

pubs.acs.org/ac

Article

¹ Challenging a Preconception: Optoacoustic Spectrum Differs from ² the Absorption Spectrum of Proteins and Dyes for Molecular ³ Imaging

⁴ Juan Pablo Fuenzalida Werner, Yuanhui Huang, Kanuj Mishra, Robert Janowski, Paul Vetschera,
⁵ Christina Heichler, Andriy Chmyrov, Clemens Neufert, Dierk Niessing, Vasilis Ntziachristos,
⁶ and Andre C. Stiel*

Cite This: http:	s://dx.doi.org/10.1021/acs.analcl	nem.0c01902	Read Online	
ACCESS	III Metrics & More	E Art	icle Recommendations	Supporting Information

7 ABSTRACT: Optoacoustic (photoacoustic) imaging has seen marked advances in detection and 8 data analysis, but there is less progress in understanding the photophysics of common optoacoustic 9 contrast agents. This gap blocks the development of novel agents and the accurate analysis and 10 interpretation of multispectral optoacoustic images. To close it, we developed a multimodal laser 11 spectrometer (MLS) to enable the simultaneous measurement of optoacoustic, absorbance, and 12 fluorescence spectra. Herein, we employ MLS to analyze contrast agents (methylene blue, rhodamine 13 800, Alexa Fluor 750, IRDye 800CW, and indocyanine green) and proteins (sfGFP, mCherry, mKate, 14 HcRed, iRFP720, and smURFP). We found that the absorption spectra do not correlate with the 15 optoacoustic spectra for the majority of the analytes. We determined that for dyes, the transition 16 underlying an aggregation state has more optoacoustic relaxation that stems from the neutral or 18 zwitterionic chromophores and unreported photoswitching behavior of tdTomato and HcRed. We 19 then crystalized HcRed in its photoswitch optoacoustic state, confirming structurally the change in



20 isomerization with respect to HcReds' fluorescence state. Finally, on the example of the widely used label tdTomato and the dye 21 indocyanine green, we show the importance of correct photophysical (e.g., spectral and kinetic) information as a prerequisite for 22 spectral-unmixing for in vivo imaging.

ptoacoustic (OA, also termed photoacoustic) imaging 23 visualizes optical contrast with ultrasound resolution 24 25 (Supporting Information 1), enabling high resolution, real-time 26 in vivo imaging well-beyond the 1 mm penetration depth of 27 typical optical methods.¹ Therefore, it is emerging as an 28 appealing tool to study tumor biology,² inflammatory 29 diseases,³ or developmental processes.⁴ However, in the 30 majority of such studies, OA imaging relies entirely on 31 endogenous absorbing-agents, like blood-hemoglobin, melanin, 32 or lipids. In contrast, targeted labels enable longitudinal, cell-33 specific in vivo imaging, drastically expanding the potential 34 applications of OA imaging.⁵ Several potential agents have 35 been tested for OA imaging, including nanoparticles and 36 targeted dyes (reviewed in Gujrati et al., 2017⁶), as well as 37 transgenic labels, such as fluorescent proteins and switchable 38 bacteriophytochromes (reviewed in Brunker et al., 2017⁷).

³⁹ Despite the fact that some labels were specifically developed ⁴⁰ for OA applications,^{8,9} most label choices and strategies for ⁴¹ their identification in images (unmixing) are based on ⁴² photophysics inferred from independent transmission-mode ⁴³ absorption and fluorescence spectroscopy data, assuming ⁴⁴ convenient generalizability. As shown by our previous work, ⁴⁵ low quantum yield and high molar absorption do not ⁴⁶ automatically translate to a high optoacoustic signal.¹⁰ In this work, we sought to develop a method of spectral analysis that 47 combines absorption and fluorescence with OA spectral data 48 to enable the full characterization of photophysical parameters 49 of OA labels. Such a method helps to identify competing 50 transitions; see Figure 1a. This data is a key to accelerate the 51 fl tailored development of OA contrast agents and improve the 52 identification of agents in imaging data. Lastly, a combined 53 method of spectral analysis would enable the study of various 54 photophysical phenomena simultaneously, such as the dye 55 aggregation, exciton coupling, and energy conversion. 56

We introduce herein a multimodal laser spectrometer 57 (MLS) that measures OA spectra with high precision and 58 spectral resolution, concomitantly with absorption and 59 fluorescence spectra. The high optoacoustic spectral quality, 60 derived from a range of common dyes and chromoproteins, 61 enables identifying significant differences between absorbance 62

 Received:
 May 3, 2020

 Accepted:
 July 8, 2020

 Published:
 July 8, 2020





Figure 1. (a) Photophysics of OA signal generation together with common competing decay channels. Abs: absorbance, S_0 : ground state, S_1 : singlet excited state, T_1 : triplet excited state, VR: vibrational relaxation, ISC: intersystem crossing, IC: internal conversion, Pho: phosphorescence, and Fluo: fluorescence. (b) MLS diagram. (c) Absorbance spectrum, raw, and corrected optoacoustic spectrum of NiCl₂. (d) Linear relation between absorbance and OA signal for brilliant black BN (BBN).

63 and OA spectra. We correlated these discrepancies either to an 64 aggregation state of the dyes or to the isomeric state of the 65 chromophores. Finally, we discuss how such high-quality MLS 66 spectra can empower future optoacoustic contrast agent 67 development and spectral unmixing to achieve high detection 68 sensitivity in molecular imaging.

69 **EXPERIMENTAL SECTION**

70 Extended experimental details can be found in the Supporting 71 Information.

Multimodality Laser Spectrometer (MLS). Based on 72 73 expertise gained measuring the OA kinetic parameters of 74 photocontrollable proteins,¹¹ we developed MLS (Figure 1b) 75 that can record fluorescence, absorbance and high quality OA 76 spectra simultaneously. An essential part of the MLS is a 77 stepwise correction procedure enabling highly accurate OA 78 spectral measurements (Figure 1c, and Supplementary Figures 79 1-5). Key characteristics of MLS are (i) illumination with 80 pulsed lasers as in OA imaging; (ii) homogenized light fluence 81 for all wavelengths; (iii) in-line reference to overcome laser 82 instabilities; (iv) simultaneous detection of fluorescence 83 excited by the laser pulse; and (v) simultaneous absorbance 84 measurement. In contrast to other devices for recording OA ss spectra, which required samples with high concentrations and/ so or large volumes, $^{12-14}$ our MLS requires only 100 μ L of 87 minimum 0.2 optical density (OD) concentration for highquality, reproducible spectra (Figure 1d). 88

Preparation of Dyes. Dyes were dissolved at 5 mg/ml in 90 DMSO and frozen; prior to measurement, a stock solution of 91 the dyes 1% v/v was prepared in 10% fetal bovine serum 92 (FBS). The dye was further diluted to provide different 93 concentrations. The absorbance spectra of each concentration 94 were measured several times before experiments to ensure no 95 change in the absorption spectra over time.

Preparation of Proteins. Proteins were expressed in 97 *Escherichia coli* BL21 and purified by Ni-NTA affinity 98 chromatography, followed by gel filtration on a HiLoad 26/ 99 600 Superdex 75pg column (Amersham Biosciences) in 100 phosphate buffer saline (PBS) buffer. Purified proteins were 101 frozen immediately in liquid nitrogen and stored at -80 °C. 102 Thawed proteins were centrifuged at 14,000 rpm for 45 min at 103 4 °C, and the supernatant was used for the measurements. **Dextran Sodium Sulfate (DSS) Colitis Model and** 104 **Sample Preparation.** Colon inflammation was induced in 10s reporter mice expressing tdTomato under control of the 106 collagen VI promoter as previously described.¹⁵ Mice were 107 euthanized at day 8, when the entire colon was removed and 108 incubated in Roti-Histofix overnight at 4 °C, before proceeding 109 to the measurements. 110

Multi Spectral OA Tomography (MSOT). Colitis model 111 MSOT images were obtained from excised colons on top of 112 agar. The sample was illuminated by nanosecond pulses at 1 113 Hz from a tunable laser (Spitlight-DPSS 250 ZHG-OPO, 114 InnoLas) Ultrasonic detection was performed using a curved 115 64-element array with a central frequency of 5 MHz (Imasonic 116 SAS, Voray, France). Ultrasound data was reconstructed to 117 images using a model-based approach. 118

ICG MSOT data has been recorded using the MSOT 119 inVision 256 (iThera Medical GmbH). MSOT spectra 120 between 680 and 900 nm of freshly euthanized mouse filled 121 rectally with ICG solution 65 μ M were recorded. IThera 122 software ViewMSOT was used for image reconstruction and 123 AMF-spectral unmixing using the ICG absorption and each of 124 our ICG OA spectra. 125

Crystal Structure of HcRed. HcRed was purified as 126 described above. The pure protein was exchanged from PBS to 127 TRIS buffer pH 8.0 containing 300 mM NaCl and 128 concentrated to 5 mg/mL. HcRed crystals grew in 0.1 M 129 BICINE pH 8.5, 22% w/v polyethylene glycol 10,000. X-ray 130 diffraction data set for HcRed was collected to 2.3 Å 131 resolution.

RESULTS AND DISCUSSION

133

Characterization of Dyes. *Methylene Blue and Rhod-* ¹³⁴ *amine 800.* The sensitivity to aggregation of some natural ¹³⁵ chromophores and organic dyes has been exploited for various ¹³⁶ sensing applications in OA imaging. ^{16,17} However, the ¹³⁷ photophysics underlying the OA signal changes have never ¹³⁸ been thoroughly understood, precluding efficient exploitation ¹³⁹ of such effects for the design of next-generation functional OA ¹⁴⁰ labels. Methylene Blue (MB) has been commonly employed as ¹⁴¹ a label in OA. ¹⁸ Rhodamine 800 (Rh800) would be even more ¹⁴² favorable for OA imaging due to its more pronounced red ¹⁴³ absorption (682 nm vs 660 nm of MB); however, it has a ¹⁴⁴ f2

¹⁴⁵ strong fluorescence at 712 nm.¹⁹ Both dyes, depending on ¹⁴⁶ environmental conditions like pH, ionic strength, and ¹⁴⁷ crowding agents, form self-aggregates through strong aromatic ¹⁴⁸ interactions.²⁰ In order to create a commonly defined medium ¹⁴⁹ for our measurements that resemble the cellular environment, ¹⁵⁰ we dissolved the dyes in PBS containing 10% FBS.²¹ In this ¹⁵¹ environment, all dyes are in a mixed aggregation state. Even ¹⁵² more critically, this provides stable environments that enable ¹⁵³ us to observe the spectra without changes over time.

For solutions of MB with concentrations ranging from 17 to 155 29 μ M, we found fluorescence emission only from excitation of 156 the red-shifted spectral band, corresponding to the fluorescent 157 monomer, while the blue-shifted band, associated predom-158 inantly with the H-aggregates (dye molecules stacked in the 159 same orientation), did not show fluorescence (Figure 2a).



Figure 2. Normalized optoacoustic signal (OAS), absorption (Abs), emission (Em), and excitation (Ex) spectra for (a) MB, methylene blue; (b) Rh800, Rhodamine 800; (c) Alexa780, Alexa Fluor 750, and (d) 800CW, IRD_{ye} 800CW. (e) Absorption and (f) optoacoustic spectra of ICG at increasing concentrations.

160 Conversely, the OA spectrum shows a stronger signal for the 161 H-aggregate band (610 nm), reflecting a higher optoacoustic 162 (photoacoustic signal generation efficiency, PGE, see Supple-163 mentary Text 1) for this transition compared to the monomer 164 (slope of 0.61 at 610 nm vs 0.37 at 665 nm in Supplementary 165 Figure 6a). It is generally accepted that the loss of fluorescence 166 in highly symmetric H-aggregates, like the one formed by MB, 167 stems from a favored ISC^{22} (Supplementary Figure 7). This is corroborated by our photofatigue analysis, which shows faster 168 169 bleaching for the H-aggregate at 610 nm (Supplementary 170 Figure 8a) than for the monomer at 665 nm. It is commonly 171 held that the vibronic relaxation following ISC, occurring in 172 the long microseconds range, was considered too slow to contribute to detectable OA signals²³ in our observation time 173 174 window of submicroseconds range and thus would not explain 175 the higher PGE of the H-aggregate. Interestingly the ISC for 176 MB is very fast (ns vs μ s),²² suggesting direct OA detection of 177 the resultant vibronic relaxations. Consequently, we show here 178 that the formation of H-aggregates by MB results in fast ISC,

which contributes to the observed strengthening of the OA 179 signal.

Dye Rh800 has thus far not been considered for applications 181 in OA; however, the increase in the PGE of MB that we 182 observed upon its aggregation prompted us also to study the 183 photophysics of Rh800. We measured Rh800 in a concen- 184 tration range from 30 to 49 μ M. Rh800 shows a similar 185 combination of monomer (695 nm) and predominantly H- 186 aggregate (635 nm) bands, similar to MB (Figure 2b). But the 187 Rh800 PGEs at the two bands are only slightly different (1.14 188 and 1.08, Supplementary Figure 6b). To our surprise, the PGE 189 of the monomer form is higher than our standard BBN (Figure 190 1d), despite its fluorescence, while the bleaching rates of both 191 transitions are moderate and suggest no strong ISC 192 (Supplementary Table 1, Supplementary Figure 8b). The 193 MLS spectra in Figure 2b showed that the fluorescence 194 excitation spectrum is shifted relative to the absorbance peak at 195 695 nm, indicating a third transition, presumably a non- 196 fluorescent J-aggregate (720 nm, arrow). Further inspection of 197 the spectra reveals an additional shoulder around 580 nm 198 (Figure 2b arrow). Such high energy transitions in xanthene 199 dyes are often correlated with higher-order aggregates.²⁴ Given 200 also a pronounced H-aggregate band suggests that most of the 201 dye is in an aggregate state. 202

The acoustic wave for Rh800 at 695nm differs from the one 203 at 635 nm and also from the one of BBN (Supplementary 204 Figure 9). It is likely that this change of the acoustic wave is 205 directly related to the higher-order aggregates. First, the 206 aggregates represent a larger emitter changing the waveform, 207 and second, the aggregates displace the water around the 208 individual dye monomers and thus change the Grüneisen 209 parameter. Both mechanisms favor the OA signals upon 210 deexcitation, thus explaining the unusually high PGE. 211

Alexa Fluor 750 and IRDye 800CW. Alexa Fluor 750 212 (Alexa750) and IRDye 800CW (800CW) show absorbance 213 peaks centered in the favorable blood window (749 and 774 214 nm, Figure 2c,d), which is why they have been developed for in 215 vivo fluorescence imaging. Despite their relatively high 216 fluorescence (quantum yields $QY_{fluo} = 0.12$), both dyes have 217 been used in OA imaging.^{25,26} 218

Analysis of the spectra of Alexa750 shows a shift for the 219 excitation maximum as well as the PGE maximum of 10 nm 220 relative to the absorbance spectrum. This points to the OA 221 transition at ~ 698 nm (Figure 2c), blue shoulder, arrow A). 222 Additionally, this shoulder bleaches less than the main peak, 223 with only 40% loss versus 80% after 60000 light pulses for the 224 shoulder and main peak, respectively (Supplementary Figure 225 8c). This low bleaching and effective deexcitation in 226 comparison to the main peak of Alexa750 points to a strong 227 triplet tendency of the main peak leaving the transition at \sim 228 698 nm as the more effective one for imaging. Similar to the 229 xanthene dyes, the absorbance spectrum of 800CW has two 230 peaks (Figure 2d) with one centered at 710 nm and the second 231 one centered at 770 nm. The excitation spectrum mirrors the 232 primary absorption peak (770 nm), with a stronger PGE in the 233 blue shoulder (710 nm), attributable to H-aggregates. Despite 234 evidence of 800CW showing excitation coupling with other 235 aromatic dyes,²⁷ there is no description in the literature of self- 236 aggregation of 800CW. The observed behavior, spectrally 237 similar to MB and Rh800 described above, might point to the 238 blue shoulder resulting from H-type aggregation for 800CW. 239 Intuitively, the formation of self-aggregates in 800CW seems to 240 be unfavorable due to 800CW's strong negative character (four 241

242 negatively charged sulfonate groups). However, it has been 243 noted that several other similarly charged dyes undergo 244 aggregation and exciton coupling in the presence of macro-245 molecules.²⁸ Thus, the FBS that is present in our solutions 246 could lead to the formation of H-aggregates of 800CW, 247 explaining the observed exciton coupling behavior.

²⁴⁸ *Indocyanine Green.* Indocyanine green (ICG) is a dye ²⁴⁹ widely used in OA due to its strong signal intensity, favorable ²⁵⁰ absorbance at 800 nm, and FDA approval.^{29–32} However, the ²⁵¹ zwitterionic and hydrophobic character of ICG promotes ²⁵² aggregation at very low concentrations.^{33,34} Thus photo-²⁵³ physical properties of ICG vary with concentration, but also ²⁵⁴ with the surrounding proteins that can modify its aggregation ²⁵⁵ state.^{21,34}

OA and absorbance spectra show a constant mismatch 256 257 (Figure 2e,f), with the whole spectral shape exhibiting a 258 redshift and an increase of the 745 nm band with increasing 259 concentration. The PGE for the OA shoulder at 745 nm is 260 0.80. Interestingly the absorbance at 795 nm does not correlate 261 linearly with the OAS (Supplementary Figure 10a). To 262 compensate for this nonlinear behavior, we used two windows 263 for the calculation of the PGEs, allowing a segmented linear fit (Supplementary Figure 10a). The PGE at 795 nm (0.62) is 264 265 significantly lower at concentrations below 10 μ M than it is 266 above 10 μ M (0.87), suggesting a higher PGE for aggregates of 267 ICG. Thus, the difference between the dimer/monomer ratio from absorbance and OA represents the aggregation tendency 268 of ICG (Supplementary Figure 10b). The PGE increases at the 270 monomer peak (795 nm) with increasing concentration 271 because the fraction of ICG aggregates also absorb at this 272 wavelength. The discrepancies between absorbance and OA 273 spectra observed could arise from different species of ICG 274 present primarily at low concentrations, such as monomer, 275 monomer bound to proteins (FBS), dimer, and dimer bound 276 to proteins, as previously suggested.³⁴ At higher concen-277 trations, aggregated ICG interacting with proteins becomes the 278 predominant species, the OA and absorption spectra of which 279 do not vary significantly. Our results for ICG prove clearly that 280 aggregation increases optoacoustic signal generation. Beyond 281 encouraging the production of nanoparticles with highly 282 aggregated ICG for optoacoustic imaging, this also suggests the use of the aggregation states as ratiometric readout. 283

Characterization of Proteins. Genetically encoded labels 284 285 are a prerequisite for longitudinal imaging and one of the 286 reasons for fluorescence imaging becoming a standard tool in 287 life sciences. GFP-like labels are prominent in fluorescence 288 imaging and have also been studied and employed as OA 289 labels.^{5,12,35} However, biliverdin bearing chromoproteins like bacteriophytochromes (BphPs) or phycobiliproteins (Pbp) are 290 advantageous due to their near-infrared (NIR) absorbance.^{10,36} 291 The hydroxybenzylidene imidazolidone chromophores of 292 GFP-like proteins share complex photophysics involving 293 different protonation states of the chromophore (neutral, 2.94 zwitterionic, and anionic) and a plethora of additional effects 295 296 like excited-state proton transfer (ESPT) and long-lived dark states (often involving isomerization along the methine 297 bridge). For a review, see Van Thor et al., 2009.³ 298

299 Classical Fluorescent Proteins. We measured MLS spectra 300 (Figure 3 and Supplementary Figure 11) for sfGFP (an 301 advanced variant of the original Aequorea victoria GFP³⁸), red 302 mKate2,³⁹ mCherry,⁴⁰ tdTomato,⁴⁰ and far-red nonfluorescent 303 HcRed.⁴¹ Comparing their OA and absorption spectra reveals 304 that the blue shoulder of the peak is pronounced in OA.

f3



Figure 3. Normalized optoacoustic signal (OAS), absorption (Abs), emission (Em), and excitation (Ex) spectra for (a) sfGFP, (b) mKate2, (c) mCherry, (d) HcRed, (e) tdTomato, (f) smURFP, and (g) IRFP720 (IRFP). Linear relationship of OAS and Abs in Supplementary Figure 11.

Sometimes even leading to a blue-shift of the OA maxima 305 compared to the absorption. sfGFP shows a maximum for 306 absorbance and OA at 480 nm; however, the shoulder in the 307 blue region of the OA spectra is more prominent, including an 308 identifiable slope rising at ~425 nm for the neutral 309 chromophore (Figure 3a and in Supplementary Figure 11a). 310 This is also reflected in a higher PGE for the blue transition 311 compared to the main peak (425 nm: 0.75; 480 nm: 0.34). 312 mKate2 has an optoacoustic spectrum that is 5 nm blue-shifted 313 for the absorption spectra (585 nm vs 580 nm) and matching 314 PGEs of 550 nm: 0.64 and 580 nm: 0.51 (Figure 3b and in 315 Supplementary Figure 11b). For mCherry, the maximum is 316 similarly blue-shifted compared to the absorbance (580 nm vs 317 585 nm) but with the corresponding PGE only slightly higher 318 for the blue-shifted peak than for the absorbance peak (545 319 nm: 0.54; 585 nm: 0.53 (Figure 3c and in the Supplementary 320 Figure 11c). For the far-red tetrameric protein HcRed, the 321 optoacoustic spectrum is blue-shifted by 20 nm with respect to 322 the absorption spectrum, with the peak center at 570 nm 323 instead of 590 nm (Figure 3d and in Supplementary Figure 324 11d). Accordingly, the PGEs at 570 and 590 nm are 0.97 and 325 0.80, respectively. 326

In general, the PGEs of the main peaks are in good 327 agreement with the reported fluorescent quantum yields 328 (Supplementary Table 1). tdTomato however, a protein 329 related to mCherry, showed an OA spectrum that is highly 330 displaced to the red, a maximum in the OA spectrum at 565 331

332 nm, rather than 555 nm, with a PGE of 0.79 at 485 nm and of 333 0.83 at 565 nm (Figure 3e and Supplementary Figure 11e). Several explanations are possible for the higher PGE of the 334 335 blue shoulder observed in most of the proteins. (i) The blue 336 shift could result from an involvement of the neutral state of 337 the chromophore.^{42,43} It is possible that the deexcitation from 338 this state results in vibronic relaxation and a higher PGE. (ii) 339 For HcRed, our data is in agreement with the theory of a 340 mixture of cis-trans isomers, with a nonfluorescent, neutral 341 trans-isomer centered at around 570 nm and a slightly 342 fluorescent cis-isomer centered at 590 nm.⁴⁴ The high PGE 343 from the peak at 570 nm could arise from the nonfluorescent 344 properties of the trans-isomer. (iii) Discrepancies between OA and absorption spectra have been observed before and 345 explained by ground state depopulation.¹² This effect primarily 346 comes into play for the strongly absorbing main transitions 347 348 leading to the "flattened" OA peaks observed by Laufer and 349 colleagues.¹² Since in our study the discrepancies are regularly 350 found at the flanks of the spectra or only for certain peaks (e.g., aggregate peaks or protonated chromophores), we believe that 351 352 the phenomenon of ground state depopulation does not explain our results. 353

Near Infrared Fluorescent Proteins. Biliverdin binding 354 355 proteins are of critical importance, as their absorption in the 356 near-infrared region gives them superior penetration depth and 357 spectral separation for endogenous contrast. Until now, only two classes of biliverdin binding proteins have been reported: 358 bacteriophytochromes and phycobiliproteins.45,46 IRFP720 359 360 and smURFP are examples of both classes. In both cases, the 361 optoacoustic spectra match the absorption spectra, with PGEs 362 of 0.56 and 0.86 for smURFP and IRFP720 at their primary 363 OA peaks, respectively (Figure 3f,g). The lower PGE of 364 smURFP correlates with its higher quantum yield 0.2 versus 365 0.06 for IRFP720. Additional reductions in PGE could come 366 from faster bleaching and transient isomerization.⁴

³⁶⁷ Photoconversion in Classical Fluorescent Proteins. Un-³⁶⁸ expectedly MLS spectra of several proteins showed a level of ³⁶⁹ photoconversion or photoswitching that was previously ³⁷⁰ unknown. This was initially observed when comparing forward ³⁷¹ (420 to 900 nm) and reverse (900 to 420 nm) modes of ³⁷² collecting spectral data.

For mCherry and tdTomato, the effects are different but and always show a long-lived change of the OA spectra. For mCherry, measuring from 420 to 900 nm results in a stronger blue shoulder than measuring in the opposite direction (Figure are accordingly (Figure 4c), resulting in an increase of the PGE of the shoulder from 0.35 to 0.45).

While for tdTomato, measuring from reversed 900 to 420 380 nm results in a stronger blue shoulder, with a change in PGE 381 for both peaks (Figure 4d and Suppl. Figure 11e), and red-382 shifted maxima of the OA spectra. Alternating illumination 383 probing with 480 nm and photoconverting with 550 nm shows 384 how the absorption spectra changed with an increase of the 385 band at 480 nm, which can be attributed to the neutral 386 chromophore in those proteins (Figure 4e,f). It could be that 387 388 the effects in tdTomato under strong pulsed laser illumination 389 are similar to photoswitching events in related rsCherry and 390 rsCherryRev proteins,⁴⁸ although we see no dark-relaxation.

For the far-red protein HcRed, continuous illumination at 392 565 nm did not produce a decrease in OA intensity 393 (bleaching) as expected but an increase in OA signal (Figure 394 4g–i). Alternating illumination between 565 and 590 nm



Figure 4. Photoswitching effects in mCherry, tdTomato, and HcRed. (a, d, and g) OA spectra from 900 to 420 nm (reverse, red) or 420 to 900 nm (forward, blue) along with the absorption spectra in the nonilluminated state. (b, e, and h) The temporal development of the OA signals at the two peaks. (c) mCherry absorption spectra before and after OA measurements; (f) tdTomato after pulses of 550/480 nm light, and (i) 590/565 nm light for HcRed, respectively. (j) In blue structure of HcRed after 60000 pulses of 565 nm light (6Y1G) and in pink structure of native HcRed (1YZW).

results in a reduction in the signal at the fluorescence peak at 395 590, but an increase at the optoacoustic peak at 565nm; the 396 absorption spectra show the same kinetics. A published crystal 397 structure of HcRed suggests inherent chromophore flexibility 398 with the trans nonplanar isomer responsible for the shoulder at 399 570 nm, and cis-isomer responsible for the peak at 590 nm, 400 with both conformer in equilibrium.⁴⁴

In order to validate our proposed mechanism, we decided to 402 crystalize our photoconverted version of HcRed. Our structure 403 (PDB ID: 6Y1G, Supplementary Table 2) shows that the 404 inherent chromophore flexibility is not present anymore after 405 pulsed illumination (Figure 4j and Supplementary Figure 12). 406 The new structure only has a trans-nonplanar chromophore 407 that is tilted and stabilizes by a new H-bond with a twisted 408 His174. Our structure corroborates that the trans-isomer is 409 responsible for the optoacoustic peak in HcRed and support 410 further our hypothesis that it is the trans chromophore 411 responsible for the higher PGE of the shoulder for sfGFP, 412 mKate2, and mCherry.

Spectral Insights to Improve Spectral Unmixing. 414 Beyond clarifying fundamental photophysical mechanisms, 415 our accurate OA spectroscopic investigations are immediately 416 f5

417 beneficial to the improvement of in vivo imaging. To 418 demonstrate this, we chose the dye ICG and the transgene 419 reporter protein tdTomato since both are widely used labels in 420 fluorescence and optoacoustic imaging. Numerous dye-label-421 ing-procedures and transgenic animals are existing for ICG and 422 tdTomato, respectively, which can immediately benefit from 423 detection strategies improved by our spectral information.

424 As described above, ICG has various aggregation states with 425 different optoacoustic spectra which depend not only on ICG 426 concentration but also on its interacting partners (lipids, 427 protein, etc.), leading to heterogeneity of ICG forms in the 428 hard to define in vivo environment. Nonetheless, most 429 (commercial) unmixing algorithms utilize only the absorption 430 spectrum of diluted monomeric ICG for spectral unmixing 431 (Figure 5a, 1). To illustrate the benefit of spectral-unmixing



Figure 5. (a) ICG spectra used for AMF spectral unmixing in b. (1) Commonly used monomeric ICG absorption spectrum, (2) library of ICG optoacoustic spectra at different aggregation states obtained with the MLS. (b) MSOT images of a mouse colon with ICG solution. Unmixed with the spectra shown in a. Shown is a maximum intensity projection of all slices. (c) Visible OA tomography images of control colon and a colon with acute DSS-induced inflammation from tdTomatoColVI reporter mice. Shown is the mean signal at wavelengths from 450 to 650 nm. (d) tdTomato signal detection after photoconversion with multiple pulses of probing and photo converting illumination, followed by differential spectra unmixing. (e) Exemplary spectra before and after conversion (top) and difference spectra together with known difference spectra from MLS used as "search spectra" for the unmixing. (f) Photoconversion of tdTomato positive pixels showing the shift of the peak ratio upon.

432 with a number of OA spectra of different ICG aggregation 433 states (Figure 5a, 2) we introduce a solution of ICG 65 μ M 434 rectally in a euthanized wildtype mouse scanned with 435 multispectral optoacoustic tomography (MSOT). While 436 unmixing using adaptive matched filter (AMF, ViewMOST 437 3.8.10.04, iThera Medical). Using the classical single 438 absorption spectrum of ICG, we found only signals with a 439 high confidence value in the center of the colon (Figure 5b, 1); 440 the same procedure using ICG spectra of different aggregation 441 states shows high-confidence unmixing in the complete colon 442 (Figure 5b, 2). This is in line with the observation that 443 different regions of the colon are differently picked up by 444 individual spectra, suggesting differences in aggregation in 479

those regions (Supplementary Figure 13). It is likely that ICG 445 remains more diluted in the high-volume colon's lumen (arrow 446 a), and it concentrates in the small volume colonics crypts 447 (arrow b). Thus, unmixing with different ICG aggregation state 448 spectra can additionally carry information on local aggregation 449 states and might allow additional conclusions on the 450 immediate environment in those regions as well as ICG 451 concentration. 452

Spectral unmixing of tdTomato in OA imaging in vivo is not 453 straightforward due to overlapping absorption with hemoglo- 454 bin in tissue. We showed that the illumination used in MLS, 455 which is similar to one used for OA imaging, is sufficient to 456 induce a photoconversion of tdTomato. Using this insight and 457 the knowledge on the spectra before and after photoconversion 458 we could readily facilitate unmixing of tdTomato signals from 459 colons. In short, we recorded uninflamed (control) or DSS- 460 induced, inflamed colons from reporter mice expressing 461 tdTomato under control of the collagen VI promoter 462 (tdTomatoColVI).^{15,49} We collected spectral data from 420 463 to 690 nm using a tomography style imaging device with a 64-464 element ultrasound transducer and 45° tilted illumination. We 465 collected spectra before and after converting tdTomato with 80 466 cycles of 500 pulses of photo converting light (550nm) and 467 500 pulses of probing light (480nm). While the signal from 468 blood made distinction between the inflamed and control 469 colon difficult in plain OA imaging (Figure 5c); unmixing, 470 based on the difference spectra (Figure 5e) allowed one to 471 selectively visualize the inflamed colon (Figure 5d) reflecting 472 accumulation and activation of colon fibroblasts and other 473 COLVI (tdTomato) expressing cells in DSS-induced inflam- 474 matory lesions. Our results are in line with corresponding 475 findings published before.^{35,50} We note that also the 476 conversion itself can be utilized if monitored by collecting 477 data at the peak wavelengths 480 and 550 nm (Figure 5f). 478

CONCLUSION

Our first results using the MLS showed a wealth of insights 480 into the behavior of a number of well-used dyes and proteins. 481 Describing the exact changes in the photophysics of OA signal 482 generation (Figure 1a and Supplementary Text 1) upon the 483 aggregation of MB, Rh800, and ICG will afford new 484 applications for sensors that exploit their environment- 485 dependent aggregation, for example, ratiometric analysis of 486 dye aggregation under different physiological conditions. 487 Finally, our results on ICG shed light on the complex behavior 488 of this molecule, which is so regularly used in OA studies.^{29,30} 489 Cognizance of the strong dependence of PGE on the 490 aggregation of dyes could help researchers to analyze 491 concentration-dependent measurements of ICG more accu- 492 rately. On the protein side, the switching or conversion 493 behavior of mCherry, tdTomato, and HcRed is extremely 494 interesting for GFP-like protein research and warrants further 495 study; however, it also has an immediate impact on employing 496 those proteins in OA applications. In particular, tdTomato, for 497 which there are numerous animal models from fluorescence 498

studies, can be used more effectively for OA measurements by 499 exploiting the transition from 550 to 480 nm and using a dual- 500 wavelength ratio analysis to identify the protein-label in tissue 501 with a strongly absorbing background. 502

Lastly, we could show that accurate measurements for OA 503 labels are a prerequisite for their understanding, their future 504 design, and especially their comparison. With regard to this, 505 the use of standards like BBN for novel OA contrast agents to 506

507 come is highly important for comparability and for 508 empowering researchers to make strategic choices on what 509 labels to use in their experiments.

510 **ASSOCIATED CONTENT**

Supporting Information 511

512 The Supporting Information is available free of charge at s13 https://pubs.acs.org/doi/10.1021/acs.analchem.0c01902.

Extended introduction and methods, process diagrams 514 for the analysis, further OA data and structural data for 515 HcRed (PDF) 516

AUTHOR INFORMATION 517

Corresponding Author 518

- Andre C. Stiel Institute of Biological and Medical Imaging 519
- (IBMI), Helmholtz Zentrum München, D-85764 Neuherberg, 520
- 521
- Email: andre.stiel@helmholtz-muenchen.de 522

523 Authors

- Juan Pablo Fuenzalida Werner Institute of Biological and 524
- Medical Imaging (IBMI), Helmholtz Zentrum München, D-525
- 85764 Neuherberg, Germany; © orcid.org/0000-0001-6651-526 3851 527
- Yuanhui Huang Institute of Biological and Medical Imaging 528 (IBMI), Helmholtz Zentrum München, D-85764 Neuherberg, 529
- Germany; Chair of Biological Imaging, Technische Universitat 530 München, D-81675 Munich, Germany; Oorcid.org/0000-531
- 0002-4774-7449 532
- 533 Kanuj Mishra – Institute of Biological and Medical Imaging
- (IBMI), Helmholtz Zentrum München, D-85764 Neuherberg, 534
- Germany; Chair of Biological Imaging, Technische Universitat 535 München, D-81675 Munich, Germany 536
- Robert Janowski Intracellular Transport and RNA Biology 537 538 Group, Institute of Structural Biology, Helmholtz Zentrum
- 539 München, D-85764 Neuherberg, Germany
- Paul Vetschera Institute of Biological and Medical Imaging 540
- (IBMI), Helmholtz Zentrum München, D-85764 Neuherberg, 541 Germany; Chair of Biological Imaging, Technische Universitat 542 München, D-81675 Munich, Germany 543
- Christina Heichler First Department of Medicine, 544
- Universitaetsklinikum Erlangen, Friedrich-Alexander-545
- Universitaet Erlangen-Nuernberg, 91054 Erlangen, Germany 546
- Andriy Chmyrov Institute of Biological and Medical Imaging 547 (IBMI), Helmholtz Zentrum München, D-85764 Neuherberg, 548
- 549 Germany; Chair of Biological Imaging and Center for
- Translational Cancer Research (TranslaTUM), Technische 550 Universitat München, D-81675 Munich, Germany; 551
- orcid.org/0000-0003-0265-019X 552
- Clemens Neufert First Department of Medicine, 553 554 Universitaetsklinikum Erlangen, Friedrich-Alexander-
- 555 Universitaet Erlangen-Nuernberg, 91054 Erlangen, Germany
- Dierk Niessing Intracellular Transport and RNA Biology 556
- Group, Institute of Structural Biology, Helmholtz Zentrum 557 München, D-85764 Neuherberg, Germany; Institute of
- 558 Pharmaceutical Biotechnology, Ulm University, 89081 Ulm, 559 Germany 560
- 561
- Vasilis Ntziachristos Institute of Biological and Medical Imaging (IBMI), Helmholtz Zentrum München, D-85764 562
- Neuherberg, Germany; Chair of Biological Imaging and Center 563
- for Translational Cancer Research (TranslaTUM), Technische 564 565 Universitat München, D-81675 Munich, Germany

https://pubs.acs.org/10.1021/acs.analchem.0c01902

Complete contact information is available at:

Author Contributions

J.P.F.W. and Y.H. contributed equally. J.P.F.W. and Y.H. built 569 the MLS with help from P.V. in early stages. Y.H. programmed 570 the measurement routine. J.P.F.W., Y.H., and K.M. conducted 571 measurements. R.J. solved the structure. A.C., C.N., D.N., and 572 V.N. contributed to the manuscript. C.K. conducted the colon 573 preparation. A.C.S. J.P.F.W., and Y.H. wrote the manuscript 574 and devised the project. All authors approved the manuscript. 575

Notes

pubs.acs.org/ac

The authors declare the following competing financial 577 interest(s): VN is a shareholder of iThera Medical. 578

ACKNOWLEDGMENTS

The authors want to thank Hong Yang, Dr. Ara Ghazaryan, Dr. 580 Yu-Shan Huang, Dr. Christian Zakian, Dr. Jan G. Laufer for 581 discussions, Sarah Glasl for help with experiments, Jessica 582 Klausmann for technical help, and G. Kollias for providing 583 ColVI-Cre mice. 584

- REFERENCES
- (1) Ntziachristos, V. Nat. Methods 2010, 7 (8), 603-614. 586
- (2) Omar, M.; Schwarz, M.; Soliman, D.; Symvoulidis, P.; 587 Ntziachristos, V. Neoplasia 2015, 17 (2), 208-214. 588
- (3) Aguirre, J.; Schwarz, M.; Garzorz, N.; Omar, M.; Buehler, A.; 589 Eyerich, K.; Ntziachristos, V. Nat. Biomed. Eng. 2017, 1, 68. 590

(4) Vetschera, P.; Koberstein-Schwarz, B.; Schmitt-Manderbach, T.; 591 Dietrich, C.; Hellmich, W.; Chekkoury, A.; Symvoulidis, P.; Reber, J.; 592 Westmeyer, G.; López-Schier, H.; Omar, M.; Ntziachristos, V. Beyond 593 Early Development: Observing Zebrafish over 6 Weeks with Hybrid 594 Optical and Optoacoustic Imaging. BioRxiv; 2019, DOI: 10.1101/ 595 586933. 596

(5) Razansky, D.; Distel, M.; Vinegoni, C.; Ma, R.; Perrimon, N.; 597 Köster, R. W.; Ntziachristos, V. Nat. Photonics 2009, 3 (7), 412-417. 598 (6) Gujrati, V.; Mishra, A.; Ntziachristos, V. Chem. Commun. 2017, 599 600

53 (34), 4653-4672. (7) Brunker, J.; Yao, J.; Laufer, J.; Bohndiek, S. E. J. Biomed. Opt. 601 2017, 22 (7), 070901. 602

(8) Mishra, K.; Stankevych, M.; Fuenzalida-Werner, J. P.; 603 Grassmann, S.; Gujrati, V.; Huang, Y.; Klemm, U.; Buchholz, V. R.; 604 Ntziachristos, V.; Stiel, A. C. Sci. Adv. 2020, 6, eaaz6293. 605

- (9) Chee, R. K. W.; Li, Y.; Zhang, W.; Campbell, R. E. J. Biomed. 606 Opt. 2018, 23 (10), 1. 607
- (10) Fuenzalida Werner, J. P. J. P.; Mishra, K.; Huang, Y.; Vetschera, 608 P.; Glasl, S.; Chmyrov, A.; Richter, K.; Ntziachristos, V.; Stiel, A. C. 609 ACS Chem. Biol. 2019, 14 (9), 1896-1903. 610
- (11) Vetschera, P.; Mishra, K.; Fuenzalida-Werner, J. P.; Chmyrov, 611 A.; Ntziachristos, V.; Stiel, A. C. Anal. Chem. 2018, 90 (17), 10527- 612 10535. 613

(12) Laufer, J.; Jathoul, A.; Pule, M.; Beard, P. Biomed. Opt. Express 614 2013, 4 (11), 2477. 615

- (13) Teng, Y. C.; Royce, B. S. H. J. Opt. Soc. Am. 1980, 70, 557. 616
- (14) Schneider, S.; Coufal, H. J. Chem. Phys. 1982, 76, 2919. 617 (15) Wirtz, S.; Neufert, C.; Weigmann, B.; Neurath, M. F. Nat. 618

Protoc. 2007, 2 (3), 541-546. 619

- (16) Peters, L.; Weidenfeld, I.; Klemm, U.; Loeschcke, A.; 620 Weihmann, R.; Jaeger, K. E.; Drepper, T.; Ntziachristos, V.; Stiel, 621 A. C. Nat. Commun. 2019, 10 (1), 1-9. 622
- (17) Wang, J.; Jeevarathinam, A. S.; Humphries, K.; Jhunjhunwala, 623 A.; Chen, F.; Hariri, A.; Miller, B. R.; Jokerst, J. V. Bioconjugate Chem. 624 2018, 29 (11), 3768-3775. 625

(18) Taruttis, A.; Ntziachristos, V. Nat. Photonics 2015, 9 (4), 219- 626 227. 627

568

566

567

579

585

576

628 (19) Alessi, A.; Salvalaggio, M.; Ruzzon, G. J. Lumin. 2013, 134, 629 385–389.

- 630 (20) Moreno-Villoslada, I.; Fuenzalida, J. P.; Tripailaf, G.; Araya-
- 631 Hermosilla, R.; Pizarro, G. D. C.; Marambio, O. G.; Nishide, H. J. 632 Phys. Chem. B **2010**, 114 (37), 11983–11992.
- 633 (21) Bongsu Jung; Vullev, V. I.; Anvari, B. *IEEE J. Sel. Top. Quantum* 634 Electron. **2014**, 20 (2), 149–157.
- 635 (22) Morgounova, E.; Shao, Q.; Hackel, B. J.; Thomas, D. D.; 636 Ashkenazi, S. J. Biomed. Opt. **2013**, *18* (5), 056004.

(23) Schaberle, F. A.; Rego Filho, F. D. A. M. G. G.; Reis, L. A.;
638 Arnaut, L. G. Photochem. Photobiol. Sci. 2016, 15 (2), 204–210.

639 (24) Moreno-Villoslada, I.; Torres-Gallegos, C.; Araya-Hermosilla,
640 R.; Nishide, H. J. Phys. Chem. B 2010, 114 (12), 4151–4158.

- 641 (25) Lakshman, M.; Needles, A. Nat. Methods 2015, 12 (4), No. iii.
- 642 (26) Chen, Z.; Deán-Ben, X. L.; Gottschalk, S.; Razansky, D. Biomed.
 643 Opt. Express 2018, 9 (5), 2229–2239.
- 644 (27) Zhang, J.; Smaga, L. P.; Satyavolu, N. S. R.; Chan, J.; Lu, Y. J. 645 Am. Chem. Soc. **2017**, 139 (48), 17225–17228.
- 646 (28) Hoffmann, S.; Fuenzalida Werner, J. P.; Moreno-Villoslada, I.; 647 Goycoolea, F. M. New Insights into the Nature of the Cibacron
- 648 Brilliant Red 3B-A Chitosan Interaction. In Pure and Applied 649 Chemistry; Walter de Gruyter GmbH, 2016; Vol. 88, pp 891–904.
- 650 (29) Beziere, N.; Lozano, N.; Nunes, A.; Salichs, J.; Queiros, D.;
- 651 Kostarelos, K.; Ntziachristos, V. Biomaterials 2015, 37, 415-424.
- (30) Wilson, K. E.; Bachawal, S. V; Willmann, J. K. Clin. Cancer Res.
 2018, 24 (15), 3572–3582.
- 654 (31) Weber, J.; Beard, P. C.; Bohndiek, S. E. *Nat. Methods* **2016**, *13* 655 (8), 639–650.
- 656 (32) Kim, C.; Song, K. H.; Gao, F.; Wang, L. V. *Radiology* **2010**, 255 657 (2), 442–450.
- 658 (33) Zweck, J.; Penzkofer, A. Chem. Phys. 2001, 269 (1-3), 399-659 409.
- 660 (34) Philip, R.; Penzkofer, A.; Bäumler, W.; Szeimies, R. M.; Abels, 661 C. J. Photochem. Photobiol., A **1996**, 96 (1–3), 137–148.
- 662 (35) Scheibe, K.; Kersten, C.; Schmied, A.; Vieth, M.; Primbs, T.;
- 663 Carlé, B.; Knieling, F.; Claussen, J.; Klimowicz, A. C.; Zheng, J.; 664 Baum, P.; Meyer, S.; Schürmann, S.; Friedrich, O.; Waldner, M. J.;
- 665 Rath, T.; Wirtz, S.; Kollias, G.; Ekici, A. B.; Atreya, R.; Raymond, E.
- 666 L.; Mbow, M. L.; Neurath, M. F.; Neufert, C. *Gastroenterology* 2019, 667 156 (4), 1082–1097.
- 668 (36) Li, L.; Shemetov, A. A.; Baloban, M.; Hu, P.; Zhu, L.; 669 Shcherbakova, D. M.; Zhang, R.; Shi, J.; Yao, J.; Wang, L. V; 670 Verkhusha, V. V. *Nat. Commun.* **2018**, *9* (1), 2734.
- 671 (37) Van Thor, J. J. Chem. Soc. Rev. 2009, 38 (10), 2935-2950.
- 672 (38) Pédelacq, J.-D.; Cabantous, S.; Tran, T.; Terwilliger, T. C.; 673 Waldo, G. S. *Nat. Biotechnol.* **2006**, *24* (1), 79–88.
- 674 (39) Shcherbo, D.; Murphy, C. S.; Ermakova, G. V; Solovieva, E. A.;
- 675 Chepurnykh, T. V; Shcheglov, A. S.; Verkhusha, V. V; Pletnev, V. Z.;
- 676 Hazelwood, K. L.; Roche, P. M.; Lukyanov, S.; Zaraisky, A. G.;
- 677 Davidson, M. W.; Chudakov, D. M. Biochem. J. 2009, 418 (3), 567-678 574.
- 679 (40) Shaner, N. C.; Campbell, R. E.; Steinbach, P. a; Giepmans, B.
 680 N. G.; Palmer, A. E.; Tsien, R. Y. Nat. Biotechnol. 2004, 22 (12),
 681 1567-1572.
- 682 (41) Gurskaya, N. G. N. G.; Fradkov, A. F. A. F.; Terskikh, A.; Matz, 683 M. V; Labas, Y. A.; Martynov, V. I.; Yanushevich, Y. G.; Lukyanov, K.
- 684 A.; Lukyanov, S. A. FEBS Lett. 2001, 507, 16–20.
- 685 (42) Cotlet, M.; Hofkens, J.; Maus, M.; Gensch, T.; Van der
 686 Auweraer, M.; Michiels, J.; Dirix, G.; Van Guyse, M.; Vanderleyden,
 687 J.; Visser, A. J. W. G.; De Schryver, F. C. J. Phys. Chem. B 2001, 105
 688 (21), 4999–5006.
- 689 (43) Miyawaki, A. Nat. Biotechnol. 2004, 22 (11), 1374-1376.
- 690 (44) Wilmann, P. G.; Petersen, J.; Pettikiriarachchi, A.; Buckle, A.
- 691 M.; Smith, S. C.; Olsen, S.; Perugini, M. a.; Devenish, R. J.; Prescott, 692 M.; Rossjohn, J. J. Mol. Biol. **2005**, 349, 223–237.
- 693 (45) Shcherbakova, D. M.; Verkhusha, V. V. Nat. Methods 2013, 10
 694 (8), 751–754.
- 695 (46) Rodriguez, E. A.; Tran, G. N.; Gross, L. A.; Crisp, J. L.; Shu, X.;
 696 Lin, J. Y.; Tsien, R. Y. Nat. Methods 2016, 13 (9), 763–769.

(47) Braslavsky, S. E.; Ellul, R. M.; Weiss, R. G.; Al-Ekabi, H.; 697 Schaffner, K.; Alekabi, H.; Schaffner, K.; Al-Ekabi, H.; Schaffner, K.; 698 Silvia, E.; Braslavsky Richard, G. Weiss; Hussain Al-Ekabi; Kurt 699 Schaffner, R. M. E. *Tetrahedron* **1983**, 39 (11), 1909–1913. 700

(48) Stiel, A. C.; Andresen, M.; Bock, H.; Hilbert, M.; Schilde, J.; 701 Schönle, A.; Eggeling, C.; Egner, A.; Hell, S. W. S. W.; Jakobs, S. 702 *Biophys. J.* **2008**, 95 (6), 2989–2997. 703

(49) Heichler, C.; Scheibe, K.; Schmied, A.; Geppert, C. I.; Schmid, 704 B.; Wirtz, S.; Thoma, O.-M.; Kramer, V.; Waldner, M. J.; Büttner, C.; 705 Farin, H. F.; Pešić, M.; Knieling, F.; Merkel, S.; Grüneboom, A.; 706 Gunzer, M.; Grützmann, R.; Rose-John, S.; Koralov, S. B.; Kollias, G.; 707 Vieth, M.; Hartmann, A.; Greten, F. R.; Neurath, M. F.; Neufert, C. 708 *Gut* **2020**, *69*, 1269. 709

(50) Knieling, F.; Gonzales Menezes, J.; Claussen, J.; Schwarz, M.; 710 Neufert, C.; Fahlbusch, F. B.; Rath, T.; Thoma, O. M.; Kramer, V.; 711 Menchicchi, B.; Kersten, C.; Scheibe, K.; Schürmann, S.; Carlé, B.; 712 Rascher, W.; Neurath, M. F.; Ntziachristos, V.; Waldner, M. J. 713 *Gastroenterology* **2018**, *154*, 807. 714