Hidden in its color: A molecular-level analysis of the beer's Maillard reaction network

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14 15 **Abstract**

We here report a comprehensive non-targeted analytical approach to describe the 16 17 Maillard reaction in beer. By Fourier-transform ion cyclotron mass spectrometry (FT-ICR-MS), we were able to assign thousands of unambiguous molecular formulae to 18 the mass signals and thus directly proceed to the compositional space of 250 analyzed 19 beer samples. Statistical data analyses of the annotated compositions showed that the 20 Maillard reaction is one of the driving forces of beer's molecular diversity leading to key 21 compositional changes in the beer metabolome. Different visualization methods 22 allowed us to map the systematic nature of Maillard reaction derived compounds. The 23 typical molecular pattern, validated by an experimental Maillard reaction model system, 24 pervades over 2,800 (40%) of the resolved small molecules. The major compositional 25 changes were investigated by mass difference network analysis. We were able to 26 27 reveal general reaction sequences that could be assigned to successive Maillard intermediate phase reactions by shortest path analysis. 28 29 Keywords: FT-ICR-MS, Maillard reaction, Foodomics, Beer metabolomics, Molecular

30 networking, Mass difference network

31 **1. Introduction**

Beer belongs to the oldest fermented beverage in the world (Michel & McGovern, 32 1993). Thousands of years ago, humankind already commenced to purposefully 33 produce durable and nutrient-rich beverages timely concordant with the domestication 34 of cereals (Dietrich et al., 2020). While the shelf life of beer is notably due to hops 35 constituents, the alcohol content and the stable pH value, the raw material's durability 36 is maintained by reducing the water content. The underlying process of malting was 37 widespread in ancient Egypt, where the good taste of heat-treated cereals already was 38 valued (Meußdoerffer & Zarnkow, 2014). It still represents one of the manifold guided 39 processes that make up modern beer brewing, the complexity of which is mirrored in 40 the diverse molecular composition of beer. Beer can be considered as an exceedingly 41 complex organic mixture in an aqueous solution, to which the brewing process 42 43 contributes as considerably as the ingredients themselves. The heat treatment of the carbohydrate source is a unique step that notably lifts the molecular complexity of beer 44 45 from that of other beverages. Malting the grain (steeping, germination, kilning/roasting) leads to a series of chemical reactions that are reflected in the "beer's metabolome". 46

Brewing science and beer analysis has been integrating empirical knowledge about its 47 chemical composition over centuries (Pieczonka et al., 2021). Using numerous 48 analytical approaches including UHPLC-MS, GC-MS and NMR spectroscopy, both 49 targeted and non-targeted strategies described the beer composition with regard to 50 metabolic profiles characteristic for beer types (Duarte et al., 2004), brewing sites 51 (Almeida et al., 2006), beer quality (Lachenmeier et al., 2005), ageing (Rodrigues et 52 al., 2011) or the evolution of hops derived compounds (Haseleu et al., 2010). Recently, 53 our group was able to demonstrate the power the ultrahigh resolution mass 54 spectrometric approach of flow injection Fourier transform ion cyclotron mass 55 spectrometry (FI-FT-ICR-MS) providing a comprehensive picture of the beer's 56 metabolome (Pieczonka et al., 2020). Out of the resolved molecular diversity, 57 58 molecular networks of plant secondary metabolites that differentiate beer types and raw materials used could be made visible and characterized. Research on the driving 59 60 force of chemical changes during the roasting process, the Maillard reaction (MR), is more so dominated by targeted approaches. Brewing research focused on 61 62 understanding the series of complex reactions by studying reaction mechanisms of certain marker molecules and aroma compounds. For example, 5-hydroxymethyl-2-63 64 furfuralaldehyde (HMF) is generated by multiple pathways including caramelization

and the MR starting from numerous possible precursors (Capuano & Fogliano, 2011). 65 By comparison, the formation of maltol, characteristic for eponymous dark malt and 66 beer, only occurs in disaccharide systems favored by stereochemistry and hindered 67 dehydration of respective monosaccharide precursors (Yaylayan & Mandeville, 1994). 68 Many studies followed this approach and studied new non-volatile or aroma active 69 compounds including their formation pathways (Hellwig & Henle, 2010; Hellwig et al., 70 2016; Mavric & Henle, 2006). However, a comprehensive and holistic approach 71 remains inadequately pursued. Comprehensive and molecular-level detection of 72 Maillard derived compounds in beer forms the basis to describe general reaction 73 sequences, driving forces and key intermediates. It carries the potential to guide the 74 75 MR related brewing processes towards desired attributes of the beer as Maillard reaction products (MRPs) play a major role in its organoleptic, physical and chemical 76 77 properties. Melanoidins as MR end products determine the color of beer (Kuntcheva & Obretenov, 1996), they contribute to the stabilization of aroma compounds (Obretenov 78 79 et al., 2002), have foam stabilizing properties (Lusk et al., 1995) and show antioxidative properties (Spreng & Hofmann, 2018). The shelf life of beer is further 80 81 increased due to the inhibition of bacterial growth (Dack et al., 2017). Overall, beer quality could benefit from optimizing the MR not only towards the formation of a few 82 targeted molecules, but addressing and eventually controlling the entire compositional 83 space, including the many still unknowns. 84

We have recently developed an analytical pipeline based on high-resolution mass 85 spectrometry and data visualization that allows to comprehensively study the early 86 Maillard reaction network on a molecular level in sugar-amino acid model systems 87 (Hemmler et al., 2019; Hemmler et al., 2017). Studying exact mass differences as 88 proxy for the reactome was shown to be a valuable tool to monitor the formation of 89 MRPs and to better understand their chemical interplay. In this study, we apply this 90 analytical strategy to better understand non-enzymatic browning reactions in beer. We 91 92 aim to capture the huge diversity of the beer metabolome, assess the contribution of MR products and extract related accurate masses. Visualization and integration of 93 molecular compositions into molecular networks will enable us to capture a 94 comprehensive picture of the Maillard reaction as it may occur in the (bio)chemically 95 96 complex beer system.

98 2. Materials and methods

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100 2.1 Beer samples and Maillard model system

A total of 250 samples of bottled beers from over 40 different countries were analyzed. They represent the variety of different beer styles, fermentation types (lager, wheat, craft, geuze, abbey) and raw materials available. The samples were purchased at local grocery stores between 2018 and 2020 and stored at -20 °C prior preparation for analyses.

For the model system, the concentration of 19 amino acids, accessible for 106 derivatization with o-phthaldialdehyde, and 5 saccharides were analyzed in a biological 107 triplicate and technical duplicate of green malt as described in Supplementary Table 108 S1. The concentration of the amino acids and sugars, as described in Supplementary 109 Table S2, were recreated in Milli-Q purified water (Merck Millipore, Darmstadt, 110 Germany) immediately prior to thermal treatment. The concentration of all amino acids 111 added up to 0.12 M and the sum of saccharides' concentration was 0.26 M. The sample 112 113 was heated in a closed glass vial until the increase in mass features flattened out and the final phase of the MR was reached (20 hours at 100°C). The model system was 114 115 created and measured in triplicates.

116 2.2 UV-Vis measurements

The beer samples and Maillard model system were diluted 1:40 in Milli-Q purified water
and centrifuged (14.000 rpm, 4 min.). An aliquot of 100 μL of the supernatant was used
for UV/Vis analysis in Nunc UV-transparent 96-well microtiter plates (Thermo Fisher
Scientific, Waltham, MO, USA). The absorption values at 294 nm and 420 nm were
measured on a Multiskan Sky UV-Vis reader (Thermo Fisher Scientific, Waltham, MO,
USA) with temperature control (23°C).

123 2.3 FI-FTICR-MS measurements

High-resolution mass spectra were acquired on a Bruker solariX ion cyclotron resonance Fourier transform mass spectrometer (Bruker Daltonics GmbH, Bremen, Germany) equipped with a 12 Tesla superconducting magnet (Magnex Scientific Inc., Yarton, GB) and a APOLO II ESI source (BrukerDaltonics GmbH, Bremen, Germany) operated in negative ionization mode. To minimize ion suppression while allowing detection of a maximum number of monoisotopic signals, we carefully optimized

sample dilution. Best compromise could be achieved, when beer samples and model 130 systems were diluted 1:500 in methanol prior to injection into the micro electrospray 131 source. The samples were measured over a period of 24 months in randomized order 132 using a representative lager beer as guality control. 80 % of all detected monoisotopic 133 signals could be assigned to a molecular formula within an error range of ± 0.2 ppm 134 and the mass resolution was stable at 400,000 at m/z 400 between and within 135 measurement days. The used reagents, sample preparation and instrumental 136 parameters are given in Supplementary Table 3. 137

138 2.4 FT-ICR-MS Data processing

The FT-ICR spectra were exported to peak lists with a cut-off signal-to-noise 139 ratio (S/N) of 6 using DataAnalysis 4.2 software. Only singly charged ions were 140 included. Spectra were first externally calibrated by ion clusters of arginine prior to 141 internal calibration by a calibration list of 2000 compositions commonly found in beer. 142 Possible space charge effects were recalibrated by mass difference mapping (Smirnov 143 et al., 2019). Processing and filtration of the peak lists (FT-side loops and isotopologue 144 filtering) were performed by an in-house R-based software tool on basis of single 145 spectra. Peak alignment was performed within a threshold of 0.5 ppm as described by 146 Lucio et al. (2001). Thereby an overall matrix of 11,500 masses was created. To obtain 147 molecular formulae, the accurate masses were subjected to mass difference network 148 149 (MDiN) analysis using the in-house NetCalc software tool (Tziotis et al., 2011). The network calculation was repeated five times and coinciding formula assignments were 150 151 kept, which led to approximately 9,500 unambiguous molecular formulae in the CHNOSPCI space. [M+CI]⁻ adducts were converted into the respective [M-H]⁻ ion. Of 152 153 those, all annotations that are featured in at least three beers were kept for statistical analysis (6,750). A full mass difference statistic was computed on the theoretical 154 neutral masses of each sample. The set of unique mass differences existing within all 155 full mass difference graphs was computed and the relative abundancies of each mass 156 difference was obtained. Mass differences that occurred at least 100 times in a single 157 beer sample (15,500) were used for further statistical analysis (PCA, OPLS) on the 158 relative abundancies of each mass difference within the different samples. 159

160 2.5 Statistical analyses

Firstly, we used an unsupervised Principal Component Analysis to separate the beer samples based on the molecular signatures that determine the biggest variance. In the

second step, an OPLS-DA was performed to extract the molecular pattern which 163 correlates with the absorption at 294 nm. The Hotelling's T² test (95%) was applied to 164 prohibit the influence of strong outliers on the models. The lists of the most important 165 masses and mass differences were defined choosing the highest loadings values. The 166 top characteristic masses were selected within the 90th percentile (674 masses for each 167 dark and pale beers) and referred to as dark and pale markers in the following, 168 respectively. The goodness of the fit and of the prediction were evaluated with the R2Y 169 and Q² values. To exclude overfitting, we computed the p-value of the Cross-Validation 170 Analysis of Variance (CV-ANOVA). The same approach was carried out with the 171 relative abundancies of mass differences occurring in the beer samples. Additionally, 172 173 based on the robustness of the models, we performed a prevision on the Maillard 174 model system. The recognition of compositional pattern could verify the MR origin of 175 the found patterns and set both models in relation. Those elaborations were done in SIMCA 13.0.3.0 (Umetrics, Umeå, Sweden). The statistical parameters of the beer 176 177 samples and Maillard model system (Supplementary Table S4) and PCA and OPLS models (Supplementary Table S5) can be found in the Supplementary information. 178

179 2.6 Mass difference network analysis

Besides the mass difference network that was used for the annotation of the FT-ICR-180 MS data (FT-ICR-MS data processing), a second MDiN was created, which includes 181 all compositions found in both the beer samples and the model system. These nodes 182 were connected by edges representing transformations from the Hodge's scheme 183 (Hodge, 1953) and expanded by reactions including MR fission products 184 (Supplementary Table S6). They are referred to as small Maillard intermediate phase 185 reactions and mass differences in the following. In total ~65,000 connections were 186 received. Based on this second network, the nine most significant compositional 187 changes elucidated by OPLS statistical treatment of the first full MDiN were broken 188 down into smaller individual reaction sequences. More precisely, we computed the 189 shortest paths connecting any source-target pair of the statistically significant, 190 composite mass differences using the unweighted Dijkstra algorithm in the Python 3.7 191 programming environment on a compatible network library (Hagberg et al., 2008). For 192 193 each statistically significant mass difference, a dominant combination of small reactions of the modified Hodge's scheme was determined. The chronological orders 194 of the individual reactions were compared, giving us a dominant reaction sequence. By 195

this approach, we received a chronological reaction sequence that build up the tenstatistically most significant compositional changes during the MR.

198 2.7 Data visualization

The marker formulae were depicted in van Krevelen. By plotting H/C versus O/C atomic 199 ratios it is possible to depict common compositional patterns within observations' 200 markers (Hertkorn et al., 2008). The degree of unsaturation of the compositions was 201 calculated as double-bond equivalents (DBE, sum of rings and double bonds in a 202 molecule) and plotted against the number of carbons. A modified Kendrick mass defect 203 analysis (Kim et al., 2003) was applied to visualize the role of dehydration reaction 204 cascades in both marker subsets. The DBEs, modified KMDs and length of 205 homologous series were calculated as described recently (Hemmler et al., 2018; 206 207 Hemmler et al., 2017). The assignment of corresponding chemical spaces to markers' compositions, their number of nitrogen and their number of oxygen atoms were plotted 208 according to the respective frequency. The developed mass difference network was 209 visualized by the open accessible Gephi Viz Platform (Bastian et al., 2009) using the 210 211 Force Atlas algorithm.

212 3. Results

3.1 Contribution of the MR to the beer's molecular complexity

In our study, we investigated the chemical diversity of a total of 250 bottled beer 214 samples that cover the many facets of beer brewing by FI-FT-ICR-MS. As shown in a 215 previous study (Pieczonka et al., 2020) our non-targeted analytical approach can 216 resolve the entire molecular complexity of beer in a single measurement. Covered 217 compounds include carbohydrates, peptides, lipids, polyphenols, hop bitter acids, 218 sulfates and phosphates as well as mostly yet inadequately characterized Maillard 219 reaction products (MRP). The richness and diversity of the selected beer samples 220 capture the great chemical space of the beer metabolome and provide a well-suitable 221 222 basis to study the contribution of the MR. We were able to assign 7,000 unambiguous molecular compositions to the accurate monoisotopic masses (Fig. 1A) within the 223 224 sample set reaching from very dark (Fig. 1B) to very pale (Fig. 1C) beers (EBC color values reaching from 5 to 150, Supplementary Table S1). The *m/z* values reached from 225 100 to 1000. The molecular formulae were annotated in the CHNOSP chemical space 226 and subjected to further statistical analyses. 227

We used Principal Component Analysis (PCA) to assess the impact of MRPs on the 228 molecular beer composition (Figure 2A). The unsupervised statistical treatment reveals 229 the greatest molecular differences between the beer samples as well as their 230 underlying brewing principles and techniques. The PCA score plot was colored 231 according to each beer's absorption at 294 nm, measured by UV-Vis spectroscopy and 232 reported characteristic to follow the evolution of MR (Yu et al., 2012). The plot reflects 233 the samples' degree of browning with the tendency to lower left positions. Therefore, 234 non-enzymatic browning can be considered to be of major importance for the 235 chemodiversity in beer. It leads to key compositional changes already visible in 236 237 unsupervised statistics.

238 We applied a second statistical analysis, a supervised OPLS-DA, to generate in depth knowledge of compositions driving the differentiation of dark beers (Figure 1B) and 239 240 pale beers (Figure 1C). Compared to PCA, OPLS-DA allowed the extraction of accurate masses without an influence of orthogonal metabolic information, which does 241 242 not contribute to the compositional changes affected by the MR. The received R2Y-, Q²- and ANOVA p-value indicate a highly significant multivariate model (Eriksson et 243 244 al., 2008; Golbraikh & Tropsha, 2002; Westerhuis et al., 2008) (Supplementary information Table S5). The gradient of absorption values, already visible in the PCA 245 and established as driving Y-variable, is reflected in the first component of the OPLS 246 score plot (Figure 2B). The comparison of both statistical models' loading plots shows 247 that the OPLS is capable to extract the same features that drive the MR related 248 separation of the beer samples in the PCA (Figure 2C). We further analyzed an 249 experimental Maillard reaction model system and integrated the results into the OPLS-250 251 DA. According to the amino acid and carbohydrate profiles and concentrations of analyzed green malt (Supplementary Table S2), we designed the MR model system, 252 which we heated to 100°C in order to simulate the processes during malting and 253 brewing. To a certain extent, this model represents Maillard reactions between multiple 254 255 sugars and amino acids in beer. The experimental model system allowed us to validate the assumption that monitoring the absorption at 294 nm can be used to study the MR 256 in beer. The prediction of the model system's position in the OPLS score plot locates 257 it to the far right validating that our OPLS model is capable to recognize the intrinsic 258 259 nature of Maillard derived complex systems (Figure 2B). The MR molecular pattern in beers, which is extracted by the statistical treatment and classified with regard to the 260 261 compounds' significance, matches the chemical space of the MR model system (Supplementary Fig. S2). We could reproduce 80% of the most significant compositions found in beer (90th percentile of most positive loadings) in the saccharide and amino acid experimental model system (Supplementary Figure S2). The overlap between the masses found in beer and those of the model system decreased with decreasing loading values of the respective masses. In comparison, compositions characteristic for pale beers (90th percentile of most negative loadings) showed an overlap of less than ten percent.

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3.2 The compositional nature of the MR in beer

The OPLS loadings plot allowed to extract compositions related to the MR from the rich diversity of beer metabolites and rank them according to their significance. To study the molecular pattern of MRPs, we focused on the top ten percent (90th percentile) of the most significant marker compositions for both the dark and pale beer characteristic. Yet, the typical compositional pattern of the Maillard reaction, reported by Hemmler et al. (2017) and reflected in the MR model system, pervades at least 40% (2.800) of all annotations (Supplementary Figure S1).

278 Several plots and visualizing tools can be used to depict and describe the compositional nature of complex (bio)chemical systems (Hemmler et al., 2019; 279 Hemmler et al., 2018; Hertkorn et al., 2008; Kim et al., 2003). The annotations of the 280 dark beer markers are almost exclusively limited to the CHO (52 %) and CHNO (48 %) 281 space (Figure 3A-I). The number of molecular formulae that contain nitrogen atoms 282 decreased linearly with the number of nitrogen atoms which implies a compositional 283 space built up by chemical kinetics (Figure 3B-I). The frequency of molecular formulae 284 is gaussian-like distributed against the number of oxygen contained, but lacks 285 compositions with less than four oxygen atoms (Figure 3C-I). Compounds with very 286 low oxygen numbers that can be detected by FT-ICR-MS in negative electrospray 287 mode are most commonly annotated as fatty acids or lipids (Pieczonka et al., 2020; 288 289 Schmitt-Kopplin et al., 2019). Such compositions can be found in the marker masses of pale beers (Figure 3C-II). Overall, in contrast to the dark beer markers, the plots of 290 beer metabolites that are characteristic for pale beers and do not come from the MR 291 do not share a distinct compositional space, as comparatively described in other 292 fermented beverages missing heat load (Gougeon et al., 2009; Roullier-Gall et al., 293 2014) (Figure 3A-C-II). 294

Furthermore, the comparison of marker masses of pale and dark beer markers in the 295 van Krevelen diagram shows substantial differences (Figure 3D). The Maillard reaction 296 leads to a highly organized compositional pattern of compounds which is mainly formed 297 through consecutive dehydration, carbonyl cleavage and redox reactions (Hemmler et 298 al., 2018) Interestingly, the extracted molecular formulae of the dark beer marker 299 masses indicate the same compositional pattern. Compositions corresponding to well-300 known MRPs like 5-hydroxymethylfurfural (HMF, C₆H₆O₃), pronyl-lysine (C₁₅H₂₄N₂O₆) 301 or Maltosine (C₁₂H₁₈N₂O₄) as well as early intermediates like desoxyosones (e.g. 302 C₆H₁₀O₅) and Amadori rearrangement products (deoxyhexosylglycine, C₈H₁₅NO₇) can 303 304 be found in both the model system and the dark beer markers. This systematics is 305 contrasted with the van Krevelen diagram of pale beer compounds (Figure 3D-II). The generally more saturated molecular formulae do not cluster in a discrete area. Merely, 306 307 the areas in the van Krevelen diagram indicate thermolabile lipids and peptides (Pieczonka et al., 2020) which may function as MR precursors. The degree of 308 309 unsaturation of MR derived compounds, expressed as double-bond equivalents (DBE), follows a highly systematic structure compared to markers for pale beers (Figure 3E). 310 311 Only a group of early MR intermediates of higher-chain saccharides (e.g. C₂₄H₄₀O₂₀, C₂₄H₃₈O₁₉, C₂₄H₃₆O₁₈ at C>20 and DBE<8) resist the clear, almost linear trend of higher 312 DBEs for higher masses (Fig. 3E-I). The biggest difference between the highly 313 structured MR compositions, which are based on defined chemical reaction cascades, 314 and other beer metabolites are shown in the modified Kendrick mass defect (KMD) plot 315 (Figure 3F). Both, in the CHO and CHNO chemical space homologous series of water 316 elimination reactions can be observed. The maximum length of water elimination 317 cascades equals seven with an average of 3.9. This is in agreement with values for 318 MR models reported in literature (7 and 3.9) (Hemmler et al., 2017) and values 319 computed for our model system (8 and 4.0). By contrast, the pale beer markers do not 320 exceed a homologous series of more than three consecutive dehydration events. 321

By these visualization methods, we could confirm the MR origin of hundreds to thousands of compositions in beer attributed significant for darker beers in the statistical data evaluation and describe their intrinsic compositional structure. The modified KMD plot furthermore implies that the reaction cascade of the MR is captured in the marker compositions.

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328 3.3 The Maillard reaction molecular network in beer

To get deeper insights into the Maillard reaction cascade that leads to the deciphered 329 molecular complexity, we applied a mass difference network (MDiN) analysis. Based 330 on the relative abundancies of mass differences that connect the elementary 331 compositions of each sample and represent chemical reactions or reaction sequences, 332 both PCA (Figure 2D) and OPLS (Figure 2E) statistical analyses were used. Similar to 333 the statistics on the compositions above, the PCA plot shows a gradient of darkening 334 colors with the tendency to lower positions of the beers in the score plot (negative PC2 335 values). Using the absorption values of beers at 294 nm as Y-variable in an OPLS-DA, 336 we were able to extract the most significant mass transformations for dark beer 337 samples. This is in agreement with the mass differences (MDs) driving the separation 338 339 in the PCA (Figure 2F). Again, the MDs match the dominant ones of the Maillard model system (Figure 2E). 340

341 These exact mass differences can be equated with changes in the molecular formulae and therefore compositional changes. They describe the compositional change a 342 343 source compound undergoes to build a target composition. The ten most significant compositional changes are almost exclusively limited to the CHO chemical space and 344 345 reach from 68 Da to 154 Da. Based on the shifts in the respective molecular formula, there are no single reaction equivalents that describe these changes. Consequently, 346 they rather represent (reaction) sequences of individual smaller compositional changes 347 and are referred to as composite mass differences in the following. 348

The van Krevelen diagram of the reaction pairs show that higher weight reactants appear in the area of early MR products (Figure 4A). The associated compounds with lower *m/z* values can be assigned to the area of unsaturated advanced MRPs (Figure 4B). Accordingly, source compounds of the composite reactions have higher masses than the target compounds. The most significant reactions could be defined as degradation processes.

To decipher the individual reactions, a MDiN analysis was applied on all annotated 355 356 compositions (N = 7000). The nodes in the network shown in Fig. 4C represent compositions annotated in both beer samples and the model system. The 357 compositions are connected by edges representing the mass differences typical for the 358 MR intermediate phase. This includes transformations, such as dehydration, 359 360 decarboxylation, and carbonyl cleavage reactions (full list of 11 transformations see Supplementary Table 1). Due to the lack of a universally applicable nitrogen-containing 361 362 mass transition, the tenth MD was omitted. We were able to connect the majority

(>95 %) of source-target pairs of the statistically significant composite mass 363 differences by individual small reactions and define the shortest paths by the 364 unweighted Dijkstra algorithm. For each big compositional change, a certain 365 combination of intermediate phase reactions was dominant (Table 1). The 366 chronological order of the respective individual Maillard intermediate phase reactions 367 was compared (Supplementary Figure S3). The order can be assumed to represent 368 the evolution of the composite MR compositional changes. With up to 175 different 369 chronological orders, for each composite mass difference one reaction sequence was 370 371 very dominant. An overview of the ten most significant compositional changes and their break down into chronological reaction sequences is given in Figure 5. They share a 372 373 similar structure: all feature a dehydration cascade, whereas most of them end with a 374 decarboxylation reaction. Fission products of early MR intermediates such as glyoxal, 375 methylglyoxal and diacetyl mark the beginning of the reaction sequence in many cases.

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377 **4. Discussion**

The progress of the early MR was followed by the absorption at 294 nm. The UV 378 absorbance at 294 nm is commonly used to indicate Maillard reaction products of the 379 intermediate phase (Yu et al., 2012). Absorption values of the beer samples measured 380 at 294 nm (MR intermediates) and 420 nm (advanced MR products) showed a very 381 strong correlation (Pearson correlation coefficient: 0.98). Consequently, our identified 382 marker candidates include MRPs from the entire reaction network (initial, intermediate 383 and final MRPs). The MR-correlating compositions lead to a differentiation of the beers 384 already in the first principal components of the unsupervised statistical analysis. It 385 386 shows that the reaction of sugars and amines define a large part of the beer metabolites. Besides the OPLS statistical parameters ($R^2 > 0.92$, $Q^2 > 0.79$ and 387 388 ANOVA p-value << 0.05), the decreasing coverage of the marker's chemical space by the MR model system with decreasing loading values confirm the power of our 389 390 approach. The typical Maillard reaction signature (Hemmler et al., 2017) is dominant and shows up to at least 40% of the whole chemical diversity resolved by FT-ICR-MS. 391

Different plots and visualization techniques confirm that the markers we found represent a highly systematic and distinct chemical space within the big variety of beer metabolites. Consistent with literature findings, the CHNOS chemical space did not significantly contribute to the universal signature of the MR in beer. This agrees with

low cysteine and cystine concentrations reported in malt, wort and beer (Otter & Taylor, 396 1976). Confirmatory, an inhibiting effect on the progression of the MR and the formation 397 of final MRPs is described for sulfur containing amino acids (Friedman & Molnar-Perl, 398 1990). The difference in the chemical signature of compositions specific to dark and 399 pale beers could be attributed to their different origins. MRPs arise from chemical 400 reactions, which follow kinetic and thermodynamic laws, and are not influenced by 401 enzymatic catalysis. As already described in model systems (Hemmler et al., 2018), 402 Maillard derived CHNO compositions can carry multiple nitrogen atoms based on 403 404 multiple condensation reactions of amino compounds to a sugar backbone. These reactions depend on the reactivity of amino acids involved and the MR intermediate's 405 406 tendency towards carbonyl cleavage, resulting in new reducing ends of the sugar backbone. The formation of such nitrogen-rich compositions are described to 407 408 accumulate with the progress of the MR (Hemmler et al., 2017). In the complex beer system, involving numerous and interacting amino compounds, we detected 409 410 compositions with up to four nitrogen atoms (CHN₁O to CHN₄O). Interestingly, we could observe a linear decrease in the composition frequencies with increasing 411 412 nitrogen number. This agrees with the formation of nitrogen-rich compositions in the later stage of the MR and might confirm the kinetic nature of the dark beer markers. 413 The number of oxygen atoms, not in the focus of previous studies, was also found to 414 be highly systematic. With oxygen numbers exceeding 20 oxygen atoms and mass 415 values over m/z 650, both oligosaccharide precursors and condensation reactions can 416 be regarded as important factors in the formation of MRPs in beer. These high-mass 417 compounds also could be classified in the MR scheme. The evolution of the MR is 418 characterized by dehydration reactions, which are reflected in the van Krevelen 419 diagram where early MRP (1.5<H/C<2; 0.75<O/C<1) evolve to highly unsaturated and 420 aromatic compositions (H/C<1.5; O/C<0.5). The dehydration reactions inevitably come 421 with introducing a DBE to the respective target formula. Both the increasing number of 422 423 DBEs with higher mass and dehydration cascades for compositions with Kendrick nominal mass > 400 reinforce the meaning of higher mass, non-volatile MRPs in the 424 complex food system. 425

Studying exact mass differences, which represent certain compositional changes, we were able to reveal general and conceptual reaction sequences that can describe a part of the Maillard reaction in beer. Condensation reactions lead to compounds with higher mass and lead to a change in the composition, which always depends on both

the carbonyl and amino compounds. Although the condensation of glycine ($C_2H_5NO_2$) 430 and isoleucine (C₆H₁₃NO₂) with a carbonyl moiety are very similar in their underlying 431 reaction mechanism, they lead to different compositional changes (C₂H₃NO and 432 $C_{6}H_{10}NO$, respectively). The same is true for the condensation and interaction of MR 433 intermediates. Other reactions like a simple dehydration or glycation are characteristic 434 to the MR but not specific as a multitude of biochemical transformations includes a loss 435 of water or glycation as well. Accordingly, compositional changes that neither depend 436 on amino acids nor correspond to the condensation of complex intermediates or very 437 simple reactions were to be expected. 438

Therefore, the ten most significant compositional changes are changes including CHO-439 440 transformations coming with a loss of mass. Consequently, at this point, our data do not allow drawing conclusions about the role of single amino compounds but describe 441 442 the complex system holistically. What was found to be statistically significant can be referred to very general chemical changes that early MRPs or intermediates of diverse 443 444 origins undergo to build a Maillard reaction end product. By our network and shortest path approach, we furthermore were able to decipher the combination and 445 chronological order of Maillard intermediate phase reactions that match these 446 compositional changes. All intermediates were found in either beer or the Maillard 447 model system and despite of hundreds of possible combinations, the chronological 448 order was consistent within the source and target pairs. This leads us to regard the 449 450 results of the network approach as reaction sequences.

451 These sequences share a common inherent structure: Starting with the condensation of a small MR fission product, a dehydration cascade and finally a decarboxylation 452 reaction occurs. These fission products like glyoxal, methylglyoxal or diacetyl arise 453 from retro-aldolization of sugar molecules or cleavage of respective dicarbonyls 454 (Hollnagel & Kroh, 2000). Dehydration cascades are well described to play a major 455 role in the formation of MRPs. In several ribose-amino acid model systems, we were 456 able to highlight the role of early diketosamine formation and its subsequent 457 degradation in the MR (Hemmler et al., 2018). Molecular formulae equivalent to six 458 consecutive dehydration products could be described. Our presented results indicate 459 460 that such a degradation process might also be caused by the condensation of a fission product, when describing a complex system in general. In the context of the MR, loss 461 of CO₂ likely occurs due to a α -dicarbonyl assisted oxidative decarboxylation (e.g. 462

Strecker degradation) (Yaylayan, 2003). In this case, the resulting imine is hydrolyzed 463 to give the so-called Strecker aldehyde. The hydrolysis reaction leads to the loss of the 464 specific amino acid residue at the initial dicarbonyl unit. These reactions would be no 465 longer tangible for our general approach. Purely thermally induced decarboxylation 466 reactions, on the other hand, could occur during the roasting process. They require 467 very high temperatures (> 200°C) (Bagdonaite et al., 2008) and thus naturally happen 468 at the end of the heating process and reaction sequences. It is worth noting that the 469 presented pathways and their interpretation are restricted to compositional information 470 471 obtained by accurate mass measurements. They describe very general and conceptual patterns within a complex food system. Mechanistic studies including 472 473 various model systems, resolved in time, should be performed to fully understand the reaction sequences we proposed to describe the MR in beer. 474

475 In industrial practice, the extensive chemical changes that are associated with the heat load are usually monitored by the unspecific reaction of 2-thiobarbituric acid (TBA) 476 (Guillén-Sans & Guzmán-Chozas, 1998). It is based on the photometric tracking of the 477 reaction of TBA with dicarbonyl functions. However, the origin of the dicarbonyls (e.g. 478 MR or lipid oxidation) and their follow-up reactions cannot be differentiated. By 479 comparison, we recorded over 2,500 compositions that describe the MR in beer 480 comprehensively alongside the reaction network leading to such a multitude of MRPs. 481 Our analytical approach may offer a unique method to guide MR related brewing 482 processes, such as malting and boiling, towards desired attributes of the final beer end 483 product. Having the opportunity to resolve the Maillard reaction cascades and resulting 484 molecular complexity, effects of changed kilning or roasting parameters can be 485 486 monitored as well as the progress of the MR throughout the whole brewing process.

487 **5. Conclusion**

Overall, this study reports a comprehensive analytical approach addressing the great variety of MR-derived products in a complex food system, the description of their compositional nature and the general reaction cascades that lead to the diversity observed. It contributes to the better understanding of the complex molecular processes involved in the MR and might be a starting point for potential process development and quality control in both malting and brewing industry.

495 **6. Conflict of interest statement**

This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors. The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.



502 Figures

503





Figure 1. Van Krevelen diagram of molecular formula annotations found in 250 beer samples (A),
 the darkest (B) and palest (C) beer sample. Color code: CHO blue; CHNO orange; CHOS green; CHNOS red;

507 P purple. Neutral molecular formulae are plotted. The bubble size indicates the mean relative intensities of

508 corresponding peaks in the spectra.

509



Figure 2. Score plot of the PCA (A) and OPLS (B) analysis of the compositional space of 250 beer 511 512 samples and the corresponding loading plots (C-I, PCA) (C-II, OPLS). Score plot of the PCA (D) and OPLS (E) analysis of the computed mass differences in 250 beer samples and the 513 514 corresponding loading plots (F-I, PCA) (F-II, OPLS). The position of the beer samples is marked by dots colored according to their absorption at 294 nm. The prediction of the Maillard model system in the 515 516 OPLS models (B and E) is highlighted as a red star. Masses in the PCA-loading plot (C-I and F-I) that 517 match the most significant masses for dark beers in the OPLS-loading plot (C-II and F-II) are colored 518 brown.



520

Figure 3. Comparison of dark (I) and pale (II) beer marker molecular formulae by different 521 522 visualizing plots (A-F). Number of annotations in the chemical spaces (A), number of nitrogen atoms 523 (B), number of oxygen atoms (C), Van Krevelen diagram (D), Double bond equivalents against Number 524 of Carbon atoms (E) and Kendrick mass defect plot with H₂O homologous series (F). Color code: CHO 525 blue; CHNO orange; CHOS green; CHNOS red; P purple. Neutral molecular formulae are plotted. The 526 bubble size indicates the mean relative intensities of corresponding peaks in the spectra (D, E). Rising 527 DBE with higher masses for dark markers is indicated in (E-I). Homologous series of H₂O-reactions are 528 marked exemplary in the KMD plot (F-I). The intrinsic systematic pattern of dark beer markers is opposed 529 to non-systematic annotations of the pale marker masses.



532 Figure 4. Van Krevelen diagrams of compositions connected by the ten most significant mass 533 differences for dark beers (A and B) and their breakdown into small reaction series by a mass difference network (C). Higher mass values (A) and lower mass values (B) of the mass pairs. The 534 entirety of compositions is in the background in gray. The lower left position of low m/z values indicate 535 536 degradation reaction sequences. Nodes in the mass difference network (C) represent all annotated 537 compositions connected by edges representing small Maillard intermediate phase reactions 538 (Supplementary Table S6). Sources and targets of the statistically most significant big composite mass 539 differences are colored.

	I					Dehy	/dratio	on ca	scade	9		
C ₁ H ₋₁₂ O ₋₈	ب	G	DH	DH	DH	DH	DH	DH	DH	DH	DC	
$C_1 H_{-12} O_{-7}$	quo	G		DH	DH	DH	DH	DH	DH	DH	DC	
$C_1 H_{-10} O_{-7}$	oud	AcA		DH	DH	DH	DH	DH	DH	DH	DC	De
$C_2H_{-10}O_{-7}$	ion	MeG		DH	DH	DH	DH	DH	DH	DH	DC	Carl
C ₂ H ₋₁₀ O ₋₆	LISS	MeG		DHy	DH	DH	DH	DH	DH	DH	DC	b X
C ₃ H ₋₈ O ₋₆		DA		DHy	DH	DH	DH	DH	DH	DH	DC	ylati
$C_1H_{\text{-}14}O_{\text{-}8}$				DHy	DH	DH	DH	DH	DH	DH	DC	n
H ₋₁₂ O ₋₈	Hy	Hy	DH	DH	DH	DH	DH	DH	DH	DH		
H ₋₁₄ O ₋₈		Hy	DH	DH	DH	DH	DH	DH	DH	DH		

DH: DehydrationDHy: DehydrogenationG: GlyoxalAcA: AcetaldehydeDC: DecarboxylationHy: HydrogenationMeG: MethylglyoxalDA: Diacetyl

541 Figure 5. Reaction sequences of the ten most significant compositional changes during the MR

in beer. All reaction sequences feature a dehydration cascade. In many cases, MR fission products start
 the reaction sequence, which ends with a decarboxylation reaction. The compositional change of
 hydrogenation (+H₂) does not indicate the involvement of elemental hydrogen, but a
 reductone/dehydroreductone reactional environment.

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547 Tables

loading	∆ m/z	formula	frequency	Decomposition into individual MDs	% of shortest
					paths
0.01581	-128.053	C1H-12O-8	543	Dehydration (8), Glyoxal, Decarboxylation	64
0.01576	-140.053	H-12O-8	714	Dehydration (8), Hydrogenation (2)	65
0.01576	-142.069	H-14O-8	576	Dehydration (8), Hydrogenation	82
0.01571	-98.0427	C ₂ H-10O-7	685	Dehydration (7), Methylglyoxal	75
0.01566	-110.043	C1H-10O-7	879	Dehydration (7), Acetaldehyde, Decarboxylation	65
0.01565	-112.058	C1H-12O-7	722	Dehydration (7), Glyoxal, Decarboxylation	66
0.01551	-82.0477	$C_2H_{-10}O_{-6}$	890	Dehydration (6), Methylglyoxal, Dehydrogenation, Decarboxylation	83
0.01549	-68.0321	C ₃ H ₋₈ O ₋₆	830	Dehydration (6), Diacetyl, Dehydrogenation, Decarboxylation	94
0.01549	-154.069	$C_{-1}H_{-14}O_{-8}$	746	Dehydration (6), Dehydrogenation, Decarboxylation	90
0.01543	-93.0637	C1H-11N1O-5	861	-	-

549Table 1. The ten most significant compositional changes during the MR in beer and their break550down into small reactions.

553 554 555	Almeida, C., Duarte, I. F., Barros, A., Rodrigues, J. E., Spraul, M., & Gil, A. M. (2006). Composition of beer by 1H NMR spectroscopy: effects of brewing site and date of production. <i>J. Agric. Food</i> <i>Chem., 54</i> , 700-706.
556 557 558	Bagdonaite, K., Derler, K., & Murkovic, M. (2008). Determination of acrylamide during roasting of coffee. <i>J. Agric. Food Chem., 56</i> , 6081-6086.
559 560 561	Bastian, M., Heymann, S., & Jacomy, M. (2009). <i>Gephie: an open source software for exploring and manipulating networks</i> International AAAI Conference on Weblogs and Social Media,
562 563 564 565	Capuano, E., & Fogliano, V. (2011). Acrylamide and 5-hydroxymethylfurfural (HMF): A review on metabolism, toxicity, occurrence in food and mitigation strategies. <i>LWT-Food Sci. Technol.,</i> 44, 793-810.
566 567 568 569	Dack, R. E., Black, G. W., Koutsidis, G., & Usher, S. J. (2017). The effect of Maillard reaction products and yeast strain on the synthesis of key higher alcohols and esters in beer fermentations. <i>Food Chem., 232</i> , 595-601.
570 571 572 573 574	 Dietrich, L., Götting-Martin, E., Hertzog, J., Schmitt-Kopplin, P., McGovern, P. E., Hall, G. R., Petersen, W. C., Zarnkow, M., Hutzler, M., Jacob, F., Ullman, C., Notroff, J., Ulbrich, M., Flöter, E., Heeb, J., Meister, J., & Dietrich, O. (2020). Investigating the function of Pre-Pottery Neolithic stone troughs from Göbekli Tepe – An integrated approach. J. Archaeol. Sci. Rep, 34, 1-20.
575 576 577 578	Duarte, I. F., Barros, A., Almeida, C., Spraul, M., & Gil, A. M. (2004). Multivariate Analysis of NMR and FTIR Data as a Potential Tool for the Quality Control of Beer. <i>J. Agric. Food Chem., 52</i> , 1031-1038.
579 580 581	Eriksson, L., Trygg, J., & Wold, S. (2008). CV-ANOVA for significance testing of PLS and OPLS models. J. Chemometrics, 22, 594-600.
582 583 584	Friedman, M., & Molnar-Perl, I. (1990). Inhibition of browning by sulfur amino acids. 1. Heated amino acid-glucose systems. J. Agric. Food Chem., 38(8), 1642-1167.
585 586	Golbraikh, A., & Tropsha, A. (2002). Beware of q2! J. Mol. Graph. Model., 20, 269-276.
587 588 589 590 591	Gougeon, R. D., Lucio, M., Frommberger, M., Peyron, D., Chassagne, D., Alexandre, H., Feuillat, F., Voilley, A., Cayot, P., Gebefügi, I., Hertkorn, N., & Schmitt-Kopplin, P. (2009). The chemical diversity of wines can reveal a metabologeography expression of cooperage oak wood. <i>PNAS</i> , 106, 9174-9179.
592 593 594	Guillén-Sans, R., & Guzmán-Chozas, M. (1998). The Thiobarbituric Acid (TBA) Reaction in Foods: A Review. <i>Crit. Rev. Food Sci. Nutr., 38</i> (4), 315-330.
595	

596 597 598	Hagberg, A. A., Schult, D. A., & Swart, P. J. (2008). Exploring network structure, dynamics, and function using networkx. Proceedings of the 7th Python in Science Conference (SciPy2008), Pasadena, CA.
599 600 601 602	Haseleu, G., Lagemann, A., Stephan, A., Intelmann, D., Dunkel, A., & Hofmann, T. (2010). Quantitative sensomics profiling of hop-derived bitter compounds throughout a full-scale beer manufacturing process. <i>J. Agric. Food Chem., 58</i> (13), 7930-7939.
603 604 605	Hellwig, M., & Henle, T. (2010). Formyline, a new glycation compounds from the reaction of lysine and 3-deoxypentosone. <i>Eur. Food Res. Technol., 230</i> (6), 903-914.
606 607 608 609	Hellwig, M., Witte, S., & Henle, T. (2016). Free and protein-bound Maillard reaction products in beer: method development and a survey of different beer types. J. Agric. Food Chem., 64, 7234- 7243.
610 611 612 613	 Hemmler, D., Gonsior, M., Powers, L. C., Marshall, J. W., Rychlik, M., Taylor, A. J., & Schmitt-Kopplin, P. (2019). Simulated sunlight selectively modifies Maillard reaction products in a wide array of chemical reactions. <i>Chem. Eur. J.</i>
614 615 616 617	Hemmler, D., Roullier-Gall, C., Marshall, J. W., Rychlik, M., Taylor, A. J., & Schmitt-Kopplin, P. (2018). Insights into the Chemistry of Non-Enzymatic Browning Reactions in Different Ribose-Amino Acid Model Systems. <i>Sci. Rep., 8</i> , 1-9.
618 619 620 621	Hemmler, D., Roullier-Gall, C., Marshall, J. W., Rychlik, M., Taylor, A. J. T., & Schmitt-Kopplin, P. (2017). Evolution of complex Maillard chemical reactions, resolved in time. <i>Sci. Rep., 7</i> (1), 3227-3233.
622 623 624 625	Hertkorn, N., Frommberger, M., Witt, M., Koch, B. P., Schmitt-Kopplin, P., & Perdue, E. M. (2008). Natural organic matter and the event horizon of mass spectrometry. <i>Anal. Chem., 80,</i> 8908- 8919.
626 627 628	Hodge, J. E. (1953). Dehydrates Foods, Chemistry of Browning Reactions in Model Systems. <i>J. Agric.</i> Food Chem., 1, 928-943.
629 630 631 632	Hollnagel, A., & Kroh, L. W. (2000). Degradation of oligosaccharides in nonenzymatic browning by formation of a- dicarbonyl compounds via a "peeling off" mechanism. <i>J. Agric. Food Chem., 48</i> , 6219-6226.
633 634 635 636	Kim, S., Kramer, R. W., & Hatcher, P. G. (2003). Graphical Method for Analysis of Ultrahigh-Resolution Broadband Mass Spectra of Natural Organic Matter, the Van Krevelen Diagram. Anal. Chem., 75, 5336-5344.
637 638 639	Kuntcheva, M. T., & Obretenov, T. D. (1996). Isolation and characterization of melanoidins from beer. Z. Lebensm. Unters. Forsch., 202(3), 238-243.
640	

641 642 643	Lachenmeier, D. W., Frank, W., Humpfer, E., Schäfer, H., Keller, S., Mortter, M., & Spraul, M. (2005). Quality control of beer using high-resolution nuclear magnetic resonance spectroscopy and multivariate analysis. <i>Eur. Food Res. Technol.</i> , 220, 215-221.
644 645 646 647 648	Lucio, M., Fekete, A., Frommberger, M., & Schmitt-Kopplin, P. (2001). Metabolomics: High-Resolution Tools Offer to Follow Bacterial Growth on a Molecular Level. In F. J. de Bruijn (Ed.), <i>Handbook</i> of Molecular Microbial Ecology I: Metagenomics and Complementary Approaches (pp. 683- 695). John Wiley & Sons.
649 650 651	Lusk, L. T., Goldstein, H., & Ryder, D. (1995). Independent role of beer proteins, melanoidins and polysaccharides in foam formation. <i>J. Am. Soc. Brew. Chem., 53</i> (3), 93-103.
652 653 654 655	Mavric, E., & Henle, T. (2006). Isolation and identification of 3,4-dideoxypentosulose as specific degradation product of oligosaccharides with 1,4-glycosidic linkages. <i>Eur. Food Res. Technol., 223</i> , 803-810.
656 657 658	Meußdoerffer, F., & Zarnkow, M. (2014). <i>Das Bier. eine Geschichte von Hopfen und Malz</i> (Vol. 2nd Ed.). Beck.
659 660 661	Michel, R. H., & McGovern, P. E. (1993). The first wine & beer. Chemical detection of ancient fermented beverages. <i>Anal. Chem., 65</i> (8), 408-413.
662 663 664 665	Obretenov, T. D., Demyttenaere, J., Abbaspour-Tehrani, K., Adams, A., Kersiene, M., & De Kimpe, N. (2002). Flavor realease in the presence of melanoidins prepared from L-(+)-ascorbic acid and amino acids. J. Agric. Food Chem., 50, 4244-4250.
666 667 668	Otter, G. E., & Taylor, L. (1976). The determination of amino acids in wort, beer and brewing materials using gas chromatography. <i>J. Inst. Brew., 82</i> , 264-269.
669 670 671	Pieczonka, S. A., Lucio, M., Rychlik, M., & Schmitt-Kopplin, P. (2020). Decomposing the molecular complexity of brewing. <i>NPJ Sci. Food, 4</i> (11), 1-10.
672 673 674	Pieczonka, S. A., Rychlik, M., & Schmitt-Kopplin, P. (2021). Metabolomics in Brewing Research. In A. Cifuentes (Ed.), <i>Comprehensive Foodomics</i> (Vol. 2, pp. 116-128). Elsevier.
675 676 677 678	Rodrigues, J. A., Barros, A. S., Carvalho, B., Brandão, T., & Gil, A. M. (2011). Probing beer aging chemistry by nuclear magnetic resonance and multivariate analysis. <i>Analytica Chimica Acta, 702</i> , 178-187.
679 680 681	Roullier-Gall, C., Witting, M., Gougeon, R. D., & Schmitt-Kopplin, P. (2014). High precision mass measurements for wine metabolomics. <i>Front. Chem.,</i> 2(102), 1-9.
682 683 684 685	Schmitt-Kopplin, P., Hemmler, D., Moritz, F., Gougeon, R. D., Lucio, M., Meringer, M., Müller, C., Harir, M., & Hertkorn, N. (2019). Systems chemical analytics: introduction to the challenges of chemical complexity analysis. <i>Faraday Discuss., 218</i> , 9-28.

686 687 688 689	Smirnov, K. S., Forcisi, S., Moritz, F., Lucio, M., & Schmitt-Kopplin, P. (2019). Mass difference maps and their application for the recalibration of mass spectrometric data in nontargeted metabolomics. <i>Anal. Chem., 91</i> , 3350-3358.
690 691 692	Spreng, S., & Hofmann, T. (2018). Activity-guided identification of <i>in vitro</i> Antioxidants in beer. <i>J.</i> Agric. Food Chem., 66, 720-731.
693 694 695 696 697	Tziotis, D., Hertkorn, N., & Schmitt-Kopplin, P. (2011). Kendrick-analogous network visualization of ion cyclotron resonance fourier transform mass spectra: improved options for the assignment of elemental compositions and the classification of organic molecular complexity. <i>Eur. J. Mass Spectrom., 17</i> , 415-421.
698 699 700 701	Westerhuis, J. A., Hoefsloot, H. C. J., Smit, S., Vis, D. J., Smilde, A. K., van Velzen, E. J. J., van Duijnhoven, J. P. M., & van Dorsten, F. A. (2008). Assessment of PLSDA cross validation. <i>Metabolomics, 4</i> , 81-89.
702 703 704	Yaylayan, V. A. (2003). Recent adcances in the chemistry of Strecker degradation and Amadori rearrangement: implications to aroma and color formation. <i>Food Sci. Technol., 9</i> (1), 1-6.
705 706 707	Yaylayan, V. A., & Mandeville, S. (1994). Stereochemical control of maltol formation in Maillard reaction. <i>J. Agric. Food Chem., 42</i> , 771-775.
708 709 710 711	 Yu, X., Zhao, M., Hu, J., Zeng, S., & Bai, X. (2012). Correspondence analysis of antioxidant activity and UV-Vis absorbance of Maillard reaction products as related to reactants. <i>Food Sci. Technol.</i>, 46, 1-9.
712	
713	