

Quantitative biokinetics of titanium dioxide nanoparticles after oral administration in rats (Part 2)

URL: http://mc.manuscriptcentral.com/tnan

³ Current address: Dept. Infrastructure, Safety, Occupational Protection, German Research Center for Environmental Health, D-85764 Neuherberg / Munich, Germany

 Current address: Bavarian Health and Food Safety Authority, D-85764 Oberschleissheim, Germany

Page 3 of 35

Nanotoxicology

24 Size-selected, radiolabeled titanium dioxide nanoparticles; gavage; gut-absorption; accumulation in

25 secondary organs and tissues; different biokinetics pattern after gavage versus intravenous injection

ABSTRACT

28 The biokinetics of a size-selected fraction (70nm median size) of commercially available and 48 V-29 radiolabeled $\int^{48}V\right]TiO_2$ nanoparticles has been investigated in female Wistar-Kyoto rats at retention 30 timepoints 1h, 4h, 24h and 7days after oral application of a single dose of an aqueous $\int^{48}V|TiO_{2}$ -31 nanoparticle suspension by intra-esophageal instillation. A completely balanced quantitative body 32 clearance and biokinetics in all organs and tissues was obtained by applying typical $\int^{48}V|TiO_{2}$ -33 nanoparticle doses in the range of 30–80 μ g•kg⁻¹ bodyweight, making use of the high sensitivity of 34 the radiotracer technique.

boss in the range of 30–80 μ g•kg⁻¹ bodyweight, making use of the high
technique.
 For Peer Review Only and Solution and for ⁴⁸V-ions not bound to TiO₂-nanopar

real extertion about 0.6% of the administered dose 35 The $\int^{48}V|TiO_2$ -nanoparticle content was corrected for nanoparticles in the residual blood retained in 36 organs and tissue after exsanguination and for ^{48}V -ions not bound to TiO₂-nanoparticles. Beyond 37 predominant fecal excretion about 0.6% of the administered dose passed the gastro-intestinal-barrier 38 after -h and about 0.05% were still distributed in the body at day-7, with quantifiable $\int^{48}V|TiO_{2}$ -39 nanoparticle organ concentrations present in liver $(0.09$ ng•g⁻¹), lungs $(0.10$ ng•g⁻¹), kidneys (0.29ng•g⁻¹), brain (0.36ng•g⁻¹), spleen (0.45ng•g⁻¹), uterus (0.55ng•g⁻¹) and skeleton (0.98ng•g⁻¹). 41 Since chronic, oral uptake of $TiO₂$ particles (including a nano-fraction) by consumers has 42 continuously increased in the past decades , the possibility of chronic accumulation of such 43 biopersistent nanoparticles in secondary organs and the skeleton raises questions about the 44 responsiveness of their defense capacities, and whether these could be leading to adverse health 45 effects in the population at large.

After normalizing the fractions of retained $\int^{48}V|TiO_2$ -nanoparticles to the fraction that passed the 47 gastro-intestinal-barrier and reached systemic circulation the biokinetics was compared to the 48 biokinetics determined after IV-injection (Part 1). Since the biokinetics patterns differ largely IV-49 injection is not an adequate surrogate for assessing the biokinetics after oral exposure to $TiO₂$ 50 nanoparticles.

Nanotoxicology

Introduction

54 On a daily basis, a typical individual in the Western world ingests an estimated average of 2.5 mg of 55 insoluble, submicrometer titanium dioxide $(TiO₂)$ particles, including a nanoparticulate fraction, 56 equivalent to an estimated 10^{12} -10¹⁴ particles, while the most elevated levels may be as high as 112 57 mg•d⁻¹ (Lomer, 2004), corresponding to an upper dose of 1.6 mg•kg⁻¹ body weight.

health care products such as swallowed toothpaste. Many of these ing
100 nm diameter. Hence by most current definitions they are not
However, all TiO₂ food additives are characterized by a wide size distri
36% of the par 58 Major sources of dietary $TiO₂$ are food additives (E171), confectionary products, pharmaceuticals, 59 cosmetics and health care products such as swallowed toothpaste. Many of these ingested particles 60 are larger than 100 nm diameter. Hence by most current definitions they are not considered as 61 nanoparticles. However, all $TiO₂$ food additives are characterized by a wide size distribution, and by 62 number, up to 36% of the particles of food grade $TiO₂$ are nano-sized (Weir, 2012). A recent study 63 supported this finding and revealed that in 27 food products and personal care products 10-25% of 64 the number of $TiO₂$ particles are below 100 nm in size (Peters, 2014).

65 A study on seven male subjects using $TiO₂$ anatase particles with a mean size of 160 nm and 380 nm 66 showed that the particles were partially absorbed by the human gut leading to peak titanium levels in 67 blood between 4 and12 hours post oral ingestion (Bockmann, 2000, Pele, 2015). The insolubility of 68 TiO₂ suggests particle uptake. However, a recent study on 9 volunteers with particles sizes of 15nm, 69 100nm and \leq 100nm and \leq 100 nm could not find significant evidence for absorption of TiO₂ nanoparticles after oral 70 application (Jones, 2015).

71 Also studies looking at TiO₂ absorption, retention and toxicity in animal models have led to 72 conflicting results depending on chosen doses, sizes and phase of the TiO2. Difficulties in 73 quantitative analysis when separating the Ti contribution of nanoparticles from a chemically 74 identical background have very likely contributed to this situation, as illustrated by (MacNicoll, 75 2015) who found no evidence of a general translocation of $TiO₂$ nanoparticles after oral application (5mg•kg⁻¹ BW) though the authors could not exclude the possibility based on Ti detection in a few 77 individual animals. Another recent study using 'low' doses (2.3 mg) of TiO₂ nanoparticles (various 78 types ranging from 107-360 nm hydrodynamic diameter) in adult healthy rats led to non-significant 79 increases of Ti in liver and spleen but accumulations in mesenteric lymph nodes (Geraets, 2014). An

80 earlier study of Jani and coworkers found that 12.5 mg/rat of orally administered TiO₂ rutile particles 81 with a mean size of 500 nm could cross the gut walls and accumulate in liver, spleen and lungs (Jani, 82 1994b) while (Tassinari, 2014) found increased Ti levels in the spleen and ovaries of rats and 83 observed DNA damage following oral application of $TiO₂NP$ doses of 2 mg \cdot kg⁻¹ BW (Tassinari, 84 2014).

85 A comprehensive review summarizing the current knowledge on toxicokinetics and toxicological 86 responses after application of TiO_2NP by various routes is that of (Shi, 2013). The authors state that 87 there is not much literature available for orally administered $TiO₂$ nanoparticles.

rapplication of TiO₂NP by various routes is that of (Shi, 2013). The auch literature available for orally administered TiO₂ nanoparticles.
 For Peer Review And Alternative Conflux Co₂NP, one has (i) to quantify the 88 To estimate the risk associated with dietary $TiO₂NP$, one has (i) to quantify their uptake following 89 ingestion, (ii) to study their biokinetics and to (iii) identify organs and tissues of concern. To date 90 there are no suitable robust data available. Hence, we aimed here to investigate the biokinetics of 91 orally applied $TiO₂$ nanoparticles by radiolabeling commercially available $TiO₂$ anatase 92 nanoparticles with radioactive V and selecting a nano-fraction (hydrodynamic diameter 70 nm) 93 which was then applied by intra-esophageal instillation (gavage) to healthy adult female rats. By 94 using γ -ray spectrometry we were able to follow the entire nanoparticle absorption, distribution and 95 excretion for each rat in a fully quantitative manner. Biodistributions of the applied $[48}V]TiO₂NP$ 96 were obtained at the time points of 1h, 4h, 24h, and 7d after application, the same retention time 97 points selected for the intravenous injection study (Part 1) in order to catch fast uptake and slower 98 clearance and relocation effects². 100 **Materials and Methods**

101 **Radiolabeling, suspension preparation and size selection of TiO₂_{NP}**

 \overline{a}

Materials and Methods

5 **URL: http://mc.manuscriptcentral.com/tnan**

 See Materials and Methods where we explain: no further animals were sacrificed for a 28-day biodistribution study after observing in the 7-day experiment that fecal excretion of $\int^{48}V|TiO_2NP$ was already complete after 4-5 days.

Page 7 of 35

Nanotoxicology

102 Two batches of 20 mg ST-01 TiO₂NP were irradiated with a proton beam current of 5 μ A and a 103 proton energy of 13.5 MeV. One yielding an activity concentration of 1.0 MBq \cdot mg⁻¹ (⁴⁸V-activity 104 per $TiO₂NP$ mass) was used for the 1h, 4h and 24h retention experiments. The second one was 105 irradiated on five consecutive days, yielded an activity concentration of 2.35 MBq \cdot mg⁻¹ and was 106 used for the 7d retention experiment. At these radioactivity concentrations the atomic ratio of ^{48}V . Ti 107 in the NP is about 2.6 \times 10⁻⁷ and 6.2 \times 10⁻⁷, respectively. Since proton irradiation and the chemical 108 difference of the radiolabel may result in a non-perfect integration of the ^{48}V in the TiO₂ matrix, the [⁴⁸V]TiO₂NP were repeatedly washed to remove released ⁴⁸V-ions.

the radiolabel may result in a non-perfect integration of the ⁴⁸V in the Tiver repeatedly washed to remove released ⁴⁸V-ions.

was performed in a repeated sequence of nanoparticle suspension, was herformed in a repeate 110 Size selection was performed in a repeated sequence of nanoparticle suspension, ultrasound 111 homogenization, washing by centrifugation and re-suspension in distilled water in order to remove 112 excess sodium pyrophosphate, to eliminate larger aggregates/agglomerates and to minimize the 113 content of free, ionic ^{48}V , as described in the Supplementary Information (SI-GAV). The final size-114 selected and radiolabeled, nano-sized aggregates or agglomerates of $\int^{48}V|TiO_2NP$ were suspended in 115 water.

116 For each of the studied retention time points a new batch of size-selected $[^{48}V]TiO₂NP$ was prepared, 117 characterized and immediately applied in a single dose to four rats for each exposure route, i.e., 118 intravenously, by gavage and intratracheal instillation, which improves the comparability between 119 the exposure routes as the studies were started with the same nanoparticle properties.

Characterization of nanoparticles

122 The hydrodynamic diameter of the size selected $[48V]TiO₂NP$ and the zeta potential were measured 123 in triplicates several times during the size-selection process for control purposes, and prior to 124 application, using a Malvern Zetasizer (DLS, Malvern, Herrenberg, Germany). Samples for 125 transmission electron microscopy, from the aqueous suspension ready for administration, were 126 prepared on glow discharged Formvar® coated 300mesh copper grids, and investigated with a 127 Philips 300 TEM at 60 kV acceleration voltage.

URL: http://mc.manuscriptcentral.com/tnan

128 In order to study the effect of the passage of $\int^{48}V|TiO_2NP$ through the gastro-intestinal tract (GIT) 129 $\int^{48}V\right]TiO_2NP$ suspensions were subjected to pH=2 for 30 minutes to simulate the passage through 130 the stomach followed by additional 2 hours at pH=9 to simulate the passage through the small 131 intestine. The evolution of the hydrodynamic diameter was followed by DLS measurements.

Experimental procedures – Study design

Frocedures – Study design
 For Equality and Solution of $[^{48}V]TiO_2NP$ with five retention time point

or gavage in four rats for each time point, as for the other exposure roughly

in the 7-day experiment that fecal 134 It was planned to study the biokinetics of $\binom{48}{110}$ V]TiO₂NP with five retention time points 1h, 4h, 24h, 135 7d and 28d after gavage in four rats for each time point, as for the other exposure routes. However, 136 after observing in the 7-day experiment that fecal excretion of $\int^{48}V|TiO_2NP$ was already complete 137 after 4-5 days, no further animals were sacrificed for a 28-day biodistribution study; as sketched 138 below.

144 Immediately after the final preparation step the $\int^{48}V|TiO_2NP$ suspensions were applied in a single 145 bolus of about 10μ g of $\int^{48}V|TiO_2NP$ per rat. The time point at 7d was studied with a higher dose of 146 about 30µg in order to preserve sufficient sensitivity in spite of longer radioactive decay, and to 147 reveal also minor redistribution and clearing processes.

148 An additional biokinetics study (MAIN-2) was performed in three other groups of four rats each in 149 order to study the amount of $\int_0^{48} V |TiO_2NP$ which remained in the GIT walls, and could possibly 150 reach systemic circulation at later time points. The accumulation of $\int^{48}V|TiO_2NP$ was investigated 151 after 1h, 4h, and 24h in the walls and chime (contents) of the stomach and of the small and large 152 intestine.

Page 9 of 35

Nanotoxicology

153 In order to investigate the absorption and biodistribution of soluble, ionic ^{48}V an AUXiliary study 154 was performed at 24h and 7d after gavage in four rats each with the purpose of correcting the 155 biodistributions of $\int^{48}V|TiO_2NP$ for contributions of $48V$ -ions possibly released from the 156 $[^{48}V]TiO_2NP$. In order to mimic ⁴⁸V released by $[^{48}V]TiO_2NP$ 0.33 $\mu g \cdot \mu L^{-1}$ ionic Ti(NO₃)₄ was 157 added to carrier-free ionic ^{48}V . The pH value was adjusted to 5. For the experiments 60 µL of 158 solution containing 27 kBq ionic ^{48}V and 20 µg of ionic Ti were administered in each rat. Based on 159 the biodistribution of $48V$ ions, the urinary excretion kinetics after gavage of $48V$ -ions and of 160 $[^{48}V]TiO_2NP$, the biodistribution of $[^{48}V]TiO_2NP$ was corrected for the contribution of ^{48}V -ions 161 according to the mathematical procedure derived in the SI-GAV.

Animals

ition of ⁴⁸V ions, the urinary excretion kinetics after gavage of ⁴⁸^x
the biodistribution of $[$ ⁴⁸VJTiO₂NP was corrected for the contribution
the biodistribution of $[$ ⁴⁸VJTiO₂NP was corrected for the contrib 164 Healthy, female Wistar-Kyoto rats (Janvier, Le Genest Saint Isle, France), 8–10 weeks of age (263 \pm 165 10 g mean body weight $(\pm$ STD)) were housed in pairs in relative-humidity and temperature 166 controlled ventilated cages on a 12h day/night cycle. Rodent diet and water were provided ad 167 libitum. After purchase, the rats were adapted for at least two weeks and then randomly attributed to 168 the experimental groups. All experiments were conducted under German federal guidelines for the 169 use and care of laboratory animals and were approved by the Regierung von Oberbayern 170 (Government of District of Upper Bavaria, Approval No. 211-2531-94/04) and by the Institutional 171 Animal Care and Use Committee of Helmholtz Centre Munich.

 $[^{48}V]TiO₂NP$ suspensions were applied to non-fasted rats by oral gavage in a single bolus. The rats 173 were first anesthetized by inhalation of 5% isoflurane in oxygen until muscular tonus relaxed, then 174 they were fixed with their incisors to a rubber band on a board at an angle of 60° to the lab bench in 175 a supine position. For intra-esophageal instillation (gavage), a flexible cannula was placed into the 176 upper third of the esophagus and the $\int^{48}V|TiO_2NP$ suspension (60 µL) was gently instilled using a 1-177 mL-insulin-syringe (0.4 μ L dead volume) followed by 100 μ L of air to accelerate the suspension 178 into the stomach.

179 After gavage, rats were kept individually in metabolism cages for separate daily collection of urine 180 and feces. At 1h, 4h, 24h and 7d after oral application, rats were anesthetized (by 5% isoflurane 181 inhalation) and euthanized by exsanguination via the abdominal aorta.

Sample preparation and radiometric analysis

184 After application the syringe and cannula used for gavage were collected for measurements of 185 residual $\int^{48} V |TiO_2NP$ retained therein.

FO₂NP retained therein.

Ectrometry, all organs, tissues, carcass and excretions were collect

were measured without any further physico-chemical processing

1, Kreyling, 2014, Schleh, 2012) to obtain quantitative, full 186 For γ-ray spectrometry, all organs, tissues, carcass and excretions were collected and ^{48}V -187 radioactivities were measured without any further physico-chemical processing (Hirn, 2011, 188 Kreyling, 2011, Kreyling, 2014, Schleh, 2012) to obtain quantitative, fully balanced biodistributions 189 of each rat. Since by exsanguination only about 60-70% of the blood volume could be recovered the 190 residual blood contents of organs and tissues after exsanguination were calculated according to the 191 findings of (Oeff, 1955) and the V-activity associated with the residual blood content was 192 subtracted as outlined in SI-GAV.

193 The radioactivity of the samples was measured by γ-ray spectrometry using shielded NaI detectors 194 properly calibrated in γ-ray energy and detection efficiency for the 511keV radiation produced by 195 decaying ⁴⁸V. Samples yielding background-corrected counts in the 511eV region-of-interest of the 196 48 V γ-ray spectrum were considered below the detection limit (DL; \leq 0.2 Bq) when the number of 197 counts was less than three standard deviations of the background counts.

198 Throughout this report, the determined background and decay corrected V-activity values of 199 organs, tissues, blood or excretions are given as percentages of the total applied 1^{48} V]TiO₂NP 200 radioactivity, determined as the sum of all samples prepared from each entire animal, including its 201 total fecal and urinary excretions. These percentages are averaged over four rats in each group and 202 are given with the standard error of the mean (SEM). These raw data were corrected (i) for the 203 residual blood content in organs or tissues after exsanguination and (ii) for the activity contribution 204 of free ^{48}V ions according to the methods presented in the SI-GAV.

Nanotoxicology

205 All calculated significances are based on the One-Way-ANOVA test and the *post hoc* Tukey test. 206 For direct comparisons between two groups, the unpaired t-test was used. $p \le 0.05$ was considered 207 significant.

Results

Physicochemical properties of [⁴⁸ V]TiO2NP

211 The size distributions of the size-selected $\int^{48}V|TiO_2NP$ determined by DLS are presented in Figure 1 212 and indicate a good reproducibility of the size selection procedure. The Z-averages (Table 1) are in a 213 narrow range of 88 ± 11 nm, and the PDI values 0.18 ± 0.04 indicate that the size distributions are 214 polydisperse but with a rather narrow size distribution. TEM investigations after the size selection 215 and dispersion process (Figure 2) revealed approximately spherical aggregated/agglomerated entities 216 of roughly 50 nm in diameter, made up of smaller primary particles.

outions of the size-selected $1^{48}V$ JTiO₂NP determined by DLS are presen good reproducibility of the size selection procedure. The Z-averages (T_i of 88 ± 11 nm, and the PDI values 0.18 ± 0.04 indicate that the size 217 With the known ⁴⁸V-activity concentrations $(1 \text{ MBq} \cdot \text{mg}^{-1} (1h, 4h, 24h)$ and 2.35 $MBq \cdot \text{mg}^{-1} (7d)$ of 218 proton irradiated nanoparticles all determined activity values were converted in $\int^{48}V|TiO_2NP$ mass. 219 The applied ⁴⁸V-activities and corresponding masses of $\int_0^{48} V |TiO_2NP$ are reported in Table 1. Since a 220 fraction of the $48V$ -activity loaded into the syringes was retained there, the effective $48V$ -activity 221 received by the rats presented in Table 1 specifies the dose effectively received by the rats. It was 222 determined from the activity balance over all organs, tissues, carcass and excretions of each rat. The 223 difference between the activity loaded into the syringes and this effective dose matches the 224 determined retained activity in the application equipment.

225 The simulation of the GIT passage by exposing the $\int^{48}V|TiO_2NP$ suspensions to different pH-values 226 (pH=2 for 30 minutes for the stomach passage; and pH=9 for 2 hours for passage through the small 227 intestine) resulted in an increase of the Z-averages from 77nm (PDI = 0.19) before simulated GIT 228 passage to 112nm (PDI 0.14) after simulated stomach passage and to 275 nm (PDI 0.45) after 229 additionally simulated passage through the small intestine (see Figure S3 SI-GAV). The results agree

For Perron Content Set To All Alterta Content Set Transform and the ensideral set of the residual blood retained in order S4). On applied ⁴⁸V-ion doses were excreted in urine (0.44% and 0.34%, resper regans was well bel 230 with the enhanced agglomeration of $TiO₂$ nanoparticles when exposed to simulated gastric fluid 231 observed by (Jones, 2015). **233 Biokinetics of soluble ionic ⁴⁸V** 234 In the auxiliary study 99.13% and 99.31% of the applied doses of soluble ^{48}V -ions were either in the 235 GIT or directly excreted *via* feces after 24h or 7d, respectively (see Figure S4). Only 0.87% and 236 0.69% of the applied V-ion doses were absorbed across the gut epithelium. At both time points 237 about half of the absorbed ^{48}V - ions were excreted in urine (0.44% and 0.34%, respectively). Total 238 uptake in the organs was well below 0.1% and only the carcass consisting of skeleton and soft tissue 239 (the latter defined as non-osseous tissues including muscles, fat, skin, connective tissue, paws) 240 contained 0.32% and 0.24% of the ionic $48V$ at 24h and 7d, respectively. The data on ionic $48V$ was 241 used to correct the biokinetics data after gavage of the $[48V]TiO₂NP$ for $48V$ -release from the

242 nanoparticles as described in the SI-GAV.

Biokinetics of [⁴⁸ V]TiO2NP

245 Most of the gavaged $\int^{48}V|TiO_2NP$ were directly excreted in feces (see Table 2). Only a small 246 fraction of about 0.6% of the applied $\int^{48}V|TiO_2NP$ dose was absorbed across the intestinal barrier 247 during the first hour after gavage. This fraction decreased to about 0.05% after 7 days as illustrated 248 in Figure 3.

URL: http://mc.manuscriptcentral.com/tnan 249 In Table 3 the raw data (%ID) are presented together with the data corrected for the radioactivity 250 attributed to the residual blood retained in organs and tissues after exsanguination as described in the 251 SI-GAV. Following this the activity contributions of free ⁴⁸V-ions were subtracted. In order to 252 estimate this contribution we assume that all ⁴⁸V-activity in urinary excretion is only due to ⁴⁸V-ions 253 since glomerular filtration in the kidneys prevents particles larger than 8 nm from passing into the 254 urine (Choi, 2007). The mathematical execution of this correction is described in the SI-GAV and 255 based on the assumption that the excretion kinetics of ionic ^{48}V is the same in the auxiliary study 256 after application of ⁴⁸V-ions and in the main study with $\int^{48}V|TiO_2NP$ suspensions that may contain **Page 13 of 35**

Nanotoxicology

257 or release ⁴⁸V-ions. The surprising result is shown in Figure S7 (supplementary Information) and 258 shows that after 1 day the activities that can be attributed to free ⁴⁸V-ions and to $\binom{48}{102}$ NP are the 259 same within the error margins.

especially during the first 4h and several data at different time points v
(DL<0.2 Bq) in spite of the high sensitivity of the radiotracer method
measurable accumulation within 1 hour after gavage, which appears to
s, hear 260 In Table 3 the percentages of absorbed $\frac{48}{10}$ TiO₂NP in all major organs, in the carcass and in the 261 blood are presented. These data (corrected for residual blood content and ^{48}V -ions are) are visualized 262 in Figure 4A-C. Due to the low absorption across the gut epithelium, the distribution patterns are 263 very variable especially during the first 4h and several data at different time points were below the 264 detection limit (DL<0.2 Bq) in spite of the high sensitivity of the radiotracer method. Nevertheless, 265 they indicate measurable accumulation within 1 hour after gavage, which appears to be delayed in 266 spleen, kidneys, heart and uterus where measurable accumulation could be observed only after 4h. 267 The retention maximum was reached in spleen, kidneys and heart after 24h. Clearance mechanisms 268 in liver, lung and blood must be effective very early and nanoparticle retention shows declining 269 values from 4h to 7d. In all organs and tissues nanoparticle retention declined after 24h towards the 270 end of the observation period with the exception of kidneys and brain where no further net clearance 271 was observable. While retention in uterus and skeleton went through a maximum after 4h, showing 272 that some net clearance can be achieved, the activity percentage retained in the brain reaches its 273 initial value (after 1h) again after 7d indicating the least efficient clearance mechanism of all 274 investigated organs. The kidneys also showed higher nanoparticle retention after 7d than after 4h, 275 however passing through a maximum after 24h indicating net clearance. The largest V-activity 276 fraction is located in the carcass consisting of skeleton and soft tissues. Separating both 277 compartments shows a retention in the skeleton between 0.03% and 0.15% (w/o free ⁴⁸V-ions) while 278 the retention in the soft tissue declines by an order of magnitude from 0.26% after 1h to below 279 0.02% after 7d. Looking at the organ/tissue concentrations in $\%$ ID•g⁻¹ the clearance from the 280 skeleton is much less effective than for the soft tissue and the concentration in the skeleton is at least 281 10 times higher than in the soft tissue except for the 1h retention data.

282 The concentrations of $\int_0^{48} V |TiO_2 NP$ per gram of organs and tissues are also provided in Table 3 and 283 selected data are visualized in Figure 4D-F. It is remarkable how similar nanoparticle concentrations

284 were in each of the secondary organs over the entire time period. The nanoparticle concentration in 285 the skeleton is of the same order of magnitude as most of the secondary organs, while in soft tissue it 286 is considerably lower.

Distinction of the nanoparticle content in gut walls and chime

289 When the walls of the gut and its contents were analyzed separately, most $\int^{48}V|TiO_2NP$ were 290 detected in the chime and only small fractions of 5.8%, 1.3%, and 0.9% in the intestinal walls, after 291 1h, 4h, and 24h, respectively (Figure 5). Since absorption through the gut wall to blood was <1% for 292 all time points the data indicate either insufficient rinsing of the gut walls, or $\int^{48}V|TiO_2NP$ 293 entrapment in mucosa, or some of the initially retained $\int^{48}V|TiO_2NP$ in the gut walls were secreted 294 back into the gut content for excretion.

Comparison of the biokinetics of [⁴⁸ V]TiO2NP absorbed through the gut epithelium with the biokinetics of intravenously injected [⁴⁸ V]TiO2NP

chime and only small fractions of 5.8%, 1.3%, and 0.9% in the intestiant, respectively (Figure 5). Since absorption through the gut wall to blooms the data indicate either insufficient rinsing of the gut walls, or mucosa, 298 In order to compare the biokinetics of $\int^{48}V|TiO_2NP$ which had been absorbed through the gut 299 epithelium and had reached systemic circulation with those $\int^{48}V|TiO_2NP$ directly administered to the $\rm 300$ blood circulation by IV injection (Kreyling, submitted), the accumulated $\rm [^{48}V]TiO_2NP$ in each organ 301 and tissue were renormalized to fractions of the nanoparticles which had been absorbed through the 302 gut epithelium. This enables a comparison of distribution patterns of some ng of $\int^{48}V|TiO_2NP$ 303 absorbed through the intestinal barriers with some 10000 ng $\int^{48}V|TiO_2NP$ intravenously injected. In 304 both applications, the corresponding retention time points were studied with the same $\int^{48}V|TiO_2NP$ 305 suspension, i.e. with the same physico-chemical properties and concentrations. Figure 6 shows the 306 retention pattern of $\int^{48}V|TiO_2NP$ absorbed through the gut epithelium on the left side and the 307 retention pattern of intravenously injected $[^{48}V]TiO₂NP$ on the right side.

Discussion

Nanotoxicology

FiO₂NP allows us, unequivocally, to address this issue. The $[^{48}V]TiO_2N$
ble when exposed to aqueous acidic and peri-neutral pH environment
er, due to the chemical difference between V and Ti a ⁴⁸V-radiolabel
urfac 314 Whether food additive TiO₂ particles are *at all* absorbed in mammals following their oral ingestion is 315 the subject of an ongoing debate (Disdier, 2015, Geraets, 2014, MacNicoll, 2015). The use of radio-316 labelled $[48V]TiO₂NP$ allows us, unequivocally, to address this issue. The $[48V]TiO₂NP$ proved to be 317 sufficiently stable when exposed to aqueous acidic and peri-neutral pH environments (Hildebrand, 2015). However, due to the chemical difference between V and Ti a 48 V-radiolabel located on the 319 nanoparticle surface or reaching it by diffusion in the $TiO₂$ matrix could be released. Alternatively, a slow dissolution process of the nanoparticles would also lead to a release of ⁴⁸V ions. (Hildebrand, 321 2015) have demonstrated that the release from proton irradiated TiO₂ (P25, Evonik) is around 2.5% 322 after 4 h at pH = 2 and well below 1% even after 7 days at pH = 7. By reference to an auxiliary study 323 on the ingestion of $48V$ -ions alone, we should be able to correct for any $48V$ -release from the $\left[^{48}V\right]TiO_2NP$ that contributed to the analytical signal, based on the rigid, conservative assumption that all ⁴⁸V-activity in urine is only ionic and not particulate. Thus, while most similar in vivo studies 326 use total Ti as a proxy for the fate of TiO₂, we have used ⁴⁸V as a proxy for the fate of TiO₂, which 327 can be detected with high sensitivity. Additionally the detection of $\int^{48}V|TiO_2NP$ by γ-ray 328 spectrometry is not affected by any chemically identical background or specimen preparation. 329 However, the same suspensions that were applied in the intravenous study, where they showed an 330 ionic activity contribution of at maximum 1% of the total retained activity, after gavage result in 331 values of about 50% of free ions after 24h. They are derived from a comparison of urinary excretion 332 data between the auxiliary and the main study and do not depend on any in-vitro assumptions on the 333 stability of the suspensions. A reason for this difference may be a preferential absorption of ions 334 through the epithelial GIT barrier in combination with a much more pronounced release of V-ions from $\binom{48}{110}$ NP in the GIT environment. In agreement with Jones et al. (2015) our DLS study 336 after simulating the ph-conditions of the GIT passage show rather aggregation than dissolution of the

337 nanoparticles, however a certain small-sized fraction of much smaller nanoparticles would not have 338 been discovered by DLS and might have been absorbed and been excreted passing renal clearance. 339 These nanoparticles would erroneously have been attributed to the ionic fraction which 340 overestimates the corrections we apply. However, as can be seen from the data in Table 3, this would 341 not invalidate our findings, but the retention in organs and tissues would be higher than indicated by 342 our conservative 'corrected' data.

the nanoparticle distribution in the whole animal and its excretions a
ineties was obtained, whereas other groups have focused on specific con
to the nanoparticle balance. For the first time differences betwee
doses and n 343 By evaluating the nanoparticle distribution in the whole animal and its excretions a quantitatively 344 balanced biokinetics was obtained, whereas other groups have focused on specific organs without 345 paying attention to the nanoparticle balance. For the first time differences between effectively 346 administered doses and nominal doses loaded into syringes could be noted, quantified and 347 considered. The radiotracer method revealed that up to 50% of the suspended $\int^{48}V|TiO_2NP$ dose to 348 be administered was retained in minimal-dead-space-syringes and cannulas, presumably due to 349 electrostatic adhesion of nanoparticles to plastic surfaces. Such effects are likely to occur in other 350 nanoparticle suspensions as well. They are highly variable, difficult to detect and most likely depend 351 on the materials used and their handling. They might be one reason for variations in reported results.

Our data confirm that already 1h following oral application $\approx 0.6\%$ of the administered $\int^{48}V|TiO_2NP$ 353 had passed through the gastrointestinal tract, reached systemic circulation and were retained in 354 various organs and tissues. The fraction retained in the body (excluding the gastrointestinal tract) 355 dropped within 4h after application to a level of ≈0.2%. This implies that not only absorption but 356 also early excretion mechanisms for $\int^{48}V|\text{TiO}_2N\text{P}$ must be active.

 To be absorbed across the gut and into the body, $\int^{48}V|TiO_2NP$ must first pass the epithelial layer. 358 This may be via M-cell capture (Powell, 1996) or regular epithelial cell endocytosis of the small 359 nanoparticle fraction (< 40 nm size) (Howe, 2014), or by "persorption" through holes left in villus 360 tips as enterocytes are shed. It may even be due to 'reach out' of intestinal dendritic cells, sampling 361 directly from the lumen. Once having passed the epithelial barrier, $\int^{48}V|TiO_2NP$ may then move

15 **URL: http://mc.manuscriptcentral.com/tnan**

Nanotoxicology

362 from the gut to the "body" most likely via the lymphatic network, either as particles alone or within 363 migrating phagocytic cells (Bockmann, 2000, Pele, 2015).

ly described after lung administration (Berry, 1977) (Geiser, 2013,
Therefore, $\binom{48}{111}$ C₂NP surface modification by proteins and/or biometron or role for the transport of gut-absorbed nanoparticles towards circuport 364 Nonetheless, based upon the presented data, a fraction of absorbed particles clearly reaches the 365 bloodstream possibly *via* the lymphatic thoracic duct into circulation. We assume that direct entry of [⁴⁸V]TiO₂NP into gut capillaries is less likely to occur as even large pore permeability via this route 367 is restricted to smaller macromolecules. In contrast, nanoparticle entry into alveolar capillaries has 368 been frequently described after lung administration (Berry, 1977) (Geiser, 2013, Geiser, 2005, Geiser, 2014). Therefore, $\int^{48}V|TiO_2NP$ surface modification by proteins and/or biomolecules seems 370 to play a minor role for the transport of gut-absorbed nanoparticles towards circulation. In this 371 respect it is important to note that the distribution of the $\int^{48}V|TiO_2NP$ in the various organs differs 372 greatly between IV-injection and oral application. The difference must be related to dose and/or 373 "pathway" of entry. Regarding dose, most local tissues and the blood itself have only a relatively 374 low capacity for acute particle uptake via their mononuclear phagocytic system (MPS). In contrast, 375 the liver has a high capacity. The 100-fold higher IV-injected doses saturate the local particle uptake 376 capacity directly after administration in most organs and tissues except the liver which collects 377 almost all of the $\int^{48}V\right|TiO_2NP$. This is precisely what is seen for intravenously delivered [⁴⁸V]TiO₂NP. In contrast, the gut barrier acts to greatly reduce the particulate dose absorbed whilst 379 the cellular and lymphatic systems described above serve to further limit vascular exposure to 380 $[^{48}V]TiO_2NP$. Strikingly, almost all retained $[^{48}V]TiO_2NP$ beyond the gut are in the organ free 381 carcass and are only very gradually released over 7 days. Retention in lymph nodes may possibly 382 explain this, however $\int^{48}V|TiO_2NP$ concentrations in samples of pure hind leg muscle with little 383 lymphoid tissue corresponded well with the integral $\int^{48}V|TiO_2NP$ concentration of soft tissue. The 384 very low levels of $\int_0^{48} V |TiO_2 NP$ that do gradually reach the circulation then appear to impact all of 385 the vascular organs to some extent presumably because their MPS is not saturated at these doses and 386 kinetics of $\int_0^{48} V |T| \cdot O_2 N$ P arrival. The likely influence of the protein corona remains speculative since 387 no *in vivo* data on the protein corona of nanoparticles absorbed through the gut are available.

388 Another interesting feature is the slowly decreasing nanoparticle retention in most organs and 389 tissues. The total nanoparticle retention in the body of about 0.2% after 4h decreases to about 0.05% 390 during the 7-day period. This hints that there may be little transport from the retention sites in the 391 parenchyma of various organs and tissues, pointing to a kind of equilibrium between the organ 392 concentrations and the $\int^{48}V|\text{TiO}_2\rangle$ NP circulating in the blood.

the GIT is very low ($\approx 0.6\%$ of the administered dose after 1h and $\approx 0.2\%$
ions in organs like liver and spleen are even an order of magnitude le
sted $\binom{48}{17102}NP$ contrasts with the results obtained on polystyr 393 The data presented here emphasize that the absorbed fraction of $TiO₂NP$ across the intestinal 394 epithelium of the GIT is very low ($\approx 0.6\%$ of the administered dose after 1h and $\approx 0.2\%$ after 4h), and 395 absorbed fractions in organs like liver and spleen are even an order of magnitude lower. The low 396 uptake of ingested $\int^{48} V |TiO_2 NP$ contrasts with the results obtained on polystyrene nanoparticles 397 (Hussain, 1998) but agrees with our previous study using monodisperse gold nanoparticles (AuNP) 398 of various sizes ranging from 1.4 nm to 200 nm (Schleh, 2012) and with an earlier study using polylysine-lipid dendrimers (Florence, 2000). Previous quantitative uptake studies for TiO₂ have 400 only considered submicron and micro-particles and showed much greater absorption and peripheral 401 distribution (Jani, 1994a). As recently discussed by Powell and co-workers, particle type and several 402 physico-chemical nanoparticle properties may be critical determinants of nanoparticles uptake in the 403 gut (Powell, 2010).

404 The observed difference between an absorbed fraction of $\approx 0.6\%$ of our 70 nm anatase TiO₂NP and 405 of 12% for gavaged 500 nm rutile $TiO₂$ particles found by Jani and coworkers is quite remarkable 406 (Jani, 1994b). Whether the different particle sizes and/or the different crystalline phases, or the 407 strikingly different doses of tens of µg per rat *versus* \approx 3 mg•d⁻¹ per rat applied over 10 days, or the 408 different detection techniques employed are responsible for these large differences remains to be 409 determined.

After 24h the absorption of $\binom{48}{10}$ TiO₂NP observed in the present study was six-fold (p<0.01), higher 411 than that of similar sized, monodisperse spherical AuNP (hydrodynamic diameter 85 nm) used in a 412 previous study (Schleh, 2012).

413 Although it is known that absorption of nanoparticles depends largely on size (Hillery, 1994, Jani, 1994a, Schleh, 2012, Sonavane, 2008), the two nanoparticle preparations of $\binom{48}{1102}$ NP and AuNP

Nanotoxicology

imple assay was a first attempt to estimate the pH-effect of the GIT of measurements of the hydrodynamic diameter after incubation indiagglomerate size increased slightly after incubation in simulated stom agglomeration co 415 with a similar size differ significantly in the amount of absorption. One explanation could be that the 416 TiO2 agglomerates break up in the digestive environment of the GIT resulting in a fraction of [⁴⁸V]TiO₂NP of primary particle size of 7-10 nm, which would probably absorb to a much greater 418 extent. The effect of the digestive conditions in the stomach and small bowel environment on the 419 stability of our TiO₂NP suspension was simulated by incubating the TiO₂NP suspension for 30min at 420 pH-2 followed by 2 hours at pH-9. Although we neglected constituents like digestive enzymes and 421 proteins, this simple assay was a first attempt to estimate the pH-effect of the GIT on nanoparticle 422 stability. The measurements of the hydrodynamic diameter after incubation indicated that the 423 average $TiO₂$ agglomerate size increased slightly after incubation in simulated stomach conditions 424 (∼120nm) and agglomeration continued further in simulated intestinal conditions (∼250nm) (see 425 Figure S3). Thus, in agreement with (Jones, 2015) who used simulated gastric fluid for such 426 simulations, a breakup of $TiO₂$ agglomerates cannot be responsible for the higher absorption with 427 respect to similar sized AuNP. Hence, the absorption of Au and $TiO₂$ nanoparticles across intestinal 428 membranes depends not only on size but also on the nanoparticle material, and possibly other factors 429 such as shape, state of aggregation/agglomeration, surface charge, *etc*., which have also been shown 430 to influence the biodistribution of nanoparticles (Arnida, 2010, Devarajan, 2010).

431 Regarding the biokinetics and accumulation of nanoparticles in secondary organs and tissues, the 432 peak of retained $\int^{48}V|TiO_2NP$ was found after 1 hour, with a maximum retention of 0.53% in the 433 carcass, i.e., in adipose tissue, skeleton, skin, and muscles. This finding is not surprising considering 434 that nanoparticles are able to penetrate adipocytes (Vaijayanthimala, 2009) or muscle cells (Suh, 435 1998, Zhang, 2009) that account for most of the carcass mass. Only small and heterogeneous 436 amounts were found in the lungs and other organs.

437 Earlier we performed another set of studies comparing IV-injection and gavage using a set of six 438 different-sized, monodisperse, virtually insoluble AuNP (1.4 nm, 2.8 nm, 5 nm, 18 nm, 80 nm, and 439 200 nm) (Hirn, 2011, Schleh, 2012). The present results are in qualitative agreement with these 440 studies. For instance, also AuNP were predominantly retained in the liver after IV-injection while 441 liver retention was ten-fold lower after absorption through the gut. Additionally, after gavage the

442 retained AuNP in the carcass dominated the biodistribution pattern, similar to the pattern for 443 $\left[\right]^{48}$ V]TiO₂NP shown in Figure 6.

1, 2012). Whether the nanoparticle uptake results from intracellula
he circulation and/or uptake of extracellularly circulating nanopart
lood proteins and/or biomolecules) cannot be decided from both of ou
Disdier et al. (144 Importantly, low but unambiguously detectable amounts of $\int^{48}V|TiO_2NP$ were found in the brain and 445 in the uterus where they were still detectable after 7 days. Although brain and uterus have tight 446 barriers, $\int^{48}V|TiO_2NP$ entry appears to be possible although we cannot exclude nanoparticle 447 entrapment in vascular endothelia. Similar but even lower retention in the brain was found for 80 nm 448 AuNP (Schleh, 2012). Whether the nanoparticle uptake results from intracellular nanoparticle 449 transport in the circulation and/or uptake of extracellularly circulating nanoparticles (surface-450 modified by blood proteins and/or biomolecules) cannot be decided from both of our studies. Only 451 the studies of Disdier et al. (Disdier, 2015) on the interaction of nanoparticles with the brain by using 452 an *in vitro* blood-brain-barrier model in addition to their animal experiments have claimed to show 453 blood-brain-barrier crossing of $TiO₂$ nanoparticles in rats after oral intake. We also note that we have 454 previously found a 10- to 100-fold enhanced accumulation of 15 nm sized AuNP in lungs, spleen, 455 kidneys, heart and brain, and to a lesser extent in the liver, between 0.5h and 48h after IV-injection 456 of AuNP that were firmly conjugated with albumin as compared to non-conjugated, citrate-stabilized 457 AuNP of the same core size and applied dose (Schäffler, 2014). These results indicate that the MPS 458 in the various organs and blood appears to respond to biomolecular AuNP surface-modifications and 459 cause strong changes in the accumulation pattern as early as 0.5h after application. Slightly lower 460 enhancements (2 to 20-fold) were found when the AuNP were pre-conjugated with apolipoprotein-E. 461 We recognize that the quantity of nanoparticles found in many organs and especially the brain is 462 very low. However, considering that, in many countries, several milligrams of $TiO₂$ are ingested per 463 person per day over decades and given the high biopersistence of TiO2, long term accumulation 464 cannot be excluded. Indeed, while there is some evidence for a moderate or low short-term risk at 465 high enough TiO2 oral doses, long-term biokinetics and toxicological NP studies are still lacking but 466 would be most important for a rational long-term low dose risk assessment.

Conclusions

Nanotoxicology

469 We have shown that the absorption of nanosized TiO₂NP across the intestinal membrane is low (less 470 than 0.6% of the applied dose) but not negligible. Absorption in the gut seems to depend not only on 471 size but also on the nanoparticle material, and probably other physico-chemical factors.

472 Seven days after oral application most organs still retain a fraction larger than 0.001% of the applied 473 dose which corresponds to about 10^7 - 10^8 nanoparticles. In view of the apparently slow excretion 474 kinetics a gradual, and possibly undesirable, accumulation of absorbed, systemically circulating 475 particles in certain cells and organs seems to be a strong possibility for subjects chronically exposed to TiO₂ nanoparticles.

tain cells and organs seems to be a strong possibility for subjects chron
articles.
biodistribution of $1^{48}V$ JTiO₂-nanoparticles retained after passage throuter with the biodistribution determined after intravenous inj 477 Comparing the biodistribution of $\int_{0}^{48}V|TiO_{2}$ -nanoparticles retained after passage through the gastro-478 intestinal barrier with the biodistribution determined after intravenous injection in Part 1 of this 479 study, the biokinetics patters are very different. Thus, intravenous injection appears not to be an 480 adequate surrogate for assessing the biodistribution and potential health effects occurring after oral 481 exposure to TiO2 nanoparticles. The differences probably depend on the doses that reach systemic 482 circulation, the dose rates and possibly the "pathway" of entry into circulation. The effect of the 483 protein corona of the nanoparticles obtained after different routes of application and the effect on the 484 biological response need to be clarified by further dedicated investigations.

ACKNOWLEDGMENTS

487 We are most grateful for in-depth discussions with Prof. Dr. J. Powell and Dr. L. Pele from 488 Cambridge University, MRC Human Nutrition Research, about the physiology of nanoparticle 489 transport from the gut towards blood circulation. We also thank Sebastian Kaidel, Paula Mayer and 490 Nadine Senger from Helmholtz Center Munich for their excellent technical assistance, as well as 491 Antonio Bulgheroni, Kamel Abbas, Federica Simonelli, Izabela Cydzik, and Giulio Cotogno from 492 the EU-Joint Research Center who strongly supported the nanoparticle radio-labeling task. We also 493 express our sincere gratitude to Barbara Rothen-Rutishauser and David Raemy, University of 494 Fribourg, Switzerland, who performed the TEM analysis of the $TiO₂NP$.

Nanotoxicology

- **J, Semmler-Behnke, M & Gibson, N submitted. Part 1: Quantitative

dred).**

ditection anoparticles after intravenous injection in rats *N*

dutchinson, C, Volkert, S, Greenfield, SM, Catterall, A, Thompson, RI

ottedinismo 572 Jones, K, Morton, J, Smith, I, Jurkschat, K, Harding, AH & Evans, G 2015. Human in vivo and in 573 vitro studies on gastrointestinal absorption of titanium dioxide nanoparticles. *Toxicol Lett,* 574 233**,** 95-101. 575 Kreyling, WG, Biswas, P, Messing, ME, Gibson, N, Geiser, M, Wenk, A, Sahu, M, Deppert, K, 576 Cydzik, I, Wigge, C, Schmid, O & Semmler-Behnke, M 2011. Generation and 577 characterization of stable, highly concentrated titanium dioxide nanoparticle aerosols for 578 rodent inhalation studies. *Journal of Nanoparticle Research,* 13**,** 511–524. 579 Kreyling, WG, Hirn, S, Moller, W, Schleh, C, Wenk, A, Celik, G, Lipka, J, Schaffler, M, Haberl, N, 580 Johnston, BD, Sperling, R, Schmid, G, Simon, U, Parak, WJ & Semmler-Behnke, M 2014. 581 Air-blood barrier translocation of tracheally instilled gold nanoparticles inversely depends on 582 particle size. *ACS Nano,* 8**,** 222-33. 583 Kreyling, WG, Holzwarth, U, Haberl, N, Kozempel, J, Wenk, A, Hirn, S, Schleh, C, Schäffler, M, 584 Lipka, J, Semmler-Behnke, M & Gibson, N submitted. Part 1: Quantitative biokinetics of 585 titanium dioxide nanoparticles after intravenous injection in rats *Nanotoxicology***,** 586 (submitted). 587 Lomer, MC, Hutchinson, C, Volkert, S, Greenfield, SM, Catterall, A, Thompson, RP & Powell, JJ 588 2004. Dietary sources of inorganic microparticles and their intake in healthy subjects and 589 patients with Crohn's disease. *Br J Nutr,* 92**,** 947-55. 590 MacNicoll, A, Kelly, M, Aksoy, H, Kramer, E, Bouwmeester, H & Chaudhry, Q 2015. A study of 591 the uptake and biodistribution of nano-titanium dioxide using in vitro and in vivo models of 592 oral intake. *Journal of Nanoparticle Research,* 17**,** 1-20. 593 Oeff, K & Konig, A 1955. [Blood volume of rat organs and residual amount of blood after blood 594 letting or irrigation; determination with radiophosphorus-labeled erythrocytes.]. *Naunyn* 595 *Schmiedebergs Arch Exp Pathol Pharmakol,* 226**,** 98-102. 596 Pele, LC, Thoree, V, Bruggraber, SF, Koller, D, Thompson, RP, Lomer, MC & Powell, JJ 2015. 597 Pharmaceutical/food grade titanium dioxide particles are absorbed into the bloodstream of 598 human volunteers. *Part Fibre Toxicol,* 12**,** 26. 599 Peters, RJ, Van Bemmel, G, Herrera-Rivera, Z, Helsper, JP, Marvin, HJ, Weigel, S, Tromp, P, 600 Oomen, AG, Rietveld, A & Bouwmeester, H 2014. Characterisation of titanium dioxide 601 nanoparticles in food products: Analytical methods to define nanoparticles. *J Agric Food* 602 *Chem*. 603 Powell, JJ, Ainley, CC, Harvey, RS, Mason, IM, Kendall, MD, Sankey, EA, Dhillon, AP & 604 Thompson, RP 1996. Characterisation of inorganic microparticles in pigment cells of human 605 gut associated lymphoid tissue. *Gut,* 38**,** 390-5. 606 Powell, JJ, Faria, N, Thomas-Mckay, E & Pele, LC 2010. Origin and fate of dietary nanoparticles 607 and microparticles in the gastrointestinal tract. *J Autoimmun,* 34**,** J226-33. 608 Schäffler, M, Sousa, F, Wenk, A, Sitia, L, Hirn, S, Schleh, C, Haberl, N, Violatto, M, Canovi, M, 609 Andreozzi, P, Salmona, M, Bigini, P, Kreyling, WG & Krol, S 2014. Blood protein coating 610 of gold nanoparticles as potential tool for organ targeting. *Biomaterials,* 35**,** 3455-66. 611 Schleh, C, Semmler-Behnke, M, Lipka, J, Wenk, A, Hirn, S, Schaffler, M, Schmid, G, Simon, U & 612 Kreyling, WG 2012. Size and surface charge of gold nanoparticles determine absorption 613 across intestinal barriers and accumulation in secondary target organs after oral 614 administration. *Nanotoxicology,* 6**,** 36-46. 615 Shi, H, Magaye, R, Castranova, V & Zhao, J 2013. Titanium dioxide nanoparticles: a review of **Nanotoxicology** 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48
- 616 current toxicological data. *Part Fibre Toxicol,* 10**,** 15. 617 Sonavane, G, Tomoda, K & Makino, K 2008. Biodistribution of colloidal gold nanoparticles after 49 50
- 618 intravenous administration: effect of particle size. *Colloids Surf B Biointerfaces,* 66**,** 274-80. 51
- 619 Suh, H, Jeong, B, Liu, F & Kim, SW 1998. Cellular uptake study of biodegradable nanoparticles in 620 vascular smooth muscle cells. *Pharm Res,* 15**,** 1495-8. 52 53 54
- 621 Tassinari, R, Cubadda, F, Moracci, G, Aureli, F, D'amato, M, Valeri, M, De Berardis, B, Raggi, A, 622 Mantovani, A, Passeri, D, Rossi, M & Maranghi, F 2014. Oral, short-term exposure to 623 titanium dioxide nanoparticles in Sprague-Dawley rat: focus on reproductive and endocrine 624 systems and spleen. *Nanotoxicology,* 8**,** 654-62. 55 56 57 58
- 59 60

Page 25 of 35

Nanotoxicology

Figure 1: Hydrodynamic diameter of the four separately prepared [48V]TiO2NP suspensions used to study the four retention times of 1h, 4h, 24h and 7d (28d not studied for gavage) measured directly before esophageal instillation.

73x52mm (300 x 300 DPI)

Figure 2: Transmission electron micrograph of size-selected TiO2NP sampled immediately after the sizeselection procedure. TEM sample preparation leads to 'clumping' together of aggregates/agglomerates on the support grid.

254x190mm (96 x 96 DPI)

Figure 3: Fractions of the applied 48V-activity that were absorbed through the gut walls, entered systemic circulation and accumulated in secondary organs and tissues. Mean \pm SEM of n=4 rats at each time point.

73x52mm (300 x 300 DPI)

the device device of the state of the st Figure 4-1: Quantified 48V-activity levels, reported as percent of applied [48V]TiO2NP dose (% ID) in various organs and tissues and in total blood at 1h, 4h, 24h and 7d after gavage in panels A-C and concentrations (%ID•g-1 of organ or tissue) in panels D-F. The [48V]TiO2NP content in the residual blood of each organ or tissue was subtracted and additionally the activity attributed to 48V-ion released from the nanoparticles. Mean \pm SEM of n=4 rats at each time point. Significant difference from [48V]TiO2NP retention at 1h: p<0.05 (*);p<0.01 (**).

70x17mm (300 x 300 DPI)

For Performing the CONSTANT CONSTANT CONSTANT CONSTANT CONSTANT CONSTANT ON A SUMPLE CONSTANT O Figure 4-2: Quantified 48V-activity levels, reported as percent of applied [48V]TiO2NP dose (% ID) in various organs and tissues and in total blood at 1h, 4h, 24h and 7d after gavage in panels A-C and concentrations (%ID•g-1 of organ or tissue) in panels D-F. The [48V]TiO2NP content in the residual blood of each organ or tissue was subtracted and additionally the activity attributed to 48V-ion released from the nanoparticles. Mean ± SEM of n=4 rats at each time point. Significant difference from [48V]TiO2NP retention at 1h: $p < 0.05$ (*); $p < 0.01$ (**). $\parallel + \parallel +$ (##One legend for both Figures 4-1 GAV and 4-2 GAV)

67x17mm (300 x 300 DPI)

Figure 5: Fractions of [48V]TiO2NP detected in the gastrointestinal tract. (A) Differentiation of [48V]TiO2NP in the total chime and the GIT walls (5.8%, 1.3%, and 0.9%); (B) fractions of [48V]TiO2NP in the chime and walls of each of the three compartments of the GIT. Mean \pm SEM of n=4 rats at each time point. Significant difference from [48V]TiO2NP retention between chime versus wall of compartment: p<0.05 (*);p<0.01 (**); p<0.001 (***); significant difference from [48V]TiO2NP retention at 1h for chime or wall of each compartment: p<0.05 (+);p<0.01 (++);p<0.001 (+++) .

91x31mm (300 x 300 DPI)

Figure 6: Comparison of [48V]TiO2NP retention and accumulation in secondary organs and tissues after gavage and intravenous injection (Part 1; (Kreyling, submitted). Panel A and B: liver, spleen, lungs; panels C and D: kidneys, heart, uterus, brain; panels E and F: carcass, skeleton, soft tissue and blood. Note that any missing data indicate fractions below the detection limit. Mean \pm SEM of n=4 rats at each time point.

198x215mm (300 x 300 DPI)

Table 1: Physico-chemical properties of the [⁴⁸V]TiO₂NP suspensions used for the four different retention times studied by gavage and the mean values of the applied $48V$ activity (kBq) and mass (µg) of $\int^{48}V$]TiO₂NP effectively received by the rats. Also the mean doses in μ g/g BW are given. Additionally, $\int^{48}V|TiO_2NP$ losses in the syringe and/or cannula are provided as detailed in SI-GAV.

Table 2: Percentages of $[^{48}V]$ TiO₂NP, detected in the GIT and feces

For Periodic Later 13.07 ± 1.22
 For Peer Replaces 22.2
 For Peer Review Only 1.222 ± 1.34
 For Peer Review Only 1.222 ± 6.38
 For Peer Review Only 1.222 ± 6.38
 For Peer Review Only 1.222 ± 6.38
 For Peer Rev

Page 35₁ a⁶{\text{\$5}}. [⁴⁸V]TiO₂NP retention in organs and tissues 1949, 249 h, 24 after gavage. The raw data are presented as retained percentage of the applied activity of $\int^{48}V|TiO_2NP$ (% ID, corrected for decay). The data values after correction for the $\int^{48}V|TiO_2NP$ content in the residual blood present in organs and tissues after exsanguination (without (w/σ)) residual blood content) and additionally for the contributions of free ^{48}V -ions to the biodistribution (w/o free ^{48}V) are also shown. After these corrections the ^{48}V -activity data were converted into $\int^{48}V|TiO_2NP$ concentrations per mass of organ or tissue, given as % ID/g and in ng•g⁻¹. Since the applied $\int^{48}V|TiO_2NP$ doses varied and also were intentionally increased for the 7d group most mass concentrations in ng•g⁻¹ exhibit an increase from 24h to 7d. The values in % ID/g are independent of the applied doses. (< DL = below detection limit). In the last line "% absorbed TiO_2 " the $[{}^{48}V]TiO_2NP$ fractions were normalized to those $[{}^{48}V]TiO_2NP$ which had entered blood circulation; see Supp-GAV.

