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1 Enantioseparation and selective detection of D amino acids by
2 ultra-high-performance liquid chromatography/ mass
3 spectrometry in analysis of complex biological samples

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5
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28 **Abstract**

29

30 The growing scientific attention in the biological function of D-amino acids leads to an
31 increasing analytical interest for enantiomeric amino acid separation, which is still very
32 challenging due to the lack of sufficiently sensitive, high-throughput analytical methods that
33 can cope with often occurring matrix interferences and very low D-amino acid concentrations.
34 Here, enantioseparation can benefit from improved resolution and chromatographic speed
35 offered by modern UHPLC techniques and the precision of MS detection. We developed a
36 RP-UHPLC-QqToF-MS method using pre-column OPA/IBLC derivatization for very precise
37 discrimination of amino acids enantiomers. The method shows a superb sensitivity with limits
38 of detection in the range of several pmol/l. It has neither shown matrix interferences in the tested
39 very complex biological matrices (serum, plasma, urine and gut) nor stability or racemisation
40 problems.

41 Keywords: chiral derivatization, chiral separation, liquid chromatography, mass spectrometry,
42 D-amino acids, complex matrices, OPA-IBLC

43

44

44 1. Introduction

45

46 Amino acids (AA) are very important chiral biomolecules. They participate in gene
47 expression, regulation of metabolism, cell signaling and immunity [1,2]. The predominantly
48 present L-enantiomers function as building blocks of peptides and proteins or precursors for
49 the synthesis of several important molecules. D-enantiomers are comparatively very low
50 abundant. While even been believed to be absent in higher animals in the past [3,4], now D-
51 AA are the focus of several studies as they are considered to be new bioactive compounds and
52 biomarkers [5]. An accumulation of D-isoleucine, D-valine, D-leucine, and D-methionine in
53 culture media was reported to stimulate the conversion of rod-shaped into spherical bacteria,
54 which illustrates how D-AA can influence peptidoglycan structure and composition [6]. D-
55 tryptophan, D-leucine, D-methionine, and D-tyrosine can inhibit biofilm formation or even
56 disrupt existing ones [7]. Additionally, D-AAs are used in nutritional industry as markers for
57 contamination from microorganism or for food age because D-AAs, like D-alanine or D-
58 glutamic acid, are natural components of bacterial cell walls [8]. Moreover, the isomerization
59 of L to less digestible D-enantiomers leads to a reduced dietary value in processed food
60 products [9]. Main natural sources of D-AAs are next to microorganisms soils, seeds, fruits
61 and tree leaves of several plants [10], but also meteorites have been extensively studied for
62 D/L-AA ratio. This ratio is used to confirm an abiotic origin or possible terrestrial
63 contamination of AAs in the analyzed material [11,12].

64 To further facilitate research on the presence and biological function of D-AA, sensitive and
65 selective methods need to be developed. The aim of this work was to establish a UHPLC-
66 QqToF-MS (ultra-high performance liquid chromatography – time of flight mass
67 spectrometry) method for AA enantiomeric analysis, which combines the enhanced separation

68 possibilities of sub-2 μm particles of UHPLCs with the sensitivity and accuracy of mass
69 spectrometric detection [12–14], and that is directly applicable to biological samples. A pre-
70 column derivatization using o-phthalaldehyde (OPA) in combination with the chiral thiol
71 isobuteryl-L-cysteine (IBLC) , a well-established tagging reagent (Figure 1), has been
72 preferred and applied in UHPLC-FLD (fluorescence detection) and UHPLC-QqToF-MS. The
73 challenges of fluorescence detection will be discussed as well as the need of detection method
74 transfer from FLD to the more selective and therefore sensitive MS. The developed method
75 and putative matrix interferences were tested analyzing four very complex biological
76 matrices. Stability issues of AA derivatives were addressed as well.

77

78 2. Materials and methods

79

80 2.1. Chemicals and reagents

81

82 Methanol (Chromasolv, Sigma-Aldrich, St. Lois, USA), acetonitrile (Chromasolv, Sigma-
83 Aldrich) and water (MilliQ) were used in LC-MS quality. OPA, IBLC, IBDC, boric acid and
84 sodium acetate were purchased from Sigma Aldrich, as well as all used amino acid standards
85 (enantiomeric mixtures: D/L-Ala, D/L-Asn, D/L-Asp, D/L-Cys, D/L-Glu, D/L-His, D/L-Ile,
86 D/L-Leu, D/L-Met, D/L-Pro, D/L-Ser, D/L-Val; enantiomer pure standards: L-Arg, L-Gln, L-
87 Lys, L-Phe, L-Thr, L-Trp, L-Tyr, D-Arg, D-Gln, D-Lys, D-Phe, D-Thr, D-Trp, D-Tyr).
88 Compounds are indicated according to IUPAC symbolism for amino acids. Ammonium
89 acetate buffer was bought from Biosolve (Valkerswaard, Netherlands) and an electrospray
90 ionization (ESI) tuning mix from Agilent (Santa Clara, USA) for MS calibration.

91

92 2.2. Preparation of human serum, human plasma and urine

93

94 The developed UHPLC-FLD and UHPLC-QqToF-MS protocols were tested with human
95 serum (HuS), EDTA-plasma and urine for general suitability and putative matrix
96 interferences. Therefore, we took samples from several individual volunteers and pooled these
97 aliquots. The samples were stored at -80°C directly after sampling until analysis. The reason
98 to test the method with such a mixture of samples taken from many individuals was the
99 increased complexity of pooled samples. We wanted to capture as many possible interfering
100 compounds as possible (e.g. different diets of different persons and therefore different
101 putative interfering compounds). We do not intent to answer any biological question with
102 these experiments..

103 HuS and EDTA-plasma required furthermore protein precipitation prior analysis. After
104 careful thawing on ice, 20 µl of each sample matrix was mixed with 80 µl ice cold methanol
105 and centrifuged (15000rpm, 4°C, 15min). The supernatant was taken, evaporated and resolved
106 in water before injection. Urine was thawed on ice as well and centrifuged to avoid particle
107 injection (15000 rpm, 4°C, 15 min).

108

109 2.3. Mouse gut

110

111 About 100 mg of frozen cecal content of ten week old C57BL/6J mouse was protein-
112 precipitated by using 500 µl cold methanol (-20°C), whereas ceramic beads were used for cell
113 disruption (NucleoSpin bead Tubes, Macherey-Nagel GmbH & Co. KG). Subsequently, the
114 sample was homogenized using TissueLyser II (Qiagen) for 5 minutes at a rate of 30 Hertz.
115 This procedure was repeated three times in order to increase the extraction efficiency.
116 Afterwards, the sample was centrifuged at 14.000 rpm for 10 min at 4°C and the supernatants

5

117 were pooled for the analyses. The supernatants were kept in -20°C conditions prior to
118 experiments.

119

120 2.4. Derivatization

121

122 The derivatization reagent OPA/IBLC was freshly prepared based on a previously described
123 HPLC-FLD method [15]. A methanolic solution of OPA (200 mM) was mixed with 200 mM
124 IBLC prepared in 0.1 M borate buffer corresponding to the moles ratio OPA/IBLC 1/3, which
125 is optimal for the stability of AA derivatives [16]. The pH of the mixture was adjusted to 9
126 with 1M sodium hydroxide. Derivatization of amino acid in standards and samples was
127 carefully carried out for 2 minutes under a fume hood by mixing 10µl of standards/samples
128 with 20µl derivatization reagents prior injection. This corresponds to an excess of OPA
129 molecules of approximately 1×10^8 (for the lowest tested amino acid concentration) and to
130 10 (for the highest tested amino acid concentration) compared to amino acids molecules in
131 solution (0.0001 - 1.5 ppm). All glassware were heat sterilized before usage [12]. Blank
132 samples consisting of pure solvent were included in all analysis.

133

134 2.5. UHPLC-FLD

135

136 Enantiomeric separation was initially achieved using a ACQUITY UPLC® system (Waters,
137 Milford, USA) coupled to a fluorescence detector, which has been set to $\lambda=300$ nm for the
138 excitation and $\lambda=445$ nm for the emission. A BEH-C18 column with dimensions of 2.1 x 150
139 mm and 1.7 µm particle size (Waters) was applied with a column temperature of 30 °C. The
140 auto-sampler was set to 25 °C. Mobile phase A consisted of 20 mM sodium acetate buffer that

141 was adjusted to pH 6.2 with acetic acid. Mobile phase B was composed of 7% acetonitrile in
142 methanol [15]. The gradient was optimized with a flow rate of 0.35 ml/min (Table 1).

143

144 2.6. UHPLC-QqToF-MS

145

146 The ACQUITY UPLC® was coupled to a QqToF-MS (maXis, Bruker Daltonics, Bremen,
147 Germany) to allow mass spectrometric detection. The maXis mass spectrometer is a hybrid
148 QqToF, dual stage reflector instrument, which uses orthogonal ion acceleration. “Q” (capital
149 letter) refers to a mass selective mass resolving quadrupole, whereas “q” indicates a
150 quadrupole collision cell. It is therefore possible to isolated selected masses in the mass
151 isolating quadrupole and fragment these isolated masses of interest in the second quadrupole.
152 The mass fragments can thereafter be measured according to their time of flight in the flight
153 tube. Slight modifications of the UHPLC-FLD method were necessary to ensure compatibility
154 with electrospray ionization. Sodium acetate was exchanged with ammonium acetate (2.5
155 mM). The flow rate was reduced to 0.2 ml/min for the same reason. Additional parameters
156 were kept constant. Mass spectrometric parameters were optimized in order to achieve highest
157 sensitivity in the mass range of the amino acid derivatives (350-500 Da) within an acquisition
158 window of 50-1500 Da and a scan rate of 2 Hertz.

159

160 3. Results

161

162 3.1. UHPLC-FLD matrix interferences followed by method transfer

163 Baseline separation of enantiomeric amino acid standards could be achieved within 34
164 minutes using the sensitive UHPLC-FLD method (Supplementary Figure 1). Nevertheless,

165 separation of standard materials is so far easily achievable, but in daily routine the analyst has
166 to deal with very complex samples. FLD detection turned out to be not well suited for such
167 studies as e.g. strong interferences from derivatization reagents and samples themselves have
168 been recognized (Supplementary figure 2). Since besides primary amino acids, amino esters,
169 amino alcohols, alkyl- and aryl-amines, and heterocyclic amines react with OPA [17], a
170 higher baseline and co-elution for some D-AA has been observed. This increases the limit of
171 detection and might lead in miss-interpretation if identification solely relies on retention time
172 comparison with standard compounds. Extensive sample clean-up might overcome this
173 problem, but requires time and bargains the risk of contaminations and sample alterations.
174 Thus, analytical methods insensitive for such interferences are on demand and consequently
175 we transferred the method to UHPLC-QqToF-MS. The flow rate and buffer of mobile phase
176 A were therefore adjusted. Figure 2 shows the extracted ion chromatogram (EIC) obtained for
177 a derivatized AA standard mixture. Proline as a secondary amine does not react with
178 OPA/IBLC reagent, glycine is achiral and was consequently not included in our study, and
179 cysteine could not be detected. Additionally, the early elution of aspartic acid (t_R 4.1 min)
180 resulted in poor resolution between its D and L forms showing a single peak. With these
181 exceptions, baseline separation was achieved for all proteinogenic AA enantiomeric pairs.
182 The high resolution ToF-MS allowed us to targeted extract m/z values of each AA derivative
183 with accuracy of $m/z \pm 0.001$ (except Asp 0.005, Glu 0.005, Phe 0.002 and Val 0.005), which
184 provided clean chromatograms.

185

186 3.2. LOD and linearity in UHPLC-QqToF-MS

187

188 Amino acid standards of 0.5 - 1 ppm concentration were used to determine the limits of
189 detection, which was calculated as $LOD = \text{conc. AA [mmol/l]} \times 3 / \text{SNR}$ for pure standards

8

190 solutions and as $LOD = \text{basic AA conc} + \text{spiked conc. AA [mmol/l]} \times 3 / \text{SNR at spiked}$
191 concentration for biological matrices. Signal to noise ratio (SNR) was elaborated with Bruker
192 Daltonics Data Analysis 4.2. It needs to be considered that the LOD may vary when analyzing
193 different sample matrices. Here, we present data for pure standards (Table 2) and spiked
194 plasma samples (Table 3). The signal-concentration response followed a linear relationship
195 for both standard solution (data not shown) and plasma in the concentration range of 0.0001
196 ppm to 1.5 ppm (Table 3). Depending on the matrix and AA of interest it might be necessary
197 to dilute the matrix before spiking the calibration curve due to the high basic concentration.

198

199 3.3. Stability of derivatization product

200

201 An additional crucial point is the stability of the derivatization product. Derivatization is
202 completed after 2 min at room temperature and has been known to be stable for at least 18
203 min [15]. We tested the peak area response of the solutions of these standards after 2, 20, 40
204 and 60 min and observed stable signals of derivatization products for at least 40 min (97.5 -
205 101.8 % of original detected value). Thereafter, the signal response decreased (Figure 3).
206 Thus, we strongly suggest the derivatization directly prior injection. A lab robot might be
207 useful for automation of this procedure. Putative racemization of amino acids during
208 derivatization and chromatography can be excluded because we confirmed the absence (or
209 very low abundance as indicated by the enantiomeric purity of the vender) of the other
210 enantiomer for several amino acids in standard mixtures and in biological samples of various
211 concentrations (exemplarily illustrated for L-Arg, L-Asp and L-Trp in Supplementary
212 Figure 3).

213

214 3.4. Applicability to important biological matrices

215 The developed UHPLC-QqToF-MS method was finally tested in four different biological
216 matrices to evaluate possible signal interferences caused by the sample matrix itself. Due to
217 the application of QqToF-MS, specific detection of proteinogenic amino acids could be
218 achieved without matrix-interferences signals. Chromatograms for alanine, serine and
219 methionine for standards and tested biological matrices are exemplarily illustrated in Figure 4
220 as their derivatization and ionization behavior can be used as approximation of the other
221 amino acids [13]. The detected exact mass of AA derivatives and retention time allow certain
222 identification. We observed very small retention time shifts when injecting different matrices
223 (Figure 4), e.g. retention time shifts for D-AA standards in water and plasma were between
224 0.1 and 0.3 min, which shows a high reproducibility. Additionally, MS² fragmentation
225 information can be easily obtained by various MS instrument if uncertainties in the identity of
226 any AA exist.

227 Due to very presumable different matrix effects of each sample type in electrospray
228 ionization, we relinquish to give absolute concentrations and strongly suggest determining
229 crucial quantification parameters for each specific application. The biological matrices were
230 used here with the purpose to test the analytical protocol for putative inferences or any other
231 analytical problem. Consequently, we repetitively analyzed pooled samples of the selected
232 matrices. We did not intent to answer any biological question.

233

234 4. Discussion

235

236 The increasing scientific interest in D-AA requires more and more sensitive and selective
237 analytical protocols. Due to several facts, like the possibility for reliable chiral separation,
238 coverage of a very high number of analytes and suitability for full automation, liquid

239 chromatography (LC) has been applied for chiral separation [4,15,18]. Existing HPLC-FLD
240 protocols are particularly challenged by matrix interferences, the very low abundances of D-
241 AAs and huge concentration differences compared to their L-enantiomers. UHPLC, which
242 offers a much higher separation efficiency and chromatographic speed due to the application
243 of sub-2 μm porous particles [19, 20], in combination with the precise and sensitive detection
244 of mass spectrometry is one way to improve enantioseparation and reach the analytical needs
245 currently occurring.

246 Two strategies can be followed to resolve enantiomers in LC systems, first the application of
247 chiral stationary phase and secondly enantioseparation based on a pre-column derivatization.
248 Several facts led us to use pre-column derivatization for the development of our UHPLC
249 method. The most important ones are a higher efficiency, which can be reached using chiral
250 derivatization reagents followed by reversed phase (RP) chromatography [4], the missing
251 availability of chiral material with sub-2 μm particles for UHPLCs and a better sensitivity that
252 can be reached with RP columns due to the high number of analytical plates. Commonly, pre-
253 column derivatization uses fluorescence tagging for detection [15,21–23]. Unfortunately,
254 although fluorescence detection is one trustworthy detection method, complex biological
255 samples contain endogenous compounds that are likely to cause signal interference or may
256 co-elute with some D-AA, which is a source of variability and inaccuracy. Mostly primary
257 amino acids, amino esters, amino alcohols, alkyl- and aryl-amines and heterocyclic amines
258 react with OPA [17]. Samples need consequently either an intensive clean-up or more precise
259 detection methods are necessary. FLD application have been therefore limited to few amino
260 acids [13]. We also observed such problems when we applied our developed UHPLC-FLD
261 method to differently composed biological matrices and decided thus to transfer the method to
262 UHPLC-QqToF-MS, which allows to isolate the exact masses of the derivatization products
263 of each amino acid and is in consequence insensitive for most matrix interferences signals.

264 This leads eventually to a method that is much more extensive in its application. This
265 potential of UHPLC applications in combination with pre-column derivatization for chiral
266 separation of amino acids has been hardly described. Only three references could be found.
267 Zhe Min developed a new fluorescence tagging reagent (DBD-PyPNCS) for the analysis of
268 amino acids in human nails [14]. Visser compared NDB-PyNCS, GITC, AMBI, OPA/IBLC,
269 S-NIFE, Marfeys- and Sangers reagent [13]. In their publication it was shown that NIFE
270 delivered best sensitivity for alanine, serine and methionine followed by OPA/IBLC [13].
271 However, we decided to proceed with OPA/IBLC due to availability issues and achieved a 10
272 times lower LOD for OPA/IBLC compared to the published method. One further argument to
273 proceed with OPA/IBLC is the possibility to exchange the elution order of D and L
274 enantiomers simply by using IBDC instead of IBLC, which allows the verification of the
275 abundance of the D-AA and might be helpful in some rare cases in which a co-eluting peak
276 has the same mass as the AA derivative. Additionally, the exchange is very advantageous if
277 the L-form is present in much higher concentration than the D-AA, which might be the case
278 of bacterial cell cultures or chemically defined medium.. The development of the presented
279 method was motivated by the fact that traditional fluorescence detection alone cannot offer
280 the level of certainty necessary to analyze D-AA in biological samples. Although there is still
281 some place for improvement, our MS detection allows sensitive, specific targeted and
282 relatively fast differentiation of proteinogenic AA enantiomers in complex matrices.
283 Analytical performance should be always evaluated for any specific application and sample
284 matrix due possible interference during ESI.

285

286 5. Conclusions

287

288 A simple applicable, specific and sensitive UHPLC-MS method for differentiation of AA
289 using chiral derivatization with OPA/IBLC reagent and RP chromatography has been
290 presented. The widely implemented stationary phase C18 prevents the analyst of buying
291 special and more expensive chiral columns. It offers moreover a high number of analytical
292 plates and consequently very good separation efficiency. The exchangeability of the elution
293 order of L- and D- enantiomers simply by usage of IBDC instead of IBLC might be very
294 useful for some applications. Most importantly, the very sensitive and selective MS detection
295 overcomes misinterpretation of detected peaks, which can result from interference signals in
296 complex samples when methods solely rely on fluorescence detection.

297

298 6. Acknowledgment

299

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301 Forcisi for providing test samples.

302

303 7. Figure caption

304

305 **Figure 1:** Derivatization reaction between OPA, primary amino acid and IBLC, which
306 contains the thiol group for enantioseparation of DL-amino acids and fluorescence detection.
307 Besides primary amino acids, their esters, amino alcohols, alkyl- and aryl-amines,
308 heterocyclic amines react with OPA resulting in strong interferences for complex samples.
309 The condensation between the amino group and the aromatic o-dicarboxaldehyde leads in an
310 N-substituted isoindolin-1-one (phthalimidine) derivative [17].

311

312 **Figure 2:** UHPLC-MS extract ion chromatogram (EIC) obtained for the reversed phase
313 enantioseparation of OPA/IBLC derivatized amino acid standards at concentration of 0.5 and
314 1 ppm (Thr, Asp, Lys, Phe, Arg, Glu, Tyr, and Trp)

315

316 **Figure 3:** Stability of detected peak area after t_0+x min

317 **Figure 4:** Extracted Ion Chromatograms (± 0.01) of alanine, serine and methionine for
318 standards, mouse gut, HuS, plasma and urine. Retention time windows of D-AA are marked
319 in grey and enlarged illustrated, if the D-enantiomer has been detected.
320
321

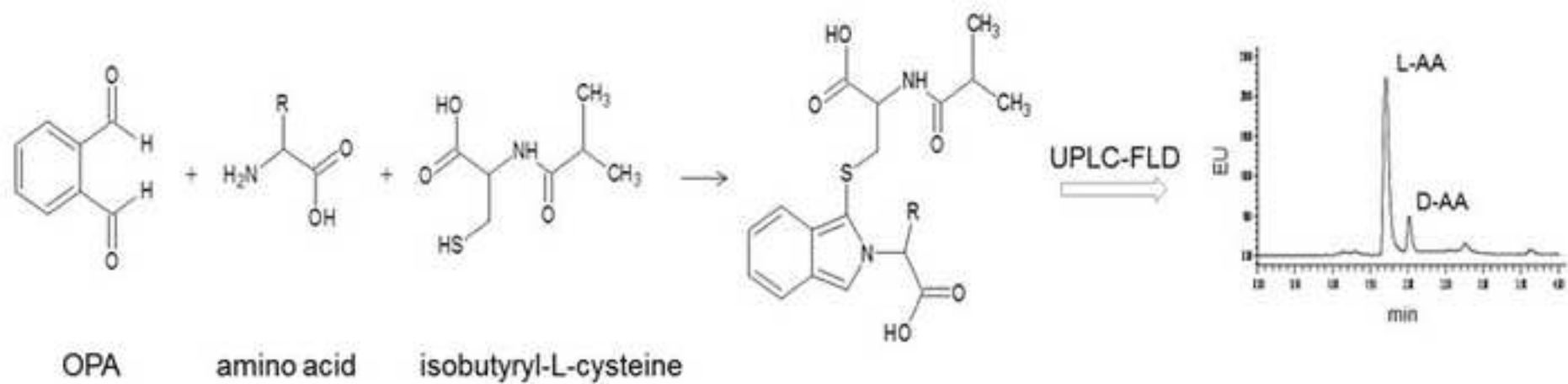
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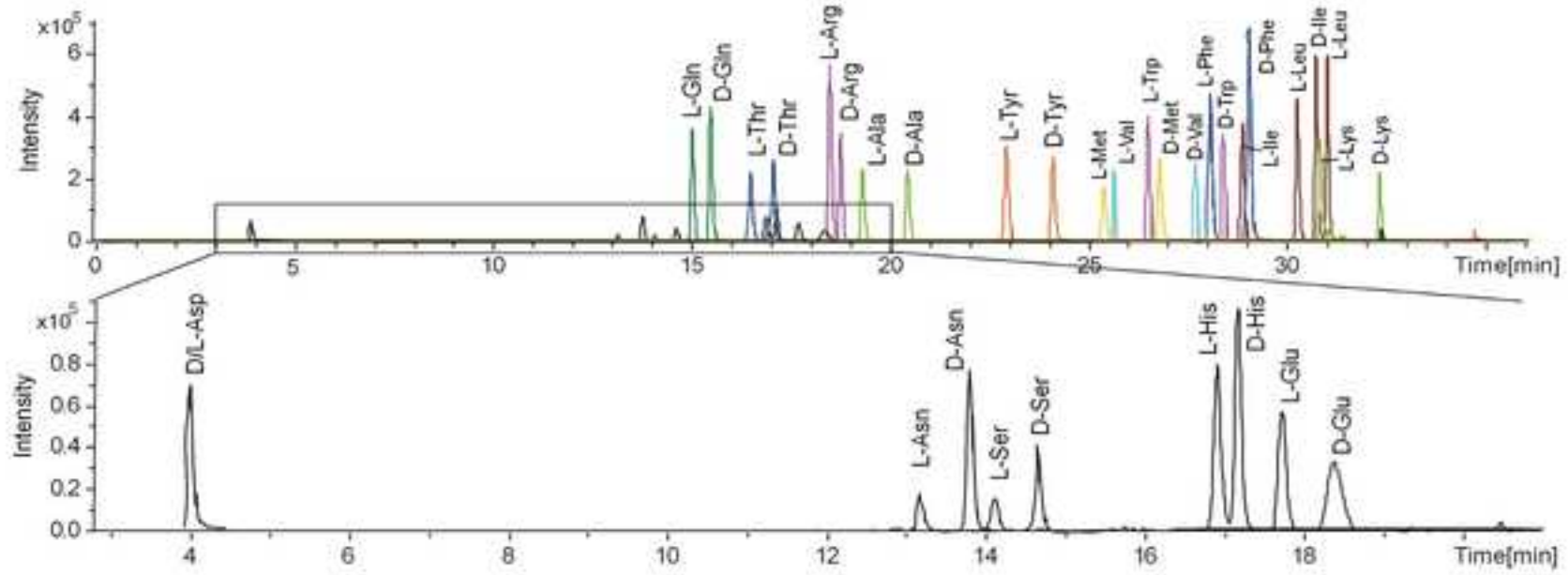
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- 355

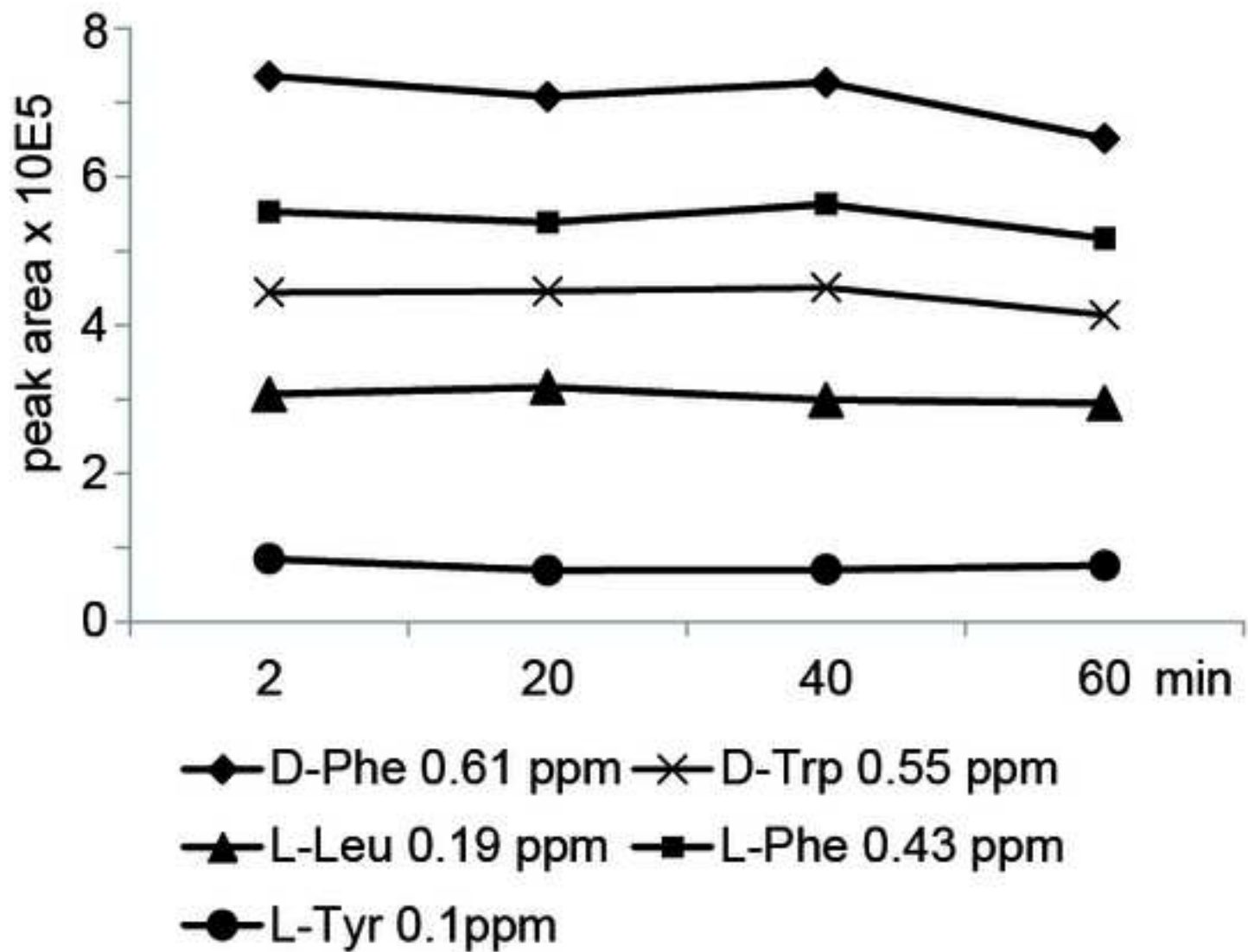
- UHPLC-QqToF-MS method using pre-column derivatization for amino acid analysis
- Discussion of the advantages of the implemented MS detection over FLD
- Enantioseparation of D and L isomers with OPA/IBLC
- No detection of matrix interferences in biological samples
- Superb sensitivity for trace level analysis of D-amino acids

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Figure

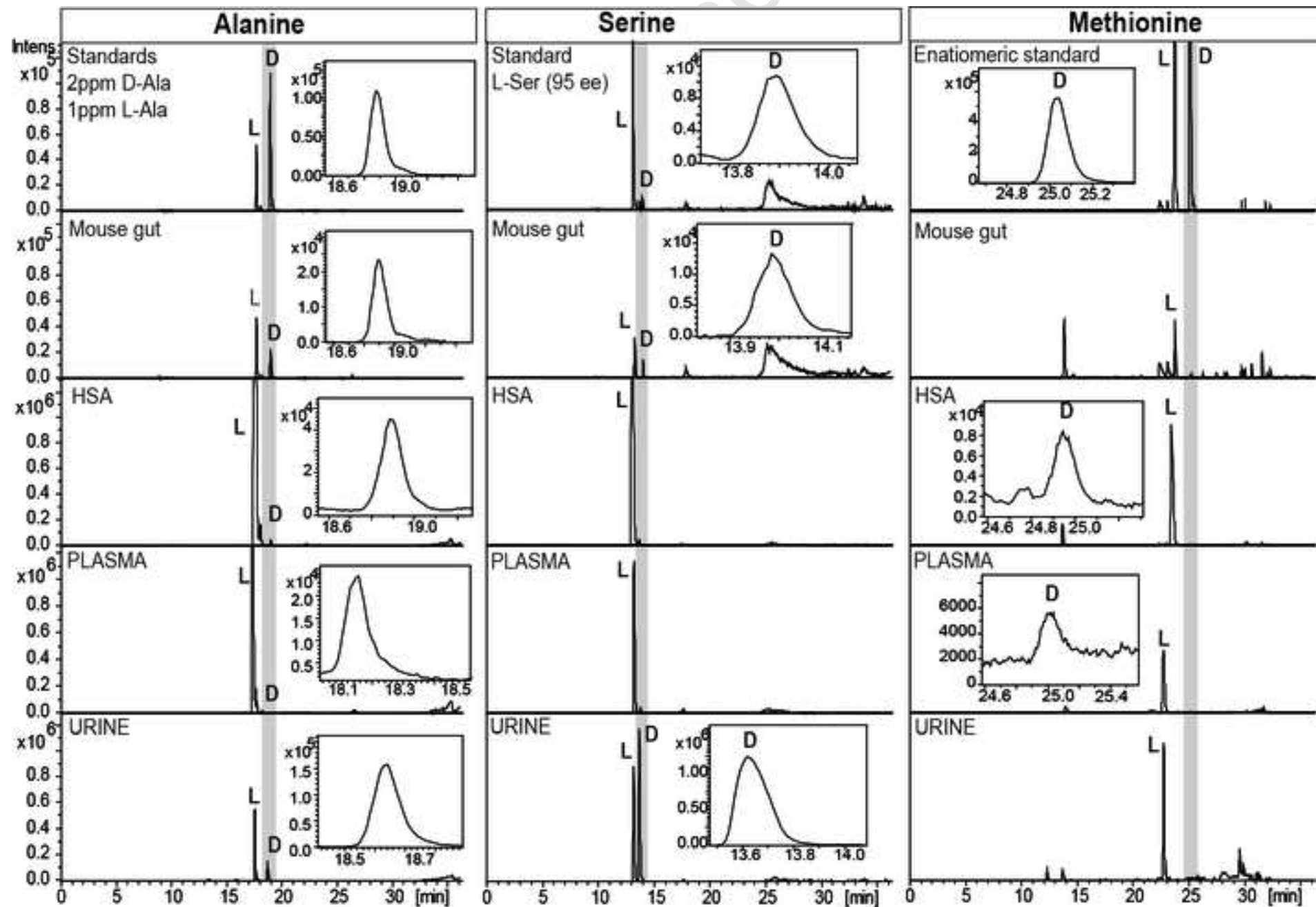


Table 1: Optimized gradient for enantiomeric separation of AA

time (min)	Flow (mL/min)	%A	%B
0	0.350	100.0	0.0
15.00	0.350	70.0	30.0
28.00	0.350	50.0	50.0
31.00	0.350	20.0	80.0
32.00	0.350	20.0	80.0
33.00	0.350	10.0	90.0
34.00	0.350	10.0	90.0
34.50	0.350	100.0	0.0

Table 2: Masses, retention time, and limits of detection for D-amino acids standards measured by UHPLC-MS using OPA/IBLC derivatization

D - AA	[M+H] ⁺ ^a (m/z)	RT (min)	LOD (pmol/L)
Ala	379.1322	20.5	11.46
Arg	464.1962	18.8	8.27
Asp	422.1380	13.8	27.09
Asn ^b	423.1221	4.1	76.43
Gln	436.1537	15.4	8.93
Glu ^b	437.1377	15.3	29.33
His	445.1540	17.1	19.22
Ile	421.1792	30.7	2.79
Leu	421.1792	31.0	2.83
Lys	436.1901	32.4	24.75
Met	439.1356	26.6	12.25
Phe	455.1635	28.8	7.31
Ser	395.1271	14.6	62.52
Thr	409.1428	17.1	40.17
Trp	494.1744	28.2	10.47
Tyr	471.1584	22.2	3.82
Val	407.1635	27.7	6.87

^a = calculated masses for protonated molecules of AA derivatives, ^b = limited soluble in water under used conditions

Table 3: Masses, LODs and calibration parameters for D-amino acids measured in plasma sample matrix by UHPLC-MS using OPA/IBLC derivatization. 4 calibration points were used except for aspartic acid, glutamic acid and lysine.

D - AA	[M+H] ⁺ ^a (m/z)	calibration range (mg/L)	calibration curve $y = aC + b$	R ²	LOD (pmol/L)
Ala	379.132	0.01 - 0.75	$y = 2\ 003\ 274.81\ C + 294\ 580.47$	0.9432	21.75
Arg	464.196	0.0002 - 1.0	$y = 740\ 377.10\ C + 127\ 178.72$	0.9807	19.6
Asn	422.138	0.01 - 0.75	$y = 641\ 037.87\ C + 66\ 622.62$	0.9577	28.37
Asp ^b	423.122	-	-	-	-
Gln	436.154	0.0002 - 1.0	$y = 1\ 532\ 437.43\ C + 239\ 481.50$	0.9808	16.83
Glu ^b	437.138	-	-	-	-
His	445.154	0.0001 - 0.5	$y = 577\ 640.12\ C + 41\ 467.93$	0.9865	61.23
Ile	421.179	0.0001 - 0.75	$y = 3\ 519\ 703.40\ C + 326\ 974.14$	0.9951	7.78
Leu	421.179	0.0001 - 0.75	$y = 2\ 880\ 663.20\ C + 163\ 095.19$	0.9965	7.08
Lys ^c	436.190	-	-	-	-
Met	439.136	0.0001 - 0.75	$y = 1\ 856\ 244.56\ C + 53\ 060.88$	0.9903	22.11
Phe	455.164	0.0002 - 1.0	$y = 2\ 859\ 517\ C + 216\ 411.03$	0.9988	20.63
Ser	395.127	0.0001 - 0.75	$y = 794\ 772.77\ C + 51\ 842.50$	0.9973	33.92
Thr	409.143	0.0002 - 1.5	$y = 1\ 158\ 402.21\ C + 65\ 359.65$	0.9969	79.43
Trp	494.174	0.0002 - 1.0	$y = 1\ 237\ 427.67\ C + 37\ 510.10$	0.9975	20.39
Tyr	471.158	0.0002 - 1.5	$y = 248\ 853\ C + 54\ 497$	0.9780	64.35
Val	407.164	0.01 - 0.75	$y = 2\ 278\ 485.89\ C + 224\ 692.79$	0.9830	159.18

^a = calculated values for protonated molecules of AA derivatives, ^b = limited soluble in water under used conditions, ^c = basic concentration too high, LOD calculation possible with 10 fold dilution of the matrix