoerucin (0.0484 mmol), 20 mg of AF (0.0540 mmol), and 10.2 mg G2 catalyst (0.0120 mmol). The reaction was heated to 50°C for 2 hr. The reaction was washed twice with water. The CHCl₃ layers were collected, and the solvent was removed by rotary evaporation. The residue was dissolved in 1 mL of acetonitrile and purified by RP-HPLC. The method was 5-95% acetonitrile (0.1% TFA) for 25 min and then holding at 95% acetonitrile for 15 min. A C18 Luna column (10 micron, 100 Å, 250 x 21.2 mm) was used. The product (**1**) eluted at 36 min. The product was dried on high vacuum overnight: 3.6 mg (1.5% yield).

Indicator Uptake Experiments with Serum Albumin and Glyceride



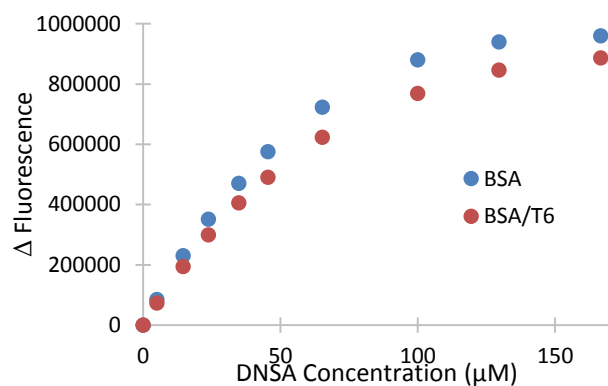


Figure S2. DNSA (0-170 μM) uptake by BSA (100 μM) in the absence and presence of trimyristin **T6** (90 μM) in 10 mM phosphate buffer, H_2O , pH 7.00, 0.02% NaN_3 , <0.3% THF, $\lambda_{\text{ex}} = 350 \text{ nm}$, $\lambda_{\text{em}} = 490 \text{ nm}$.

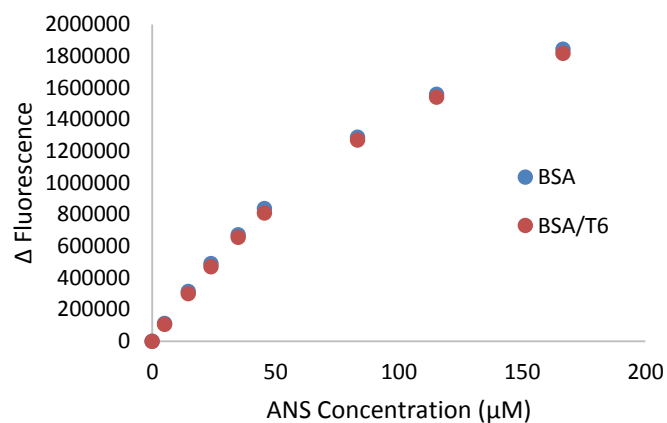


Figure S3. ANS (0-170 μM) uptake by BSA (100 μM) in the absence and presence of trimyristin **T6** (90 μM) in 10 mM phosphate buffer, H_2O , pH 7.00, 0.02% NaN_3 , <0.3% THF, $\lambda_{\text{ex}} = 400 \text{ nm}$, $\lambda_{\text{em}} = 475 \text{ nm}$.

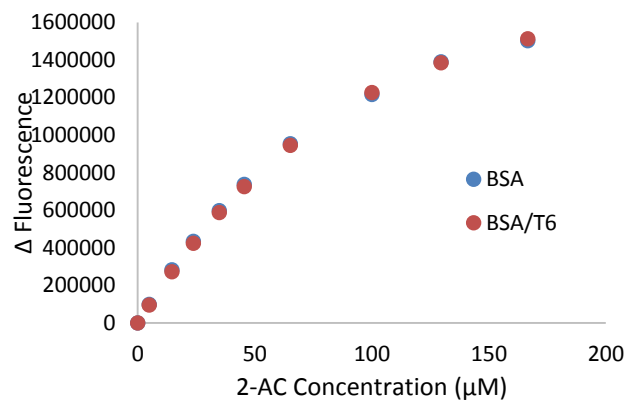


Figure S4. AC (0-170 μM) uptake by BSA (100 μM) in the absence and presence of trimyristin T6 (90 μM) in 10 mM phosphate buffer, H_2O , pH 7.00, 0.02% NaN_3 , <0.3% THF, $\lambda_{\text{ex}} = 386 \text{ nm}$, $\lambda_{\text{em}} = 422 \text{ nm}$.

Glyceride Array (Part 1)

Table S1. Plate read parameters

	NBD-FA	DNSA	ANS
λ_{ex}	485/20	360/40	380/20
λ_{em}	540/25	485/20	485/20
Mirror	Top 510	Top 400	Top 400

Olefin metathesis of glycerides

Table S2. Reaction conditions for metathesis of monoerucin

Reaction	Solvent	Temperature (°C)	Reaction time	M1 (equiv)	AF (equiv)	G2 (mol%)	CuI (mol%)	Reaction success
1	CDCl ₃	55, vacuum	24 hr	1.1	1.0	0.1	NA	Yes
2	CHCl ₃	55	24 hr	1.1	1.0	0.1	NA	No
3	THF	50	24hr	1.1	1.0	0.1	NA	No
4	DCE	70	2 hr	1.0	15	1.5	NA	Yes
5	DCE	75	24 hr	1.0	15	2.0	3.0	Yes
6	DCE	75	24 hr	1.0	5.0	20	25	Yes
7	CHCl ₃	50	2 hr	1.0	2.0	20	25	Yes
8	CHCl ₃	50	2 hr	1.0	1.1	20	30	Yes
9	CHCl ₃	50	6 hr	2.0	1.1	50*	NA	Yes
10	CHCl ₃	50	2 hrs	3.0	1.0	20*	NA	Yes
11	CHCl ₃	50	24 hrs	1.0	1.1, added after 2 hrs	20*	NA	No
12	CHCl ₃	50	2 hr	1.0, prepared from a THF stock of M1	1.1	20*	NA	No
13	CHCl ₃	50	2 hr	1.0	1.1	20*	NA	Yes

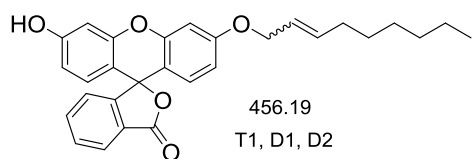
*New batch of catalyst was used from Sigma

Notes: Although the reaction also worked in DCE, chloroform was preferred as the solvent because it's nonflammable and hence can be heated more safely in a plastic well plate. The addition of copper provided no significant improvement in the reaction. The origin of reaction 2's failure is unclear. THF was not used as the solvent during the reaction, but the reagents were first dissolved in THF, and then the THF was removed by rotary evaporation. Perhaps the presence of trace amounts of THF was enough to affect the reaction. Using more M1 compared to AF did improve the yield of the M1-AF mixed product/s, but we did not want to use more material than necessary in the array. Pre-reacting M1 with the catalyst before adding AF hindered the formation of any M1-AF mixed products.

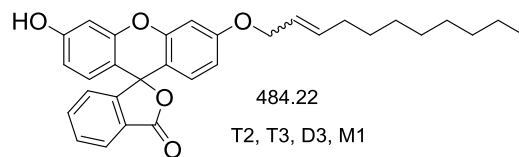
There were some issues with repeating the metathesis reaction with an old batch of catalyst. Additionally, the presence of even very small amounts (<10%) of THF appeared to hinder or entirely halt the reaction from occurring in some cases. Once the final satisfactory reaction conditions were identified (13 in the Table S2), the metathesis reaction was tested several times with all of the unsaturated glycerides in order to ensure reproducible conversions (data in Table 1).

Table S3. Summary of the olefin metathesis reaction for the unsaturated glycerides and **AF**

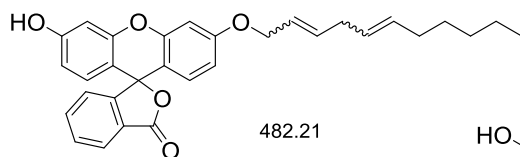
Glyceride	Cis or Trans	m/z (M+1) of mixed metathesis product	Percent conversion of AF to mixed product (%)
T1- Tripalmitolein	Cis	457.2	68
T2- 1,3-Dioleoyl-2-palmitoyl glycerol	Cis	485.2	52
T3- Trielaidin	Trans	485.2	65
T4- Tripetroselaidin	Trans	527.3	54
T5- Trilinolein	Cis	443.2	49
	Cis	483.2	23
D1- 1,3-Dipalmitelaidin	Trans	457.2	55
D2- 1,3-Dipalmitolein	Cis	457.2	56
D3- 1,2-Diolein	Cis	485.2	59
D5- 1,3-Diarachidonin	Cis	443.2	33
	Cis	483.2	16
	Cis	523.2	9
	Cis	563.3	10
M1- Monoerucin	Cis	485.2	8



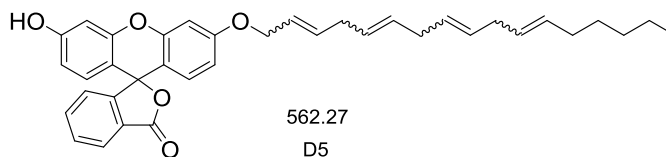
456.19
T1, D1, D2



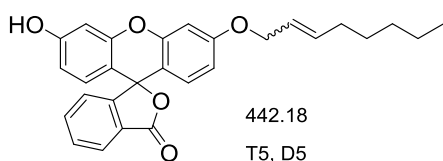
484.22
T2, T3, D3, M1



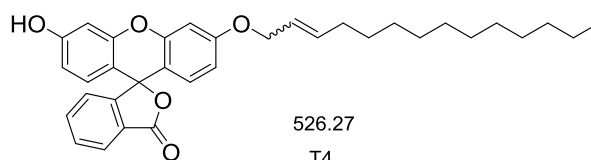
482.21
T5, D5



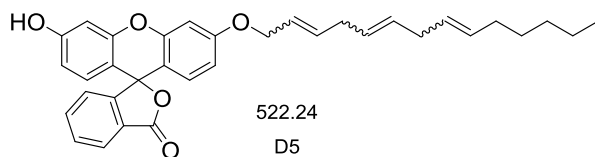
562.27
D5



442.18
T5, D5



526.27
T4



522.24
D5

Structures and m/z of mixed metathesis products for each unsaturated glyceride.

Exploring FRET with Dansyl Amide and Allyl Fluorescein

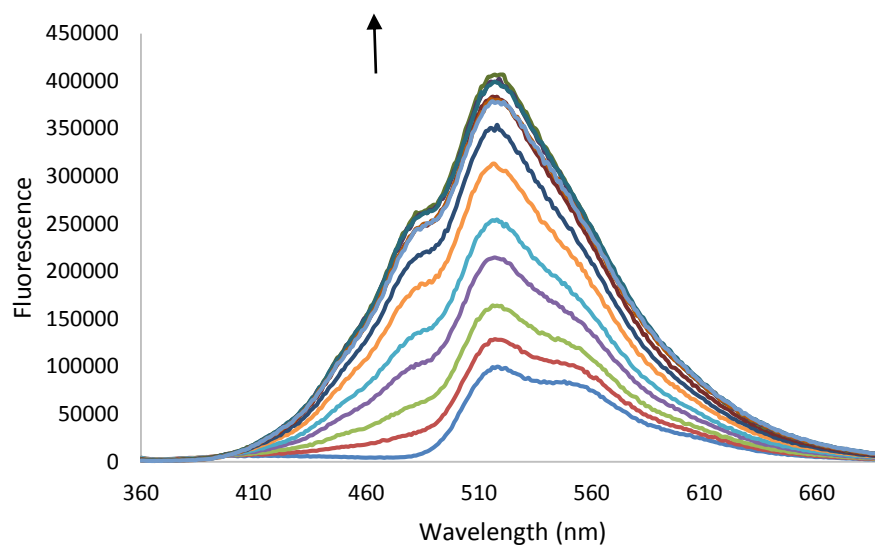


Figure S5. DNSA (0-200 μM) uptake by BSA/1 (100 μM /90 μM) in 10 mM phosphate buffer, H₂O, pH 7.00, 0.02% NaN₃, <0.3% THF, $\lambda_{\text{ex}} = 350$ nm.

Glyceride Array with Metathesized Glycerides (Part 2)

Table S4. Plate read parameters

	NBD-FA	DNSA1	ANS	AF1/2	FRET	DNSA2
λ_{ex}	485/20	360/40	380/20	485/20	360/40	360/40
λ_{em}	540/25	485/20	485/20	528/40	528/40	485/20
Mirror	Top 510	Top 400	Top 400	Top 510	Top 510	Top 400

Array Reproducibility

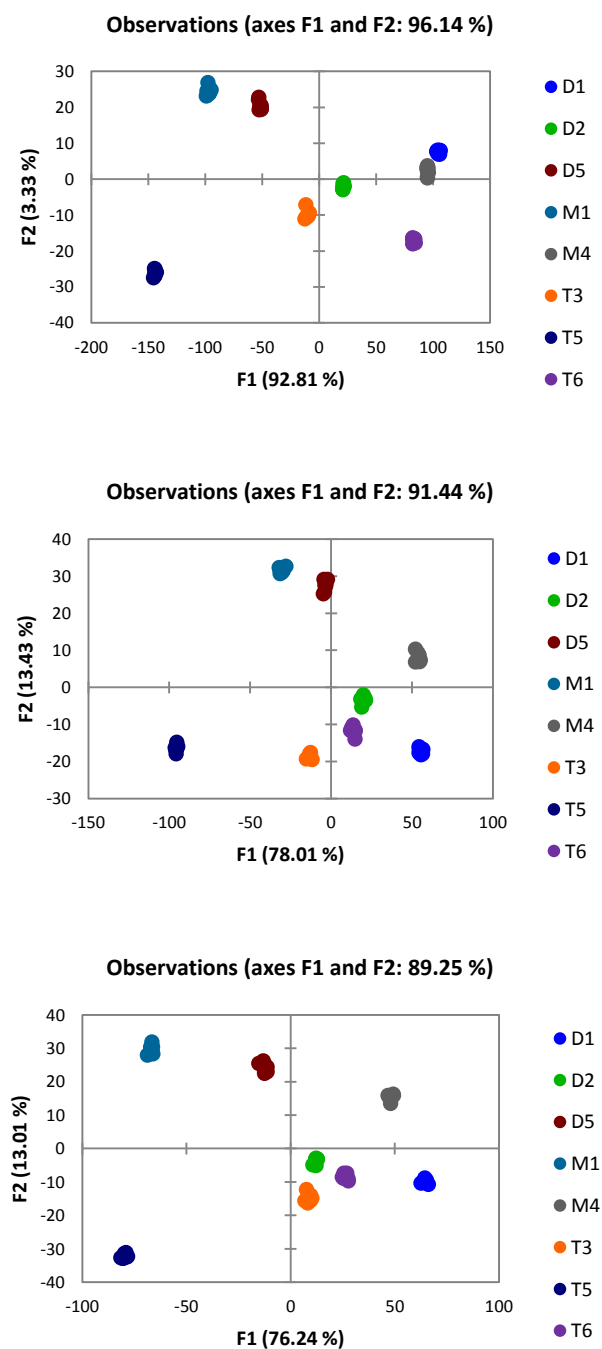


Figure S6. LDA plots of data collected from 96-well plates in three independent repetitions: BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), and NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with < 5% THF (see Table S3 for read parameters). The fluorescence counts were express as a percentage of the total fluorescence for each receptor/variable.

Table S5. Summary of prediction accuracy for the six combinations of the three repetitions in Figure S6

Training Set	Prediction Set	% Correct Classification (of 64 cases)
Rep 1	Rep 2	75
Rep 1	Rep 3	84
Rep 2	Rep 1	80
Rep 2	Rep 3	100
Rep 3	Rep 1	84
Rep3	Rep 2	100

Analysis of the Full Glyceride Panel

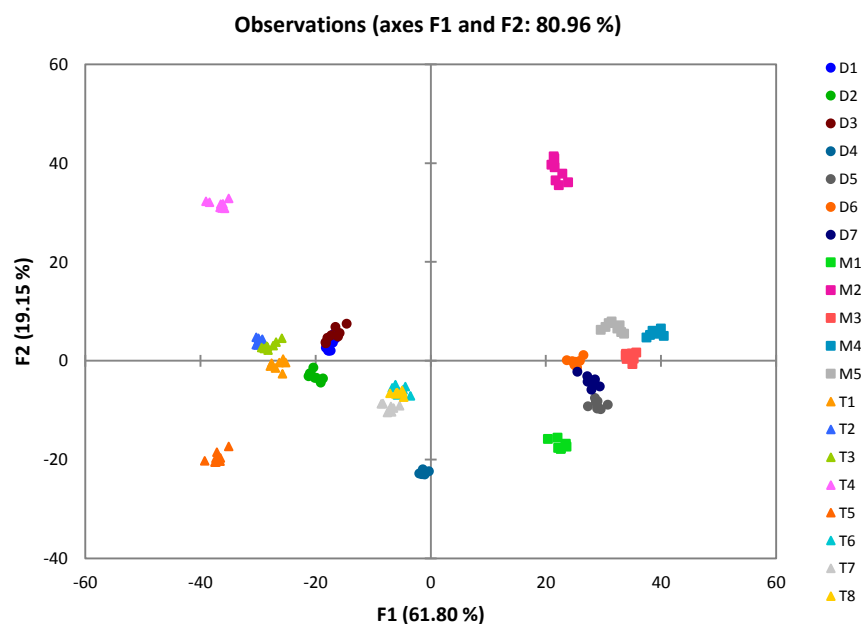


Figure S7. LDA plot of data collected from 96-well plates in an independent repetition of the array on all twenty glycerides. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with < 5% THF (see Table S3 for read parameters). Cross validation: 98%

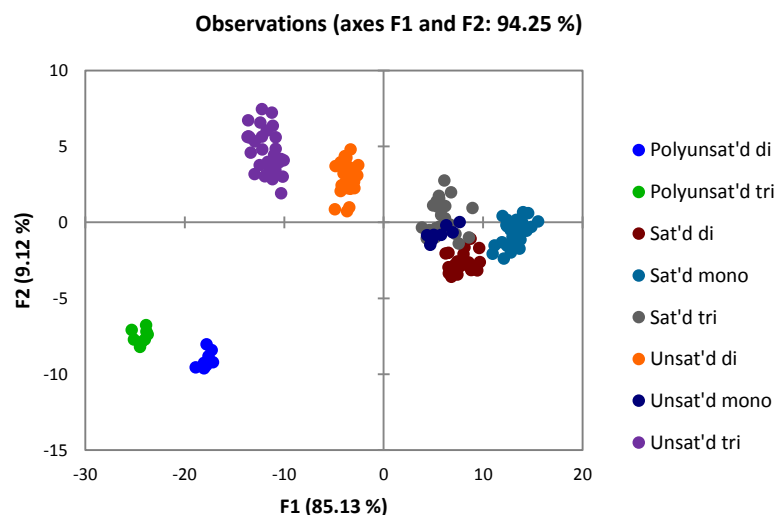


Figure S8. LDA plot of second training set of glycerides with analyte classes designated as structural features. The array components consisted of BSA and HSA (100 uM), glyceride (90 uM), DNSA (60 uM), ANS (60 uM), NBD-FA (60 uM), metathesized glyceride (90 uM), AF (100 uM), and DNSA (60 uM) in phosphate buffer with < 5% THF (see Table S3 for read parameters). The fluorescence counts were express as a percentage of the total fluorescence for each receptor/variable. Cross validation: 98%.

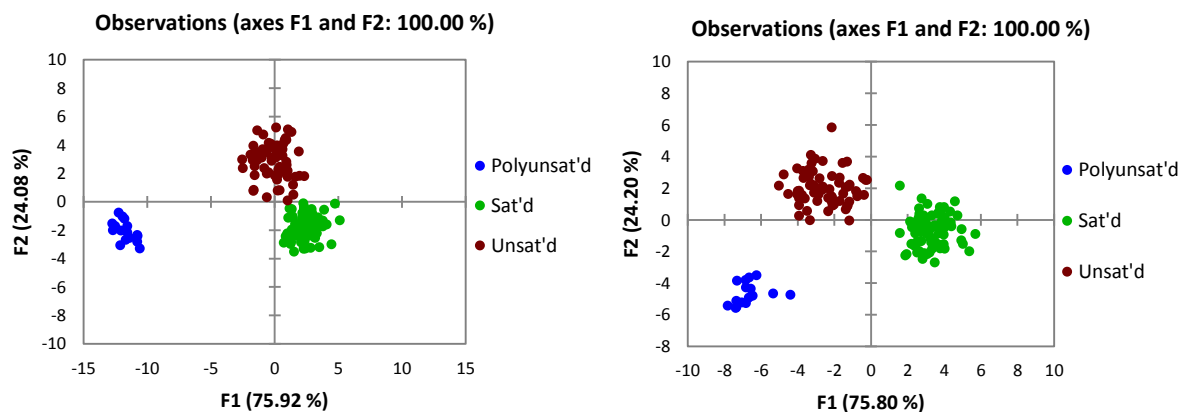


Figure S9. LDA plots of the training sets of glycerides with analyte classes designated as saturated, unsaturated, and polyunsaturated. The array components consisted of BSA and HSA (100 uM), glyceride (90 uM), DNSA (60 uM), ANS (60 uM), NBD-FA (60 uM), metathesized glyceride (90 uM), AF (100 uM), and DNSA (60 uM) in phosphate buffer with < 5% THF (see Table S3 for read parameters). The fluorescence counts were express as a percentage of the total fluorescence for each receptor/variable. Cross validation: 99% (left) and 100% (right).

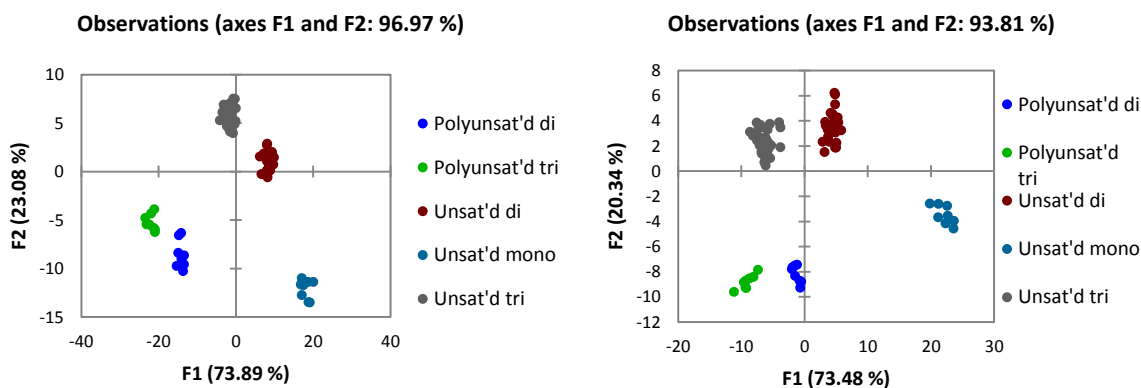


Figure S10. LDA plots of the training sets of unsaturated glycerides with analyte classes designated as shown in the key. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with < 5% THF (see Table S3 for read parameters). The fluorescence counts were express as a percentage of the total fluorescence for each receptor/variable. Cross validation: 100% (left) and 100% (right).

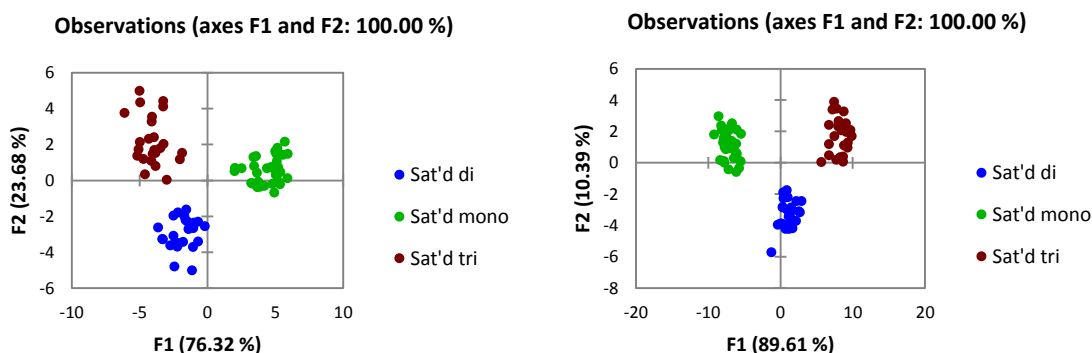


Figure S11. LDA plots of the training sets of saturated glycerides with analyte classes designated as shown in the key. The array components consisted of BSA and HSA (100 μ M), glyceride (90 μ M), DNSA (60 μ M), ANS (60 μ M), NBD-FA (60 μ M), metathesized glyceride (90 μ M), AF (100 μ M), and DNSA (60 μ M) in phosphate buffer with < 5% THF (see Table S3 for read parameters). The fluorescence counts were express as a percentage of the total fluorescence for each receptor/variable. Cross validation: 100% (left) and 100% (right).

Performing assay on glyceride mixtures

Mixtures of triglycerides were prepared by weighing out each component in a vial, dissolving each in chloroform, and consolidating the component solutions into a single weighed vial. The chloroform was removed by rotary evaporation.

Table S6. Composition of glyceride mixtures by weight

Mixture	% T1 by weight	% T3 by weight	% T5 by weight	% T6 by weight	% T7 by weight	Total weight of glyceride material (mg)
A	11.61	26.47	22.13	16.01	23.79	17.49
B	7.74	17.95	31.63	24.00	18.67	34.87
C	0	16.34	52.81	30.86	0	23.69
D	0	32.37	41.88	25.75	0	25.24

SANAS

Mixtures of triglycerides were prepared, and a fixed amount was used for each analyte solution/metathesis reaction well. To the constant quantity of mixture in each vial/metathesis reaction well was added increasing amounts of pure **T5** as described in the text above. These solutions were then subjected to the same conditions as described for parts 1 and 2 of the array. The final concentration of the glyceride mixture in the well plates of the array is given in Table 2 as well as the final concentration within that mixture of **T5**. The content of the other triglyceride components is detailed in Table S5. For the additions, the final amount of **T5** added was 0, 0.005, 0.01, 0.02, 0.03, 0.04, 0.05, and 0.06 mg/mL for mixtures A and B and 0, 0.003, 0.005, 0.01, 0.015, 0.02, 0.025, and 0.03 mg/mL for mixtures C and D. The raw fluorescence data was normalized by rescaling within each variable from values 0-100 using XLSTAT. The eight replicates collected for each addition were averaged after normalization, and this averaged data matrix was subjected to the SANAS calculation in MATLAB.

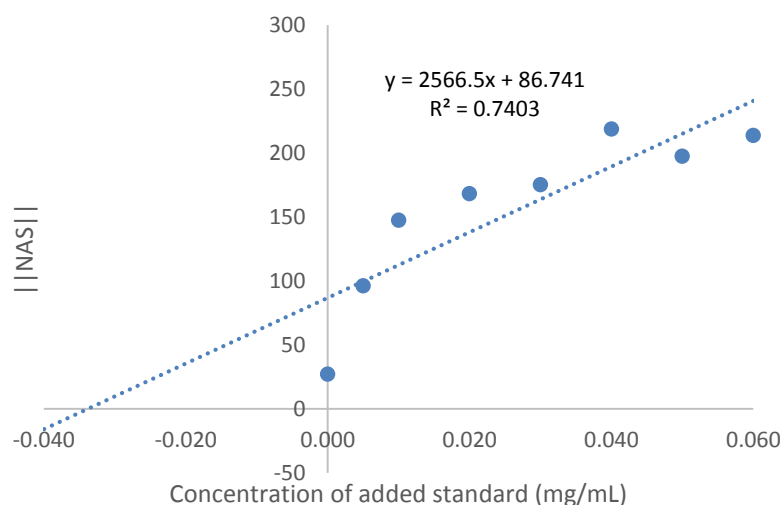


Figure S12. SANAS plot generated from the standard addition of mixture A.

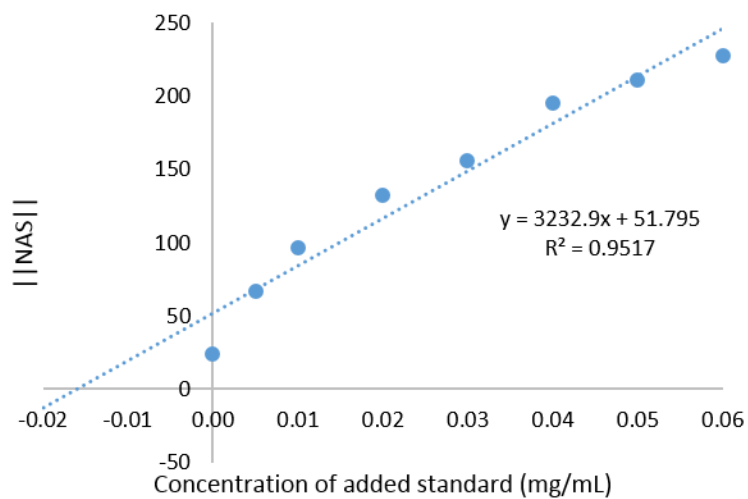


Figure S13. SANAS plot generated from the standard addition of mixture B.

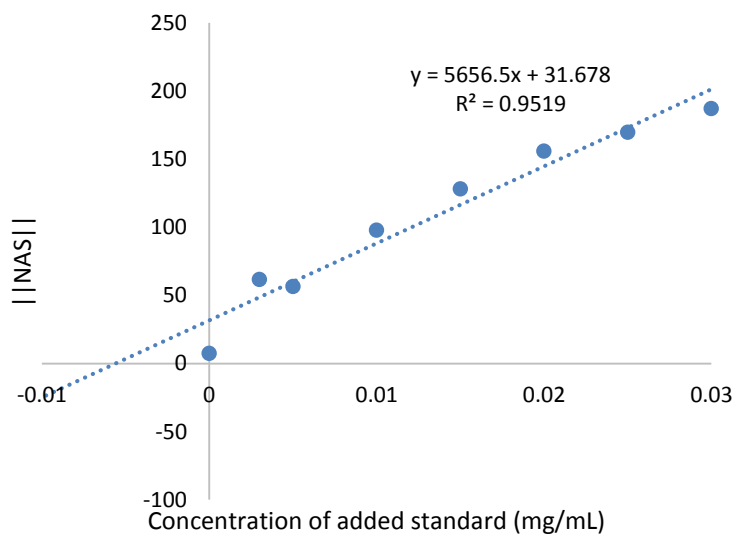


Figure S14. SANAS plot generated from the standard addition of mixture C.