



Article

Size-resolved identification, characterization and quantification of primary biological organic aerosol at a European rural site

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26	Abstract
27	Primary biological organic aerosols (PBOA) represent a major component of the coarse
28	organic matter (OM _{COARSE} , aerodynamic diameter >2.5µm). Although this fraction affects

Primary biological organic aerosols (PBOA) represent a major component of the coarse organic matter (OM_{COARSE}, aerodynamic diameter >2.5μm). Although this fraction affects human health and climate, its quantification and chemical characterization currently remain elusive. We present the first quantification of the entire PBOA_{COARSE} mass and its main sources by analyzing size-segregated filter samples collected during summer and winter at the rural site of Payerne (Switzerland), representing a continental Europe background environment. The size-segregated water soluble OM was analyzed by a newly developed offline aerosol mass spectrometric technique (AMS). Collected spectra were analyzed by 3-dimensional positive matrix factorization (3D-PMF), showing that PBOA represented the main OM_{COARSE} source during summer and its contribution to PM₁₀ was comparable to that of secondary organic aerosol. We found substantial cellulose contributions to OM_{COARSE}, which in combination with gas chromatography mass spectrometry molecular markers quantification, underlined the predominance of plant debris. Quantitative polymerase chain reaction (qPCR) analysis instead

- 40 revealed that the sum of bacterial and fungal spores mass represented only a minor OM_{COARSE}
- 41 fraction (<0.1%). X-ray photoelectron spectroscopic (XPS) analysis of C and N binding energies
- 42 throughout the size fractions revealed an organic N increase in the PM₁₀ compared to PM₁
- 43 consistent with AMS observations.

Introduction

- 45 Primary biological organic aerosol (PBOA) is a major source of coarse aerosol organic matter
- 46 (OM). The detection of these particles has been the subject of studies for one and a half
- 47 centuries.¹⁻³ Studies⁴ have related single PBOA components to adverse health effects,⁵ and
- 48 revealed their important role as ice and cloud condensation nuclei. 6-10 Emissions of primary
- 49 biological particles (PBAP) are estimated to be among the largest contributors of pre-industrial
- organic aerosols, 11 therefore a precise estimate of their sources is also important for the
- 51 development of accurate climate models.⁴ Nevertheless, PBOA characterization and
- quantification has received less attention than other types of aerosol sources and processes (e.g.
- traffic, mineral dust, sulfate, wood combustion and secondary organic aerosol), possibly because
- of technical limitations hindering the understanding of the sources and composition of this
- 55 fraction.
- 56 Traditional analytical techniques for the PBOA characterization include optical microscopy,
- 57 cultivation of specific viable bacteria, fungi and algae and fluorescence microscopy for the
- 58 quantification of functionalized or autoflorescent specific components.⁴ More recent approaches
- 59 are classified into molecular techniques (e.g. chemical tracers determination, nucleic acids
- extraction and amplification), optical techniques (fluorescent and Raman spectroscopy), and non-
- optical techniques. Fluorescence techniques are of particular relevance because biological

materials contain fluorophores. 12,13 Non-optical approaches include different types of mass
spectrometers; among these, we note the recent use of online-aerosol mass spectrometry (AMS)
for the study of the submicron fraction. 14-16
Despite the vast literature focusing on the quantification of individual PBOA components, the
quantification of the total PBOA mass and the main processes by which this fraction enters the
atmosphere remains elusive. As a consequence, the International Panel on Climate Change
2013 ¹⁷ reported the global terrestrial PBOA emission to range between 50 and 1000 Tg/yr,
highlighting the large gap in our knowledge about this fraction. Within this fraction, 28 Tg/yr
were estimated to comprise fungal spore emissions using arabitol and mannitol as tracers. ¹⁸ The
use of these compounds as specific fungal spores tracers is still subject of discussion in the
scientific community 19,20 and there is a general indispensable need for the determination of
PBOA concentrations and major emission processes through size-resolved field observations
against which the global models can be evaluated.
In this study, we present the first quantification of the total water-soluble PBOA (WSPBOA)
mass using an offline Aerodyne Time-of-Flight Aerosol Mass Spectrometer (ToF-AMS). The
analysis was performed on PM_1 , $PM_{2.5}$ and PM_{10} (particulate matter with an aerodynamic
diameter $<$ 1, 2.5 and 10 $\mu m)$ filter samples collected concomitantly at the rural site of Payerne,
Switzerland. WSPBOA quantification was achieved by 3-dimensional positive matrix
factorization analysis (3D-PMF) of water soluble OA mass spectra, following the recently
developed methodology described by Daellenbach. ²¹ In comparison with previous PBOA online
AMS observations, 14-16 the filter samples water extraction step enabled accessing the
WSOM _{COARSE} fraction. For the characterization of the main PBOA sources, the dataset was
complemented with an unprecedented combination of measurements, including enzymatic

cellulose determination, quantification of bacterial and fungal spore DNA via quantitative polymerase chain reaction (qPCR), and gas chromatography mass spectrometry analysis (GC-MS) of organic molecular markers. In this study, we discuss the quantification of the total PBOA mass via 3D-PMF, the quantification of its major components and their possible usage as PBOA tracers including bacteria and fungal spores measured via qPCR, plant debris estimate from *n*-alkanes measurements, and carbohydrates.

Material and Methods

Sample collection. We collected in total 87 24h-integrated aerosol samples (Batch A) on quartz fiber filters at the rural background site of Payerne during June-July 2012 and January-February 2013. Batch A included PM₁, PM_{2.5}, and PM₁₀ samples collected in parallel using three High-Volume samplers (Digitel DA-80H equipped with PM₁, PM_{2.5} and PM₁₀ size-selective inlets) operating at 500 L min⁻¹. In total 45 samples were collected during summer (15 samples per size fraction), and 42 during winter (14 samples per size fraction). Additionally, PM₁₀ filters were collected every fourth day throughout 2013 following the same procedure (Batch B). In the following, the subscript *coarse* will denote for a generic aerosol component, the fraction contained between 2.5 and 10 μm.

Aerosol characterization. An overview of the auxiliary analytical measurements can be found in Table 1, Table S2, and in the Supplementary Information (SI). In this section only offline-AMS, qPCR, and x-ray photoelectron spectroscopy (XPS) will be discussed in details.

Table 1. Supporting measurements

Measured variable	Batch A	Batch B

PM	Gravimetry	All filters	-
WSOM mass spectral fingerprint	Offline-AMS ²¹	All filters	All filters
EC/OC	Thermal Optical Transmittance using a Sunset Lab Analyzer ²² (EUSAAR2) ²³	All filters	-
ions	Ion Chromatography ²⁴	All filters	-
WSOC	Water extraction Thermal Decomposition ND-IR determination using TOC analyzer (SI)	All filters	-
Cellulose	Cellulose enzymatic conversion to D-glucose and photometric determination ²⁵	32 filters (9 summer PM ₁₀ filters, 4 winter PM ₁₀ , 5 summer PM _{2.5} , 9 summer PM ₁ , and 5 summer PM ₁)	-
molecular markers (Table S2)	In-Situ Derivatization Thermal Desorption Gas Chromatography Time-of- Flight Mass Spectrometry (IDTD-GC-MS) ²⁶	40 samples (15 summer PM ₁ , 15 summer PM ₁₀ , 5 winter PM ₁₀ , 5 winter PM ₁₀)	-
C1s, N1s Binding energies	X-Ray Photoelectron Spectroscopy	6 samples (3 summer PM ₁₀ , 3 summer PM ₁)	-
bacterial and fungal spore DNA	Quantitative Polymerase Chain Reaction genetic analysis ^{27,28}	58 samples (all summer PM ₁ , PM _{2.5} , and PM ₁₀ , all winter PM ₁ and PM ₁₀)	-
Carbohydrates (Table S2)	IC coupled to a Pulsed Amperometric Detector (IC-PAD) ²⁹	All samples	-

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Offline-AMS. The Offline-AMS analysis entails an extraction of two 16 mm diameter punches per sample in 10 mL of ultrapure water (18.2 M Ω cm, Total Organic Carbon < 5 ppb) via ultrasonication for 20 min at 30°C. Liquid extracts were subsequently homogenized for 40 s using a

vortex mixer and then filtered through 0.45 µm nylon membrane syringe filters. Filtered extracts
were aerosolized and the generated particles were dried using a silica gel diffusion drier before
measurement by HR-ToF-AMS. 30 On average 10 mass spectra (60 s each) of the bulk WSOM
were collected per extract. Before each sample measurement, 5 blank mass spectra were
collected by nebulizing ultrapure water, and their average was subtracted from the corresponding
individual sample mass spectra. The signal of field blank samples measured following the same
procedure was statistically not different from the ultrapure water mass spectra.
XPS. XPS analysis enabled monitoring the binding energies (BE) of C, S and N, providing
insight into their oxidation state (typically higher BE are related to higher oxidation numbers),
and thereby quantifying the organic N (N_{org}) mass through the size fractions. The same analysis
was conducted on 3 field blanks and on N-containing surrogate standards deposited on blank
quartz fiber filters. Tested standards included $NaNO_3$ and $(NH_4)_2SO_4$ for the characterization of
the most abundant forms of inorganic N, while horseradish peroxidase and chloroperoxidase
from caldariomyces fumago were used as surrogates for amine and amide containing proteins in
PBOA. Signal identification and integration proceeded as follows. The obtained spectra were
first aligned with a two-point BE calibration using the Si_{2p} and the O_{1s} peaks deriving from the
quartz fiber filters as reference points. We estimated an energy accuracy of 0.3 eV, and an
average fitting error of 1.4% by fitting the signals of replicate measurements of standard
compounds and blanks and assuming a single Gaussian peak for each atom. These parameters

were then used for the fitting of the blank-subtracted C_{1s} , and N_{1s} signals in environmental

samples, which consisted of several peaks from different chemical components. The number of

these peaks was determined such that fitting residuals (fraction of signal) equaled the fitting

errors determined from the fitting of single compounds. The N_{1s} peak widths were constrained to

be equal to the one derived from $(NH_4)_2SO_4$ standard, while the C_{1s} peak width was determined
from blank filters. From the analysis of standard $(NH_4)_2SO_4$ we derived an average N_{1s}/S_{2p} ratio
of 0.80±0.02, which was used to estimate the N_{1s} contribution from $(NH_4)_2SO_4$ $(N_{1s(NH_4)_2SO_4})$.
This contribution was fixed in proportion of that of S_{2p} using the aforementioned $N_{1s}\!/S_{2p}$ ratio
and N_{1s} peak width. This estimate neglected the contribution from organic or non-(NH ₄) ₂ SO ₄
sulfate. The uncertainty on the $N_{1s(NH_4)_2SO_4}$ area was estimated based on the integration of the
S_{2p} peak. N_{1s} fitting sensitivity analysis was performed by varying the $N_{1s(NH_4)_2SO_4}$ peak position
and area within our uncertainties. Only fittings of $N_{1s(NH_4)_2SO_4}$ with residuals lower than our
errors were retained.
qPCR. We performed a qPCR analysis in order to quantify total bacterial and fungal spore DNA.
DNA extraction was conducted following the procedure presented in the SI and specific
universal primers (Table S3) were selected for total DNA quantification of bacterial and fungal
spores. The extracted DNA was amplified using the qPCR technique described in Lang-
Yona. ^{27,28} The total number of bacterial cells and fungal spores was estimated assuming a DNA
content of $4.74 \cdot 10^{-3}$ pg per bacterial cell and $3 \cdot 10^{-2}$ pg per fungal spore respectively, based on the
Escherichia coli and Aspergillus fumigatus genome lengths (4,639,221 bp and 29,384,958 bp,
respectively). 31 Total bacterial mass was estimated for PM_1 and PM_{10} samples assuming as a
reference the dry and wet <i>E. coli</i> cell weights (3.10^{-13}) and 1.10^{-12} g, respectively), 32 while total
fungal spores mass was based on the A. fumigatus spore weight of 2.9·10 ⁻¹² g. ³³

3D-PMF

OA mass spectra collected by offline-AMS were analyzed using 3D-PMF to apportion the timedependent size-segregated (PM₁, PM_{2.5}, PM₁₀) contributions of the water soluble organic sources.³⁴ We adopted a vector-matrix approach,³⁵ also known as "Tucker1" approach³⁶ in which we assumed constant mass spectra throughout the size fractions. The 3D-PMF algorithm describes the variability of the multivariate data-matrix (x) as the linear combination of static factor profiles (f) and their corresponding time and size-dependent contributions (g), such that

$$x_{i,j,k} = \sum_{z=1}^{p} g_{i,j,z} \cdot f_{z,k} + e_{i,j,k}$$
 (1)

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- Here, $x_{i,j,k}$ denotes an element of the data matrix, while subscripts i, j and k represent time, size and organic ions (250 fitted organic ions in the range m/z 12 to 115) respectively. The subscripts p and z indicate the total number of factors selected by the user, and a discrete factor number $(1 \le z \le p)$ respectively, while $e_{i,i,k}$ represents an element of the residual matrix.
- PMF was solved using the multi-linear engine algorithm (ME-2)^{37,38} (using the source finder, SoFi)³⁸ which enabled an efficient exploration of the rotational ambiguity by directing the solution toward environmentally relevant rotations. This was achieved by a-priori constraining $f_{z,k}$ and/or $g_{i,j,z}$ elements, and allowing the constrained elements to vary within a predetermined range defined by a scalar a, such that the returned $f_{z,k}$ or $g_{i,j,z}$ values satisfy eq 2.

$$f_{z,k}' = f_{z,k} \pm a \cdot f_{z,k} \tag{2}$$

- Here we constrained the f matrix elements for only one factor, related to hydrocarbon-like organic aerosol (HOA) from traffic³⁹ (SI).
- PMF data and error input matrices (*x* and *s*) were constructed including ten mass spectral repetitions per filter sample. Data and error matrices were rescaled to WSOM_i in order to compare source apportionment results with external tracers. WSOM_i concentrations were estimated from the WSOC_i measurements multiplied by the OM/OC_i ratios determined from offline-AMS HR analysis (measured OM/OC_i distribution 1st quartile 1.89, 3rd quartile 2.01).⁴⁰ In

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 $(\sigma_{WSOC.i}).$

total, the 3D-PMF input matrices comprised 87 samples corresponding to 29 filters per size fractions.

The error matrix elements $s_{i,j,k}$ were determined according to eq 3 by propagating the blank standard deviation $\sigma_{i,j,k}$ and the signal error $\delta_{i,j,k}$ accounting for electronic noise, ion-to-ion variability at the detector, and ion counting statistics.^{41,42}

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$$s_{i,j,k} = \sqrt{\delta_{i,j,k}^2 + \sigma_{i,j,k}^2}$$
 (3)

The optimization of the 3D-PMF results is thoroughly presented in the SI. Briefly, to improve the factor separation we up-weighted selected variables dividing their corresponding uncertainties by a scalar c (>1).⁴³ The sensitivity of model outputs to c and a-values was assessed and only solutions matching selected criteria were retained (SI). The variability of the results amongst the selected solutions was considered our best estimate of model errors.

PMF factor contributions to total OM were estimated after PMF analysis as:

$$ZOA_i = \frac{WSZOA_i}{R_z} \tag{4}$$

Here, [WSZOA] and [ZOA] denote for a generic Z source the concentration of the ambient water soluble organic aerosol and the total organic aerosol respectively, while R_z indicates the recovery efficiency for that source. In total, 5 OA factors were separated including HOA, summer oxygenated OA (S-OOA), winter oxygenated OA (W-OOA), biomass burning OA (BBOA), and primary biological OA (PBOA). The $R_{z,med}$ determined by Daellenbach²¹ were applied to all factors except for PBOA, whose recovery was not previously estimated. Accordingly, we shall report hereafter the concentration of WSPBOA and estimate the PBOA water solubility.

Source apportionment errors ($\sigma_{S.A.,Z,i}$) were estimated according to eq 5, which accounts for R_Z and rotational uncertainty ($\sigma_{PMF,RZ,i}$), measurement repeatability ($\sigma_{REF,i}$), and WSOM uncertainty

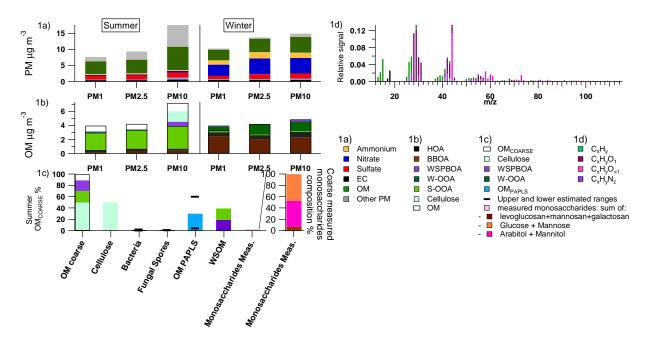
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$$\sigma_{S.A.,Z,i} = \sqrt{\sigma_{PMF,RZ,i}^2 + \sigma_{REP,Z,i}^2 + f_{Z,i}^2 \cdot \sigma_{WSOM,i}^2}$$
 (5)

Here f_Z denotes the relative contribution of the generic factor Z to WSOM. $\sigma_{WSOM,i}$ includes WSOC blank variability and measurement repeatability. The $\sigma_{PMF,RZ,i}$ term includes the variability of the rescaled PMF solutions and represents our best estimate of recovery errors and rotational ambiguity. The $\sigma_{REP,Z,i}$ term was considered as our best estimate of experimental repeatability/errors and represents the variability of PMF results for the measurements repetitions.

Results and Discussion

PM major components

A complete overview of the size-segregated chemical composition of winter and summer PM components is presented in Figure 1a. In the following, average and median values are indicated with the subscripts *avg* and *med*, respectively.

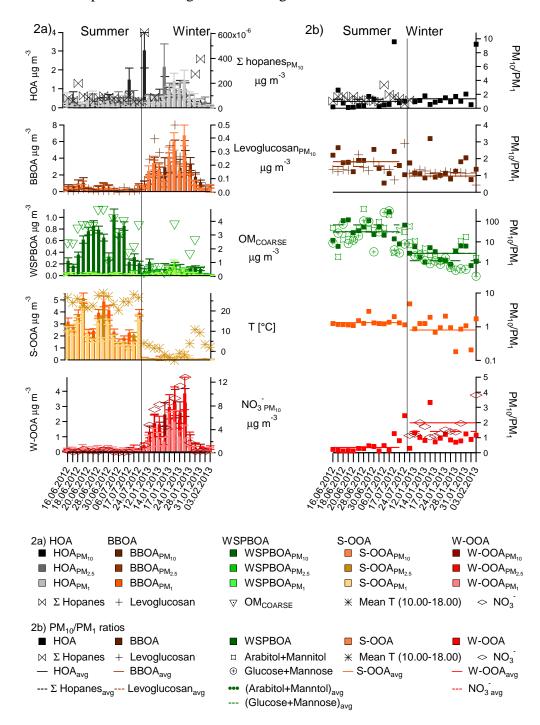


215 216 217 218 219 220 221	Figure 1. 1a) Seasonal PM chemical composition of the different size fractions. The OM _i estimate was calculated from OC _i measurements multiplied by the corresponding OM/OC _i retrieved from offline-AMS HR analysis. 1b) Average seasonal aerosol sources contributions to OM in the different size fractions. White are consistent with our estimate of the water insoluble PBOA fractions (Figure S8). 1c) Summer OM _{COARSE} major components. 1d) WSPBOA high resolution AMS mass spectrum.
222	OM represented a major component of PM during summer and winter. While during winter
223	large part of the OM_{10} (87%) was comprised in the $PM_{2.5}$ fraction, during summer this fraction
224	represented only 58%. In contrast, during summer secondary inorganic species (SO ₄ ²⁻ , NH ₄ ⁺ , and
225	NO ₃ ⁻) did not manifest a comparable increase in PM _{COARSE} (85% of the mass comprised in the
226	PM _{2.5} fraction) suggesting a small contribution of additional secondary aerosols in the coarse
227	fraction. Overall OM_{COARSE} accounted for 3 μg m ⁻³ avg during summer, and as will be shown in
228	the following, large part of this fraction constituted of PBOA (Figure S13).
229	Similarly to OM, dust likely from resuspension ⁴⁴ was enhanced in the coarse fraction
230	especially during summer. The upper limit for the inorganic dust _{COARSE} concentration was
231	estimated as the difference between inorganic PM_{10} and inorganic $PM_{2.5}$ ($PM_{COARSE,inorg}$), and
232	accounted for $31\%_{avg}$ during summer and $5\%_{avg}$ during winter, although this estimate can include
233	small sea salt contributions (SI). The obtained (Ca ²⁺ /PM) _{COARSE,inorg} value of 4.2% _{med} (1 st quartile
234	3.2%, 3 rd quartile 7.7%) was consistent with the ratios reported by Chow ⁴⁵ for 20 different dust
235	profiles (3.5±0.5%), and with values reported by Amato in Zürich. ⁴⁶ As a comparison, the total
236	OM_{COARSE} concentration represented 36% $_{avg}$ of PM_{COARSE} (8.4 μg m ⁻³), compared to the 62% $_{avg}$
237	for dust _{COARSE,inorg} .

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Size resolved OA source apportionment

In this section we present the validation of the 3D-PMF factors (HOA, BBOA, W-OOA, S-OOA, and WSPBOA) which enabled the quantification of WSPBOA. Average source apportionment results are presented in Figure 1b and Figure 2.



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244245246247	Figure 2 . 3D-PMF source apportionment results. 2a) Size fractional time series of PMF factors, corresponding tracers, and temperature. Error bars represent source apportionment uncertainty. 2b) Size fractional increase (PM ₁₀ /PM ₁) time series of PMF factors, and corresponding tracers.
248	3D-PMF factors were associated to aerosol sources or processes according to mass spectral
249	features, seasonal contributions, size fractional contributions, and correlation with tracers (Figure
250	2). Given the lack of widely accepted methodologies to estimate the uncertainty of PMF results,
251	in this work we considered $\sigma_{S.A.,k,i}$ (Methodology section) as our source apportionment
252	uncertainty, while the statistical significance of the factor contributions for each size fraction was
253	based on our best error estimation ($\sigma_{S.A.,k,i}$, Table S4).
254	HOA and BBOA contributions represented the only anthropogenic primary sources resolved in
255	Payerne. In particular, HOA correlated with hopanes present in lubricant oils with a R =0.54 (SI).
256	This correlation is also supported by the summer (HOA/EC) _{med} ratio (0.63 _{med}) being consistent
257	with other European studies reported by El Haddad and references therein. ⁴⁷ BBOA instead
258	correlated with levoglucosan produced by cellulose pyrolysis (R=0.94). A levoglucosan/BBOC
259	ratio of $0.18_{\rm med}$ was found, consistent with values reported (Huang and references therein 48) for
260	ambient BBOA observations. Both HOA and BBOA showed statistically significant
261	contributions ($>3\sigma$) only in the submicron fractions. The seasonal trend of these anthropogenic
262	factors was also significantly different: while the HOA (traffic) contribution was relatively stable
263	and small across the year, BBOA showed a strong seasonality, rising from $6\%_{avg}$ of OM_1 during
264	summer to 73% _{avg} during winter.
265	Two OOA factors characterized by high $\mathrm{CO_2}^+$ contributions were separated according to their
266	different seasonal trends. While W-OOA showed a strong correlation with NO ₃ (R=0.94), S-
267	OOA showed a positive non-linear correlation with temperature, following the behavior of
268	biogenic volatile organic compounds emissions. 49 The relative contribution of W-OOA to OM1

rose from $5\%_{avg}$ during summer to $22\%_{avg}$ during winter, while the S-OOA contribution to OM_1
decreased from $59\%_{avg}$ during summer to $4\%_{avg}$ during winter. W-OOA was the only factor
significantly contributing (within $3\sigma)$ to OM in the size range 1-2.5 μm (48% $_{avg}$ of the W-OOA
mass in winter), while the W-OOA $_{\text{COARSE}}$ contribution was never statistically significant.
NH_4NO_3 behaved similarly with $31\%_{avg}$ of the mass in winter comprised in $PM_{2.5}$ - PM_1 . During
summer instead S-OOA showed a different behavior in the three size fractions: its contribution
was significant for PM_1 , but not in the size range 1-2.5 μm . The overall S-OOA _{2.5} fraction
accounted for $82\pm2\%_{avg}$ of the mass, while the remaining $18\pm2\%_{avg}$ was included in OM_{COARSE} .
Considering the sum of both OOA factors, the OOA/ $NH_4^+_{med}$ ratio for PM_1 was 2.1, consistent
with values reported by Crippa ⁵⁰ for 25 different European rural stations, suggesting that Payerne
can be representative of typical European rural environments.
The last PMF factor showed an unusual size fractionation with 96% _{avg} of its mass comprised in
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The last PMF factor showed an unusual size fractionation with 96% avg of its mass comprised in the PM _{COARSE} during summer (0.54±0.02 μg m ⁻³), corresponding to 49% of the WSOM _{COARSE} (or 19% avg of the OM _{COARSE}). This factor was ascribed to water soluble primary biological organic aerosol, given its striking mass spectral resemblance to biological carbohydrates and plant debris extracts with high contribution from C ₂ H ₄ O ₂ ⁺ , C ₂ H ₅ O ₂ ⁺ and C ₃ H ₅ O ₂ ⁺ (Figure 1d, S3, S10), its enhancement in OM _{COARSE} especially during summer, and its correlations with
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Also during winter WSPBOA showed a smaller but still significant contribution to the OM_{COARSE} (30% of WSOM_{COARSE} or 8% of OM_{COARSE}) with $68\%_{avg}$ of the mass comprised in the coarse fraction. This result was corroborated by a minor but statistically significant enhancement in the coarse fraction (in comparison with $PM_{2.5}$) of biological carbohydrates (monosaccharides_{BIO}: Σ (glucose, mannose, arabitol and mannitol)), cellulose, and fungal spores. The chemical characteristics and origin of this fraction will be thoroughly discussed in the following sections.

Composition of OM_{COARSE}.

This section presents a detailed characterization of OM_{COARSE} , of which 91% avg of the mass was ascribed to PBOA.

*Water soluble and insoluble OM*_{COARSE}. Figure 1c displays the relative chemical composition of OM_{COARSE} during summer. The major part of OM_{COARSE} could be ascribed to cellulose $(50\pm20\%_{avg})$ and WSOM_{COARSE} $(38\%_{avg})$. Given the low cellulose water solubility, and consequently its negligible contribution to WSOM, the two fractions together accounted for $88\%_{avg}$ of the OM_{COARSE}. Regarding the origin of the WSOM_{COARSE} fraction, 3D-PMF results revealed that only WSPBOA and WSS-OOA contributed significantly to WSOM_{COARSE} during summer, explaining respectively $51\%_{avg}$ and $49\%_{avg}$ of the WSOM_{COARSE} mass. Assuming the water insoluble OM_{COARSE} fraction not ascribed to S-OOA to be entirely related to PBOA, we calculated a R_{PBOA} lowest estimate of 0.18_{med} (1st quartile 0.15, 3rd quartile 0.25) according to eq 82, 83 and 84. This assumption was corroborated by the high cellulose contributions to the water insoluble OM_{COARSE} fraction ($82\%_{avg}$) and by the good correlation of WSPBOA with OM_{COARSE} S-OOA_{COARSE} (R=0.54), especially considering that the water insoluble OM_{COARSE} fraction represented $82\%_{avg}$ of the total OM_{COARSE}.

Contribution of carbohydrates to PBOA and OM_{COARSE} . Measured carbohydrates
$(carbohydrates_{meas}: \; \; \; \; \; \; \; \; (monosaccharides_{BIO}, \; \; mannosan, \; \; levoglucosan, \; \; and \; \; galactosan))$
represented 3% of OM_{COARSE} (8% of $WSOM_{COARSE}$), of which $93\%_{avg}$ was related to
monosaccharides $_{\rm BIO}$. This fraction, albeit minor, was highly correlated with PBOA (R =0.73) and
cellulose (R =0.85), showing a size fractionation similar to WSPBOA especially during summer
with $96\%_{avg}$ of the mass included in the OM_{COARSE} . A similar behavior was noted in winter, with
29% avg of the carbohydrates _{meas,COARSE} consisting of monosaccharides _{BIO} , suggesting a minor, but
statistically significant contribution of primary biological emissions, consistent with WSPBOA
from 3D-PMF results (figure 2). Also other biological components, such as cellulose and fungal
spores showed a small but significant contribution in winter (respectively 0.06 $\mu g\ m^{3}$ and $2{\cdot}10^1$
spores·m $^{-3}$ detected on the 31^{st} of January 2013 PM_{10} filter sample). However, the overall
correlation of single monosaccharides $_{\mbox{\footnotesize BIO}}$ with each other and with other PBOA components was
relatively poor, indicating a high variability in the molecular composition of the carbohydrates.
Such variability highlighted the diversity of biological processes producing these sugars, clearly
hindering their use as single tracers for reliably estimating PBOA concentrations in our
conditions.
By ascribing all the monosaccharides _{BIO,COARSE} to WSPBOA we estimated a contribution of
monosaccharides $_{\text{BIO}}$ to WSPBOA of 15% $_{\text{avg}}$. Consistently, the WSPBOA average mass spectrum
(Figure 1d), similarly to BBOA, showed a typical fingerprint deriving from carbohydrate
$fragmentation^{15} \ as \ evidenced \ by \ strong \ contributions \ from \ C_2H_4O_2^{\ +}, \ C_2H_5O_2^{\ +} \ and \ C_3H_5O_2^{\ +}$
fragments (Figure 1b, S3, S4, S10). We estimated that >89% of the remaining WSPBOA fraction
could be related to water soluble polysaccharides (after the subtraction of the
monosaccharides _{BIO} mass spectrum using D-mannitol and D-glucose as surrogates). This

337	estimate was based on the non-monosaccharides _{BIO} -WSPBOA mass spectrum, assuming
338	$C_2H_4O_2^+$, $C_2H_5O_2^+$ and $C_3H_5O_2^+$ as specific carbohydrates fragmentation tracers ¹⁵ (Figure S4),
339	and using amylopectin and starch (Figure S10) as surrogates for polysaccharides. This result,
340	together with the high cellulose contribution to OM_{COARSE} , indicated that the majority of PBOA
341	consisted of carbohydrates.
342	Part of the remaining WSPBOA fraction instead was attributed to N_{org} . 3D-PMF results
343	showed that WSPBOA explained great part of the variability of minor N-containing fragments
344	$(C_3H_9N^+,\ C_3H_8N^+,\ C_5H_{12}N^+)$, consistent with XPS observations of an increased N_{org} signal in
345	PM_{COARSE} . The WSPBOA spectrum as expected showed a higher N/C ratio (0.061) than other
346	factors. Overall both the carbohydrate signature and the increased N/C content were consistent
347	with the interpretation of our factor as WSPBOA.
348	Quantification of OM related to particulate abrasion products from leaf surfaces (OM_{PAPLS})
349	using n-alkanes. n-alkanes (C18-C39) measured via gas chromatography mass spectrometry
350	(IDTD-GC-MS) showed distinct signatures during the different seasons and particle sizes. While
351	during winter most of the alkane mass was contained within PM ₁ (90% for alkanes with an odd
352	number of C; 97% for alkanes with an even number of C), during summer only 50% avg and
353	$70\%_{avg}$ of the odd and even alkanes were contained within PM ₁ . The summer-time signatures
354	were consistent with Rogge's ⁵⁴ observations of alkane emissions from OM _{PAPLS} dominated by
355	odd alkanes with the highest contributions from hentriacontane (C31) followed by nonacosane
356	(C29) and tritriacontane (C33) (Figure S9). By contrast, in winter we observed a higher
357	contribution of smaller alkanes (C19-C24), without a clear odd/even predominance pattern,
358	which was consistent with winter urban observations ⁵⁵ possibly related to temperature-driven
359	partitioning of combustion emissions, and consistent with vehicular fuel combustion profiles. ^{47,56}

360	This was corroborated by a slight increase in the average HOA concentration during winter
361	compared to summer (Figure 2). We estimated the contribution of OM_{PAPLS} by applying a
362	chemical mass balance approach (SI) using the <i>n</i> -alkanes/OM _{PAPLS} ratios reported by Rogge. ^{56,57}
363	Assuming either green or dead leaves, and a possible (OM/OC) _{green,dead leaves} range between 1.2
364	and 2.2, the total estimated range for $OM_{PAPLS,COARSE}$ spanned from 0.5 to 1 μg m ⁻³ avg,
365	corresponding to $16\text{-}32\%_{avg}$ of the OM_{COARSE} . This result, together with high cellulose
366	contributions, indicated that plant debris was the dominating source of OM_{COARSE} .
367	Fungal spores. Fungal spores measured by qPCR represented a minor component of OM. During
368	summer, their contribution was above the detection limit only in the coarse fraction, representing
369	just $0.01\%_{avg}$ of the OM_{COARSE} mass (corresponding to 0.4 ng m ⁻³ , or $2\cdot10^2$ spores·m ⁻³).
370	Nevertheless, the measured fungal spore/m³ concentration during summer was consistent with
371	ranges reported in other studies. ⁵⁸ During winter, only one PM ₁₀ sample showed concentrations
372	above the detection limits. The summer arabitol/fungal spore $(5\cdot10^2\ pg/spore_{avg})$ and
373	mannitol/fungal spore $(8\cdot10^2~pg/spore_{avg})$ ratios were noticeably variable and higher than those
374	reported by Bauer ¹⁹ (1.2 pg arabitol/fungal spore, 1.7 pg mannitol/fungal spore), suggesting that
375	these compounds are not unique fungal spore tracers, but given the high levels of cellulose and
376	OM _{PAPLS} could be related to plant debris, as already proposed by other studies. ²⁰
377	Bacteria. Likewise, total bacterial mass estimated by qPCR represented a minor contributor to
378	OM _{COARSE} . Assuming dry or wet <i>E. coli</i> cellular weights (SI), the total PM ₁₀ bacterial mass
379	during summer was estimated as 1.3±0.7 ng m ⁻³ _{avg} or 4±0.2 ng m ⁻³ _{avg} , corresponding to 2·10 ³
380	cells m ⁻³ _{avg} . This is consistent with the ranges reported in other studies, ⁵⁸⁻⁶⁰ especially
381	considering that low concentrations are commonly observed at remote and rural locations. ⁶¹ The
382	bacterial size fractionation seasonality was similar to the other biological components: while

383	$69\%_{avg}$ of the bacterial mass was comprised between the PM_{10} and PM_1 fraction during summer,
384	all bacterial mass $(2\cdot10^3 \text{ cells m}^{-3}_{avg})$ was detected in the submicron fraction during winter.
385	Surface chemical composition from XPS analysis. Another approach to look at the entire
386	aerosol is to study the chemical composition of its surface. This was performed by XPS
387	measurements, which enabled monitoring the evolution of the C_{1s} and N_{1s} BE throughout the
388	different size fractions and thus providing chemical information also about the water insoluble
389	fraction. Although XPS sensitivity was limited to the particle surface (7 nm thickness) and low
390	volatility compounds (XPS technique operates under high vacuum at 10 ⁻¹⁰ torr), results showed a
391	significant increase of N_{org} in the PM_{COARSE} . We resolved both an inorganic and organic N_{1s}
392	peak, with $N_{1s,org}$ occurring at a lower BE (397.7±0.3 eV, Figure 3a) than that of $N_{1s(NH_4)_2SO_4}$
393	and NaNO $_3$ (400.0±0.8 eV and 407.7±0.4 eV respectively). Likewise, tested N $_{org}$ surrogates
394	(horseradish peroxidase and chloroperoxidase from $\it caldariomyces\ fumago$) showed the N_{1s} peak
395	occurring at similar BE (398.7 \pm 0.3 eV) corroborating our interpretation of the N_{org} peak position.
396	Overall we observed a substantial increase of the N_{org} signal in PM_{10} in comparison to PM_{1}
397	(Figure 3a) reflected by an N_{org}/C_{1s} ratio increase from 0.022 ± 0.001 in PM_1 to 0.027 ± 0.005 in
398	PM_{10} . From the N_{org}/C_{1s} ratio and from the bulk total C measurements (TC=EC+OC) _{Sunset} , we
399	estimated the $N_{org,1}$ and $N_{org,10}$ concentrations to be 0.05±0.03 μg m $^{-3}{}_{avg}$ and 0.13±0.01 μg m $^{-3}{}_{avg}$
400	respectively. This estimate assumed N_{org} to follow the TC intra-particle concentration gradient.
401	While a crude assumption, this is the best and only methodology providing an estimate of the
402	N _{org} total mass.
403	Figure 3b displays the C_{1s} peak fitting for a PM_1 and a PM_{10} filter sample. We report an
404	increase of the less oxidized C_{1s} fraction (C_{1s} peak at lower BE) in PM_{10} , which was qualitatively
405	consistent with the odd-alkanes size fractionation. Overall, in all size fractions, the dominant C_{1s}

contribution did not derive from the most oxidized C_{1s} peak (Figure 3b), but from the intermediate oxidized C peak, which could be related to alcohols, ketones, and aldehydes. This result, although relative only to the surface and to the less volatile fractions, seemed in agreement with other studies.⁶²

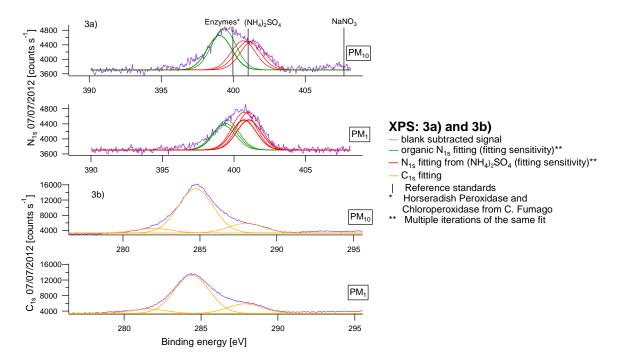


Figure 3. 3a) XPS measurements: N_{1s} peak fitting (PM₁ and PM₁₀ sample from 04/07/2012). 3b) XPS measurements: C_{1s} peak fitting (PM₁ and PM₁₀ sample from 04/07/2012).

Yearly estimate of PBOA relative contribution to OM₁₀

From 3D-PMF analysis we identified a set of AMS fragments as potential PBOA tracers (figure S4). Among these fragments we selected $C_2H_4O_2^+$ and $C_2H_5O_2^+$ to estimate the PBOA contribution for the entire year 2013 (batch B) given their relatively high signal to noise, and because they are commonly fitted in HR analysis. Both fragments showed a contribution statistically higher than 0 within 1σ only to the BBOA, PBOA, and HOA factors. However, given the low HOA concentration at the rural site (Figure 2a), and given the low contribution of

- the two fragments to the HOA profile (0.02% and 0.03% respectively) we neglected the HOA contribution to $C_2H_4O_2^+$ and $C_2H_5O_2^+$. Therefore the water soluble $C_2H_5O_2^+$ and $C_2H_4O_2^+$ fractional contribution to WSOM ($WSfC_2H_5O_2^+_i$ and $WSfC_2H_4O_2^+_i$) could be expressed as:
- 424 $WSfC_{2}H_{5}O_{2}^{+}{}_{i} = fC_{2}H_{5}O_{2}^{+}{}_{WSPBOA} \cdot \frac{WSPBOA}{WSOM}{}_{i} + fC_{2}H_{5}O_{2}^{+}{}_{WSBBOA} \cdot \frac{WSBBOA}{WSOM}{}_{i}$ (6)
- 425 $WSfC_{2}H_{4}O_{2}^{+}{}_{i} = fC_{2}H_{4}O_{2}^{+}{}_{WSPBOA} \cdot \frac{WSPBOA}{WSOM}{}_{i} + fC_{2}H_{4}O_{2}^{+}{}_{WSBBOA} \cdot \frac{WSBBOA}{WSOM}{}_{i}$ (7)
- Where $fC_2H_5O_2^+_{PBOA}$, $fC_2H_4O_2^+_{PBOA}$, $fC_2H_5O_2^+_{BBOA}$, $fC_2H_4O_2^+_{BBOA}$ denote the $C_2H_5O_2^+$, and
- 427 C₂H₄O₂⁺ fractional contributions to the WSPBOA and WSBBOA mass spectra.
- 428 (WSPBOA/WSOM), values could be derived by solving the two linear equation system. This
- approach will be referred to as "60/61 methodology" in the following. We assessed the accuracy
- of the 60/61 methodology by comparing the (WSPBOA/WSOM)_i values obtained from 3D-PMF
- with the values predicted from the 60/61 methodology for the Batch A PM₁₀ filter samples.
- During summer the (WSPBOA/WSOM)_{med,3D-PMF}/(WSPBOA/WSOM)_{med,60/61 methodology} ratio was
- 433 0.98, while during winter 0.85. The winter discrepancy was likely due to non-negligible
- contributions of W-OOA or other sources to fC₂H₄O₂⁺ and fC₂H₅O₂⁺. However the two
- methodologies yielded highly correlated time series (R^2 =0.81) and agreed within 15%, with
- 436 much better agreement during summer.
- From the 60/61 methodology we estimated a WSPBOA/WSOM of $20\%_{avg}$ in summer, and $6\%_{avg}$
- 438 in winter. Assuming a R_{PBOA} of 0.18_{med} (SI), the average PBOA contribution to OM₁₀ was
- estimated as 37%_{avg}, with higher values during summer (60%_{avg} vs. 19%_{avg} in winter).
- Overall, these results revealed that the contribution of PBOA to OM₁₀, mainly from plant debris,
- 441 may be as high as SOA contribution during summer in Payerne. While Payerne can be
- considered as representative of typical European rural environments⁵⁰ and therefore results here
- may be extended to other sites, other field observations are indeed required. This work represents

a benchmark for future field studies providing a methodology for the thorough determination of PBOA mass and origin, and one of the first size-segregated datasets necessary to constrain PBOA in global models.

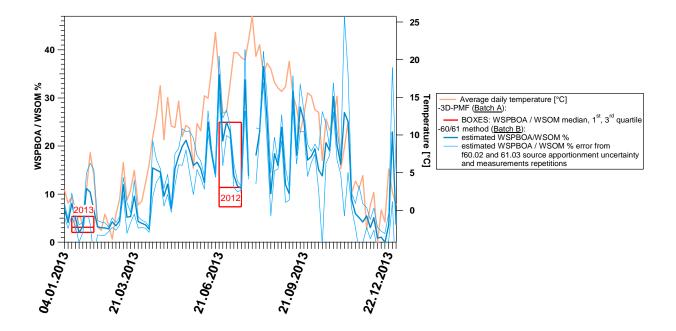


Figure 4. 2013 yearly WSPBOA₁₀ relative contribution to WSOM₁₀ estimated from the 60/61 methodology (Batch B). Red boxes denote WSPBOA relative contribution (median, 1^{st} and 3^{rd} quartiles) to WSOM₁₀ during June-July 2012 and January-February 2013 determined by 3D-PMF analysis (Batch A). The uncertainty relative to measurements repetitions and to the apportionment of $fC_2H_4O_2^+$ and $fC_2H_5O_2^+$ can be interpreted as a precision estimate, while the sensitivity analysis comparing 3D-PMF and 60/61 methodology results, shows an underestimate of the WSPBOA/WSOM ratio calculated with the 60/61 methodology of 2% during summer and 15% during winter.

ASSOCIATED CONTENT

- Supporting Information. Detailed methodology descriptions of WSOC, qPCR, XPS, and
- 460 IDTD-GC-ToF-MS measurements; OM_{PAPLS} determination; source apportionment optimization.
- This material is available free of charge via the Internet at http://pubs.acs.org.

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468 **Author Contributions**

- 469 [†]C.B. wrote the manuscript. [†]C.B. and ^{†*}I.E.H performed the data analysis and source
- apportionment. †*A.S.H.P., †*I.E.H., †C.B. and †J.G.S. designed the experiment. †C.B. and †A.K.
- performed the offline-AMS analysis. *P.F. and *R.G. performed WSOC measurements. *J.S.
- measured carbohydrates_{meas} and EC/OC. [‡]C.H. collected the samples, and measured ions and
- 473 EC/OC. G.A., G.A., and J.S.-K. performed IDTD-GC-ToF-MS measurements. Y.R., T.S.M.
- and Y.M. performed qPCR measurements. M.E.K., C.B. and I.E.H. performed XPS
- 475 measurements. ⁸A.K.-G. and ⁸M.F. performed cellulose measurements. All authors gave
- approval to the final version of the manuscript.

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- 486 REFERENCES
- 487 (1) Pasteur, L. Mémoire sur les corpuscles organises qui existent dans l'atmosphere. Examen
- de la doctrine des générations spontanées. Ann. Sci. Nat. Zool. **1861**, 16, 5–98.
- 489 (2) Carnelly, T.; Haldane, J. S.; Anderson, A. M. The carbon acid, organic matter, and micro-
- organisms in air, more especially of dwellings and schools. Philos. Transact. R. Soc. Lond.
- 491 B. **1887**, 178, 61–111.
- 492 (3) Fuzzi, S.; Baltensperger, U.; Carslaw, K.; Decesari, S.; Denier van der Gon, H.; Facchini,
- 493 M. C.; Fowler, D.; Koren, I.; Langford, B.; Lohmann, U.; Nemitz, E.; Pandis, S.; Riipinen,
- 494 I.; Rudich, Y.; Schaap, M.; Slowik, J. G.; Spracklen, D. V.; Vignati, E.; Wild,
- 495 M.; Williams, M.; Gilardoni. S. Particulate matter, air quality and climate: lessons learned
- and future needs. Atmos. Chem. Phys. **2015**, 15, 8217-8299.
- 497 (4) Després, V. R.; Huffman, J. A.; Burrows, S. M; Hoose, C.; Safatov, A. S.; Buryak, G.;
- 498 Fröhlich-Nowoisky, J.; Elbert, W.; Andreae, M. O.; Pöschl, U.; Jaenicke, R. Primary
- biological aerosol particles in the atmosphere: a review. Tellus B. **2012**, 64, 15598.
- 500 (5) Douwes, J.; Thorne, P.; Pearce, N.; Heederik, D. Bioaerosol health effects and exposure
- assessment: progress and prospects. Ann. Occup. Hyg. **2003**, 47, 187–200.

- 502 (6) Hiranuma, N.; Möhler, O.; Yamashita, K.; Tajiri, T.; Saito, A.; Kiselev, A.; Hoffmann, N.;
- Hoose, C.; Jantsch, E.; Koop, T.; Murakami M. Ice nucleation by cellulose and its
- potential contribution to ice formation in clouds. Nature Geosci. **2015**, 8, 273-277.
- 505 (7) Hader, J. D.; Wright, T. P.; Petters. M. D. Contribution of pollen to atmospheric ice nuclei
- 506 concentrations Atmos. Chem. Phys. **2014**, 14, 5433-5449.
- 507 (8) Gurian-Sherman, D.; Lindow., S. E.; Bacterial ice nucleation: significance and molecular
- 508 basis. FASEB J. **1993**,14, 1338-1343.
- 509 (9) Andreae, M. O.; Rosenfeld, D. Aerosol-cloud-precipitation interactions. Part 1. The nature
- and sources of cloud-active aerosols. Earth Sci. Rev. **2008**, 89, 13–41.
- 511 (10) Ariya, P. A.; Sun, J., Eltouny, N. A.; Hudson, E. D.; Hayes, C. T; Kos, G. Physical and
- 512 chemical characterization of bioaerosols–implications for nucleation processes. Int. Rev.
- 513 Phys. Chem. **2009**, 28, 1–32.
- 514 (11) Andreae, M. O. Aerosols before pollution. Science. **2007**, 315, 50-51.
- 515 (12) Fu, P.; Kawamura, K.; Chen, J.; Qin, M.; Ren., L.; Sun, Y.; Wang, Z.; Barrie, L. A.;
- Tachibana, E.; Ding, A.; Yamashita, Y. Fluorescent water-soluble organic aerosol in the
- 517 High Arctic atmosphere. Sci. Rep. **2015**, 5, 9845.
- 518 (13) Pöhlker, C.; Huffman, J. A.; Pöschl U. Autofluorescence of atmospheric bioaerosols -
- fluorescent biomolecules and potential interferences. Atmos. Meas. Tech., **2012**, 5, 37–71.
- 520 (14) Chen, Q.; Farmer, D. K.; Schneider, J.; Zorn, S. R.; Heald, C. L; Karl, T. G.; Guenther,
- A.; Allan, J. D.; Robinson, N.; Coe, H.; Kimmel, J. R.; Pauliquevis, T.; Borrmann, S.;
- Pöschl, U.; Andreae, M. O.; Artaxo, P.; Jimenez, J. L.; Martin, S. T. Mass spectral

- 523 characterization of submicron biogenic organic particles in the Amazon Basin. Geophys.
- 524 Res. Lett. **2009**, 36, L20806.
- 525 (15) Schneider, J.; Freutel, F.; Zorn, S. R.; Chen, Q.; Farmer, D. K.; J. L. Jimenez, Martin, S.
- T. Artaxo, P.; Wiedensohler, A.; Borrmann, S. Mass-spectrometric identification of
- 527 primary biological particle markers: indication for low abundance of primary biological
- material in the pristine submicron aerosol of Amazonia. Atmos. Chem. Phys. Discuss.
- **2011**, 11, 19143–19178.
- 530 (16) Pöschl, U., Martin, S. T., Sinha, B., Chen, Q., Gunthe, S. S.; Huffman, J. A.; Borrmann,
- S.; Farmer, D.K.; Garland, R. M.; Helas, G.; Jimenez, J. L.; King, S. M.; Manzi, A.;
- Mikhailov, E.; Pauliquevis, T.; Petters, M. D.; Prenni, A. J.; Roldin, P.; Rose, D.;
- Schneider, J.; Su, H.; Zorn, S. R.; Artaxo, P.; Andreae, M. O. Rainforest aerosols as
- biogenic nuclei of clouds and precipitation in the Amazon. Science. **2010**, 329, 1513–1516.
- 535 (17) IPCC, 2013: Climate Change 2013: The Physical Science Basis. Contribution of Working
- Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change
- [Stocker, T.F.; Qin, D.; Plattner, G.-K.; Tignor, M.; Allen, S. K.; Boschung, J.; Nauels, A.;
- Xia, Y.; Bex, V.; Midgley, P. M. (eds.)]. Cambridge University Press, Cambridge, United
- Kingdom and New York, NY, USA, 1535 pp.
- 540 (18) Heald, C. L.; Spracklen D. V. Atmospheric budget of primary biological aerosol particles
- from fungal spores. Geophys. Res. Lett. **2009**, 36, L09806-L09806.
- 542 (19) Bauer, H.; Claeys, M.; Vermeylen, R.; Schueller, E.; Weinke, G.; Berger, A.; Puxbaum,
- H. Arabitol and mannitol as tracers for the quantification of airborne fungal spores, Atmos.
- 544 Environ. **2008**, 42, 588–593.

- 545 (20) Burshtein, N.; Lang-Yona, N.; Rudich, Y. Ergosterol, arabitol and mannitol as tracers for 546 biogenic aerosols in the eastern Mediterranean. Atmos. Chem. Phys. **2011**, 11, 829–839.
- 547 (21) Daellenbach, K. R.; Bozzetti, C.; Krepelova, A.; Canonaco, F.; Huang, R.-J.; Wolf, R.;
- Zotter, P.; Crippa, M.; Slowik, J.; Zhang, Y.; Szidat, S.; Baltensperger, U.; Prévôt, A. S.
- H.; El Haddad, I. Characterization and source apportionment of organic aerosol using
- offline aerosol mass spectrometry. Atmos. Meas. Tech. **2016**, 9, 23-29.
- 551 (22) Birch, M. E.; Cary, R. A. Elemental carbon-based method for monitoring occupational
- exposures to particulate diesel exhaust. Aerosol Sci. and Tech. **1996**, 25, 221–241.
- 553 (23) Cavalli, F.; Viana, M.; Yttri, K. E.; Genberg, J.; Putaud, J. P. Toward a standardised
- thermal-optical protocol for measuring atmospheric organic and elemental carbon: the
- 555 EUSAAR protocol. Atmos. Meas. Tech. **2010**, 3, 79-89.
- 556 (24) Piazzalunga, A.; Bernardoni, V.; Fermo, P.; Vecchi, R.Optimisation of analytical
- procedures for the quantification of ionic and carbonaceous fractions in the atmospheric
- aerosol and applications to ambient samples. Anal Bioanal Chem. **2013**, 405, 1123-32.
- 559 (25) Kunit, M.; Puxbaum, H. Enzymatic determination of the cellulose content of atmospheric
- 560 aerosols. Atmos. Environ. **1996**, 30, 1233-1236.
- 561 (26) Orasche, J.; Schnelle-Kreis, J.; Abbaszade, G.; Zimmermann, R. Technical Note: In-situ
- derivatization thermal desorption GC-TOFMS for direct analysis of particle-bound non-
- polar and polar organic species. Atmos. Chem. Phys. **2011**, 11, 8977-8993.
- 564 (27) Lang-Yona, N.; Dannemiller, K.; Yamamoto, N.; Burshtein, N.; Peccia, J.; Yarden, O.;
- Rudich, Y. Annual distribution of allergenic fungal spores in atmospheric particulate

- matter in the Eastern Mediterranean; a comparative study between ergosterol and quantitative PCR analysis. Atmos. Chem. Phys. **2012**, 12, 2681–2690.
- 568 (28) Lang-Yona, N.; Lehahn, Y.; Herut, B.; Burshtein, N.; Rudich, Y. Marine aerosol as a possible source for endotoxins in coastal areas. Sci. Total Environ. **2014**, 499, 311–318.
- 570 (29) Yttri, K. E.; Schnelle-Kreis, J.; Maenhaut, W.; Abbaszade, G.; Alves, C.; Bjerke, A.;
- Bonnier, N.; Bossi, R.; Claeys, M.; Dye, C.; Evtyugina, M.; García-Gacio, D.; Hillamo, R.;
- Hoffer, A.; Hyder, M.; Iinuma, Y.; Jaffrezo, J.-L.; Kasper-Giebl, A.; Kiss, G.; López-
- Mahia, P. L.; Pio, C.; Piot, C.; Ramirez-Santa-Cruz, C.; Sciare, J.; Teinilä, K.;
- Vermeylen, R.; Vicente, A.; Zimmermann, R. An intercomparison study of analytical
- 575 methods used for quantification of levoglucosan in ambient aerosol filter samples, Atmos.
- 576 Meas. Tech., **2015**, 8, 125-147.
- 577 (30) DeCarlo, P. F.; Kimmel, J. R.; Trimborn, A.; Northway, M. J.; Jayne, J. T.; Aiken, A. C.;
- Gonin, M.; Fuhrer, K.; Horvath, T.; Docherty, K. S.; Worsnop, D. R.; Jimenez, J. L. Field-
- deployable, high-resolution, time-of-flight aerosol mass spectrometer. Anal. Chem. **2006**,
- 580 78, 8281–8289.
- 581 (31) Hospodsky, D.; Yamamoto, N.; Peccia, J.; Accuracy, precision, and method detection
- limits of quantitative PCR for airborne bacteria and fungi. Appl. Environ. Microb. **2010**,
- 583 76, 7004-7012.
- 584 (32) Sundararaj, S.; Guo, A.; Habibi-Nazhad, B.; Rouani, M.; Stothard, P.; Ellison, M.;
- Wishart, D. S. The CyberCell Database (CCDB): a comprehensive, self-updating,
- 586 relational database to coordinate and facilitate in silico modeling of Escherichia coli.
- 587 Nuclei Acid Res. **2004**, 32, D263-D265.

- 588 (33) Crilley, L. R.; Ayoko, G. A.; Morawska, L. Analysis of organic aerosols collected on 589 filters by Aerosol Mass Spectrometry for source identification. Anal. Chim. Acta. **2013**, 590 803, 91–96.
- 591 (34) Paatero, P.; Tapper, U. Positive matrix factorization a nonnegative factor model with optimal utilization of error-estimates of data values. Environmetrics **1994**, 5, 111-126.
- (35) Ulbrich, I. M.; Canagaratna, M. R.; Cubison, M. J.; Zhang, Q.; Ng, N. L.; Aiken, A. C.;
 Jimenez, J. L. Three-dimensional factorization of size-resolved organic aerosol mass
 spectra from Mexico City. Atmos. Meas. Tech. 2012, 5, 195–224.
- (36) Tucker, L. R. Some mathematical notes on 3-mode factor analysis. Psychometrika 1966,
 31, 279–311.
- 598 (37) Paatero, P.; Hopke, K. Rotational tools for factor analytic models. J. Chemometr. **2009**, 599 23, 91–100.
- 600 (38) Canonaco, F.; Crippa, M.; Slowik, J. G.; Baltensperger, U.; Prévôt, A. S. H. SoFi, an
 601 IGOR-based interface for the efficient use of the generalized multilinear engine (ME-2) for
 602 the source apportionment: ME-2 application to aerosol mass spectrometer data. Atmos.
 603 Meas. Tech. **2013**, 6, 3649-3661.
- (39) Mohr, C.; DeCarlo, P. F.; Heringa, M. F.; Chirico, R.; Slowik, J. G.; Richter, R.; Reche,
 C.; Alastuey, A.; Querol, X.; Seco, R.; Penuelas, J.; Jimenez, J. L.; Crippa, M.;
 Zimmermann, R.; Baltensperger, U.; Prevot, A. S. H. Identification and quantification of
 organic aerosol from cooking and other sources in Barcelona using aerosol mass
 spectrometer data. Atmos. Chem. Phys. 2012, 12, 1649-1665.

- 609 (40) Aiken, A. C.; DeCarlo, P. F.; Kroll, J. H.; Worsnop, D. R.; Huffmann, J. A.; Docherty, K.
- S.; Ulbrich, I. M.; Mohr, C.; Kimmel, J. R.; Sueper, D.; Sun, Y.; Zhang, Q.; Trimborn, A.;
- Northway, M.; Ziemann, P. J.; Canagaratna, M. R.; Onasch, T. B.; Alfarra, M. R.; Prevot,
- A. S. H.; Dommen, J.; Duplissy, J.; Metzger, A.; Baltensperger, U.; Jimenez J. L. O/C and
- OM/OC ratios of primary, secondary, and ambient organic aerosols with high-resolution
- time-of-flight aerosol mass spectrometry. Environ. Sci. Technol., **2008**, 42, 4478-4485.
- 615 (41) Allan, J. D.; Jimenez, J. L.; Williams, P. I.; Alfarra, M. R.; Bower, K. N.; Jayne, J. T.;
- Coe, H.; Worsnop, D. R. Quantitative sampling using an Aerodyne aerosol mass
- spectrometer 1. Techniques of data interpretation and error analysis. J. Geophys. Res.,
- 618 **2003**, 108 (D3), 4090.
- 619 (42) Ulbrich, I. M.; Canagaratna, M. R.; Zhang, Q.; Worsnop, D. R.; Jimenez, J. L.
- Interpretation of organic components from positive matrix factorization of aerosol mass
- 621 spectrometric data. Atmos. Chem. Phys. **2009**, 9, 2891-2918.
- 622 (43) Crippa, M.; Canonaco, F.; Slowik, J. G.; El Haddad, I.; DeCarlo, P. F.; Mohr, C.;
- Heringa, M. F.; Chirico, R.; Marchand, N.; Temime-Roussel, B.; Abidi, E.; Poulain,
- 624 L.; Wiedensohler, A.; Baltensperger, U.; Prévôt; A. S. H. Primary and secondary organic
- aerosol origin by combined gas-particle phase source apportionment. Atmos. Chem. Phys.
- **2013**, 13, 8411-8426.
- 627 (44) Barmpadimos, I.; Nufer, M.; Oderbolz, D. C.; Keller, J.; Aksoyoglu, S.; Hueglin, C.;
- Baltensperger, U.; Prevot A. S. H. The weekly cycle of ambient concentrations and traffic
- emissions of coarse (PM(10)-PM(2.5)) atmospheric particles, Atmos. Environ. **2011**, 45,
- 630 4580-4590.

- 631 (45) Chow, J.; Watson, J.; Ashbaugh, L. L.; Magliano, K. L. Similarities and differences in
- PM10 chemical source profiles for geological dust from the San Joaquin Valley, California.
- 633 Atmos. Environ. **2003**, 37, 1317-1340.
- 634 (46) Amato, F.; Pandolfi, M; Moreno, T.; Furger, M.; Pey, J.; Alastuey, A.; Bukowiecki, N.;
- Prevot, A.S.H.; Baltensperger, U.; Querol X. Sources and variability of inhalable road dust
- particles in three European cities. Atmos. Environ. **2011**, 45, 6777-6787.
- 637 (47) El Haddad, I.; Marchand, N.; Drona, J.; Temime-Roussel, B.; Quivet, E.; Wortham, H.;
- Jaffrezo, J.-L.; Baduel, C.; Voisin, D.; Besombes, J. L.; Gille, G. Comprehensive primary
- particulate organic characterization of vehicular exhaust emissions in France. Atmos.
- Environ. **2009**, 43, 6190-6198.
- 641 (48) Huang, R.-J.; Zhang, Y.; Bozzetti, C.; Ho, K.-F.; Cao, J.; Han, Y.; Dällenbach, K. R.;
- Slowik, J. G.; Platt, S. M.; Canonaco, F.; Zotter, P.; Wolf, R.; Pieber, S. M.; Bruns, E. A.;
- 643 Crippa, M.; Ciarelli, G.; Piazzalunga, A.; Schwikowski, M.; Abbaszade, G.; Schnelle-
- Kreis, J.; Zimmermann, R.; An, Z.; Szidat, S.; Baltensperger, U.; El Haddad, I.; Prévôt, A.
- S. H. High secondary aerosol contribution to particulate pollution during haze events in
- 646 China, Nature. **2014**, 514, 218-212.
- 647 (49) Canonaco, F.; Slowik, J. G.; Baltensperger, U.; Prévôt, A. S. H. Seasonal differences in
- oxygenated organic aerosol composition: implications for emissions sources and factor
- analysis. Atmos. Chem. Phys. **2015**, 15, 6993-7002.
- 650 (50) Crippa, M.; Canonaco, F.; Lanz, V. A.; Äijälä, M.; Allan, J. D.; Carbone, S.; Capes, G.;
- 651 Ceburnis, D.; Dall'Osto, M.; Day, D. A.; DeCarlo, P. F.; Ehn, M.; Eriksson, A.; Freney, E.;
- 652 Hildebrandt Ruiz, L.; Hillamo, R.; Jimenez, J. L.; Junninen, H.; Kiendler-Scharr, A.;

- Kortelainen, A.- M.; Kulmala, M.; Laaksonen, A.; Mensah, A. A.; Mohr, C.; Nemitz, E.;
- O'Dowd, C.; Ovadnevaite, J.; Pandis, S. N.; Petäjä, T.; Poulain, L.; Saarikoski, S.; Sellegri,
- K.; Swietlicki, E.; Tiitta, P.; Worsnop, D. R.; Baltensperger, U.; Prévôt, A. S. H. Organic
- aerosol components derived from 25 AMS data sets across Europe using a consistent ME-2
- based source apportionment approach. Atmos. Chem. Phys. **2014**, 14, 6159–6176.
- 658 (51) Medeiros, P. M.; Conte, M. H.; Weber, J. C.; Simoneit, B. R. T. Sugars as source
- indicators of biogenic organic carbon in aerosols collected above the Howland
- Experimental Forest, Maine. Atmos. Environ. **2006**, 40, 1694-1705.
- 661 (52) Jia, Y.; Clements, A. L.; Fraser, M. P. Saccharide composition in atmospheric particulate
- matter in the southwest US and estimates of source contributions. J. Aerosol Sci. **2010**, 41,
- 663 62-73.
- 664 (53) Williams, L. R.; Gonzalez, L. A.; Peck, J.; Trimborn, D.; McInnis, J.; Farrar, M. R.;
- Moore, K. D.; Jayne, J. T.; Robinson, W. A.; Lewis, D. K.; Onasch, T. B.; Canagaratna, M.
- R.; Trimborn, A.; Timko, M. T.; Magoon, G.; Deng, R.; Tang, D.; de la Rosa Blanco, E.;
- Prévôt, A. S. H.; Smith, K. A.; Worsnop D. Characterization of an aerodynamic lens for
- transmitting particles greater than 1 micrometer in diameter into the Aerodyne aerosol mass
- spectrometer. Atmos. Meas. Tech. **2013**, 6, 3271–3280.
- 670 (54) Rogge, W. F.; Hildemann, L. M.; Mazurek, M. A.; Cass, G. R; Simoneit, B. R. T.
- Sources of fine organic aerosol. 4. Particulate abrasion products from leaf surfaces of urban
- 672 plants, Environ. Sci. Technol., **1993**, 27 (13), 2700–2711.

- 673 (55) Kotianová, P.; Puxbaum, H.; Bauer, H.; Caseiro, A.; Marrb, I. L.; Čìk, G.; Temporal
- patterns of n-alkanes at traffic exposed and suburban sites in Vienna. Atmos. Environ.
- **2008**, 42, 2993–3005.
- 676 (56) Rogge, W. F.; Hildemann, L. M.; Mazurek, M., A.; Caw, G. R. Sources of Fine Organic
- Aerosol. 2. Noncatalyst and Catalyst-Equipped Automobiles and Heavy-Duty Diesel
- 678 Trucks. Environ. Sci. Technol. **1993**, 27, 636-651.
- 679 (57) Hildemann, L. M.; Mazurek, M. A.; Cass, G. R.; Simoneit, B. R. Quantitative
- characterization of urban sources of organic aerosol by high-resolution gas-
- chromatography, Environ. Sci. Technol. **1991**, 25, 1311-1325.
- 682 (58) Borodulin, A.; Safatov, A.; Belan, B.; Panchenko, M. Measurement errors in determining
- tropospheric bioaerosol concentrations in the southern region of Western Siberia. Dokl.
- 684 Biol. Sci. **2005**, 403, 260–262.
- 685 (59) Vlodavets, V.; Mats, L. The influence of meteorological factors on the microflora of the
- atmospheric air in Moscow. J. Microbiol. **1958**, 59, 539–544.
- 687 (60) Pady, S.; Kelly, C. Aerobiological studies of fungi and bacteria over the Atlantic Ocean.
- 688 Can. J. Botany. **1954**, 32, 202–212.
- 689 (61) Burrows, S. M.; Elbert, W.; Lawrence, M. G.; Pöschl, U. Bacteria in the global
- atmosphere–Part 1: Review and synthesis of literature data for different ecosystems.
- 691 Atmos. Chem. Phys. **2009**, 9, 9281–9297.
- 692 (62) El Haddad, I.; D'Anna, B.; Temime-Roussel, B.; Nicolas, M.; Boreave, A.; Favez, O.;
- Voisin, D.; Sciare, J.; George, C.; Jaffrezo, J.-L.; Wortham, H.; Marchand, N.

Towards a better understanding of the origins, chemical composition and aging of oxygenated organic aerosols: case study of a Mediterranean industrialized environment, Marseille. Atmos. Chem. Phys. **2013**, 13, 7875-7894.

Graphical TOC Entry

